

Determination of iodine in iodides. R. L. RAOGORODSKA and E. S. BINOVA (Farm. Shur., 1935, No. 1, 23-25).—Free I is titrated with  $\text{Na}_2\text{S}_2\text{O}_3$  and total I' with  $\text{AgNO}_3$ .  
Ca. Ann. (e)

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CIA-RDP86-00513R001344020008-9"

Determination of salicylic acid. R. I. Raigorskaya and E. S. Vinova. *Farm. Zhur.* 1934, 137-40. Salicylic acid can be detd. by treatment with  $I_2$  and titration of the excess, or by a bromometric modification of Koltzoff's method, which is very accurate. L. Nasarevich

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Detection and determination of salicylic and benzoic acids and their salts. V. S. Il'ina and R. I. Ruzgorskaya. Izv. Akad. Nauk. 1934, no. 4. Detection is effected by sublimation in a Hg sublimate. Benzoic acid volatilizes at 250-260° and the salicylic acid at 70-85°, to form their characteristic crystal sublimes. Salts must be converted into the acids. For quantitative work the methods of Messinger (J. Prakt. Chem., 61, 20, 1890) or Baumgärtel (ibid., 2, 250) and of Kesthelyi (ibid., 15, 280) are recommended.

I. N. G.

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17  
PROCESSES AND PROPERTIES INDEX  
  
Microchemical determination of alkaloids of ipecacuanha. B. S. Binova and R. L. Raigorod'ska, *Trans. Ukraine. Inst. Exptl. Pharm.*, 1, 151 (1936). The qual. reagent used was a mixt. of NH<sub>4</sub> molybdate and H<sub>2</sub>SO<sub>4</sub>. This gives a green color with the alkaloid. For the detn., MgO, tragacanth and chloroform were used. One hundred mg. of powd. root can be analyzed by this method.  
R. Levine

## AIA-SLA METALLURGICAL LITERATURE CLASSIFICATION

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Determination of Iodine with Iodides R. E. Babu  
govinda and K. S. Bhava *Zeta Zeta* 1975, No. 1,  
20 A. Free I<sub>2</sub> is titrated with Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, the iodides are  
then titrated with AgNO<sub>3</sub> and the free I<sub>2</sub> equiv. is sub-  
tracted from total iodides. 1 Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>

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1st AND TWO COLUMNS  
PROCESSES AND PROPERTIES INDEX  
3rd AND 4th COLUMNS

**Acetylene derivatives. LXXXVI. Heterocyclic compounds. 7. Synthesis of 4-vinylethynyl-4-hydroxypiperidines by condensation of vinylacetylenes with 4-piperidones.** I. N. Nazarov, V. Ya. Ralgorodskaya, and V. A. Rudenko. *Izvest. Akad. Nauk S.S.R., Otdel. Khim. Nauk* 1949, No. 1, 68-75; cf. *C.A.* 43, 2094.  $\text{CH}_3\text{CH}=\text{CH}_2$  (7 g.) and 14.1 g. 2,5-dimethyl-4-piperidone in 30 ml.  $\text{Et}_2\text{O}$  added to 10 g. powd. KOH in 30 ml.  $\text{Et}_2\text{O}$  with stirring at  $-5^\circ$ , stirred 8 hrs. in the cold, and let stand overnight, gave 12.3 g. 2,5-dimethyl-4-vinylethynyl-4-hydroxypiperidine (I),  $b_f$  135-6 $^\circ$ . It hydrogenated with Pt catalyst in  $\text{Et}_2\text{O}$  gave 2,5-dimethyl-4-butyl-4-hydroxypiperidine, needles, m. 106-7 $^\circ$  (sublimed for purification). The latter, b. 85-6 $^\circ$  (liquid form), was obtained also in 7.3 g. yield from  $\text{Bu}_2\text{MgCl}$  (30 g.  $\text{Bu}_2\text{Cl}$ ) and 20 g. 2,5-dimethyl-4-piperidone, after heating in  $\text{Et}_2\text{O}$ , then in  $\text{C}_6\text{H}_6$  to 70 $^\circ$ , prior to decomprn. with 15% HCl and ice.  $\text{CH}_3\text{CH}=\text{CH}_2$  (6 g.) and 12.6 g. 2,5,6-trimethyl-4-piperidone, treated as above, gave 13.2 g. 2,5,6-trimethyl-4-vinylethynyl-4-hydroxypiperidine,  $b_f$  140-3 $^\circ$ , m. 123.5-4.5 $^\circ$  (from  $\text{C}_6\text{H}_6$ ); 4-Bu analog, m. 135-8 $^\circ$  (from  $\text{C}_6\text{H}_6$ ). 2,5-Dimethyl-6-ethyl-4-piperidone (11.3 g.) gave 10 g. 2,5-dimethyl-6-ethyl-4-vinylethynyl-4-hydroxypiperidine,  $b_f$  143-5 $^\circ$ ; 4-Bu analog,  $b_f$  142-1 $^\circ$ ,  $n_D^{20}$  1.4825,  $d_4^{20}$  0.9301. Similarly, 11.3 g. 2-methyl-4-ketodecahydroquinoline gave 9.8 g. 2-methyl-4-hydroxy-4-vinylethynyldecahydroquinoline,  $b_f$  140-1 $^\circ$ ; 4-Bu analog, m. 136-8 $^\circ$  (from  $\text{C}_6\text{H}_6$ ). Analogously, 23 g. 1,2,5,6-tetramethyl-4-piperidone gave 14 g. 1,2,5,6-tetramethyl-4-vinylethynyl-4-hydroxypiperidine,  $b_f$  122-2.5 $^\circ$ ,  $n_D^{20}$  1.6138,  $d_4^{20}$  0.9658; 4-Bu analog,  $b_f$  104-1 $^\circ$ ,  $n_D^{20}$  1.4894,  $d_4^{20}$  0.9418 [also prep'd. in 25% yield from  $\text{Bu}_2\text{MgCl}$  and the corresponding piperidone; in this case, the product,  $b_f$  113-15 $^\circ$ , crystall., and m. 58-9 $^\circ$  (from petr. ether); apparently the Grignard syntheses yield isomers of the compds. obtained by the  $\text{CH}_3\text{CH}=\text{CH}_2$  condensation]. **LXXXVII. Mechanism of dehydration and cyclization of diynes.** 17. **Hydration and cyclization of 5-propyl-1,5-octadien-3-yne.** I. N. Nazarov and I. I. Zuretskaya. *Ibid.* 178-83; cf. *C.A.* 43, 115.— $\text{CH}_3\text{CH}=\text{CH}_2$  (159 g.) in 200 ml. dry  $\text{Et}_2\text{O}$  and 230 g.  $\text{Pr}_2\text{CO}$  added to 155 g. powd. KOH in 400 ml.  $\text{Et}_2\text{O}$  in the cold over 4 hrs., and stirred 8 hrs. in the cold and 4 hrs. at room temp. gave, after the usual aq. treatment, 251 g. (77%) dipropyl(vinylethynyl)carbinol,  $b_f$  73-6 $^\circ$ ,  $n_D^{20}$  1.4780. This (103 g.) and 100 g. 50%  $\text{H}_2\text{SO}_4$ , stirred 5 hrs. at 60-5 $^\circ$  gave 81% 5-propyl-1,5-octadien-3-yne (I),  $b_f$  65-7 $^\circ$ ,  $n_D^{20}$  1.4900. I (6 g.) hydrogenated in  $\text{AcOH}$  with Pt oxide gave 5-propylacetate, b. 174-0 $^\circ$ ,  $n_D^{20}$  1.4190,  $d_4^{20}$  0.7447. I (77 g.), 300 g. 90%  $\text{MeOH}$ , 1.5 ml.  $\text{H}_2\text{SO}_4$ , and 4 g.  $\text{Hg}$  sulfate stirred 1 hr. at 65 $^\circ$ , then treated with a total of 10 g.  $\text{Hg}$  sulfate in 2 portions spaced 2.5 hrs. (total time 6 hrs.) gave, after concn. and neutralization, 87 g. 5-propyl-1,5-octadien-4-one (contaminated with some 5-propyl-2-methoxy-5-octen-4-one),  $b_f$  90-8 $^\circ$ ; the impurity was removed by 1 hr. at 60-100 $^\circ$  and 75 mm. with 0.5 g.  $p$ -MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H (3 (faux))

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*LXXXVII.*  $b_2$  103-0°,  $n_D^20$  1.4770,  $d_4^{\circ}$  0.8900, which does not give a semicarbazone or dinitrophenylhydrazone, nor undergo hydration under the above conditions. Hydrogenation over Pt oxide in EtOH gave 5-propyl-4-octanone,  $b_2$  89-92°,  $b_2$  201-4°,  $n_D^20$  1.4300,  $d_4^{\circ}$  0.8200, which yields a 2,4-dinitrophenylhydrazone, m. 61° (from EtOH), but not an oxime or semicarbazone. The structure of the dienone was shown by ozonization, which gave  $\text{HCO}_2\text{H}$ ,  $\text{AcO}_2\text{H}$ ,  $\text{EtCO}_2\text{H}$ , and  $\text{PrCO}_2\text{H}$ . The dienone (25 g.) and 45 g.  $\text{H}_3\text{PO}_4$  (d. 1.77) after 2.5 hrs. at 80-5° gave 20 g. 1-propyl-2-ethyl-3-methyl-1-cyclopenten-5-one,  $b_2$  103-5°,  $n_D^20$  1.4770,  $d_4^{\circ}$  0.9116; semicarbazone, m. 185-8° (from EtOH); 2,4-dinitrophenylhydrazone, m. 137-8°. The same product is obtained with undiminished yield on treatment with  $\text{H}_3\text{PO}_4$  11 hrs. at 10-14°. Hydrogenation over Pt oxide gave in poor yield the corresponding cyclopentanone deriv.; semicarbone, m. 173-4°. Treatment of (28 g.) 5-propyl-1,5-octadien-3-yne with 50 g.  $\text{H}_3\text{PO}_4$  at 80-5° for 8 hrs. gave the above cyclopentenone directly but in very poor yield (1 g.), the rest of the dienone being unchanged. The structure of the cyclopentenone is shown by ozonization, which yields  $\text{PrCO}_2\text{H}$  and  $\text{EtCO}_2\text{H}$ ;  $\text{CH}_3\text{MeCH}_2\text{CO}_2\text{H}$ ,  $b_2$  130-9°,  $n_D^20$  1.4680, and by  $\text{KMnO}_4$  oxidation, which yields the same products; the above keto acid yields a 2,4-dinitrophenylhydrazone, m. 115-18°. *LXXXVIII.* The mechanism of hydration and cyclization of diynes. 18. Hydration and cyclization of 5-methyl-1,5-tetradecadien-3-yne. *Ibid.* 184-0. Adm. of 344 g.  $\text{C}_5\text{H}_11\text{COMe}$ , 150 ml.  $\text{CH}_3\text{CHClCH}_2$ , and 200 ml. dry  $\text{Et}_2\text{O}$  to 130 g. powd. KOH in 500 ml.  $\text{Et}_2\text{O}$  with stirring and cooling over 4 hrs., followed by 3 hrs. stirring in the cold and 4 hrs. at room temp. gave, after the usual aq. treatment and neutralization, 322 g. mixed methylnonyl- (vinylethynyl)carbinol and 5-methyl-1,5-tetradecadien-3-yne (I),  $b_2$  103-30°; repeated fractionation gave 160 g. of the former,  $b_2$  120-2°,  $b_2$  131-3°,  $n_D^20$  1.4725,  $d_4^{\circ}$  0.8580, which on hydrogenation over Pt oxide in EtOH gave 5-methyl-5-tetradecanol,  $b_2$  135-8°,  $n_D^20$  1.4440,  $d_4^{\circ}$  0.8303, while heating the carbisol (80 g.) with 80 g. 65%  $\text{H}_2\text{SO}_4$  4 hrs. to 65° gave 79% pure I,  $b_2$  100-2°,  $n_D^20$  1.4850,  $d_4^{\circ}$  0.8183. Hydrogenation of I over Pt oxide in  $\text{AcOH}$  gave 5-methyldecanoate,  $b_2$  125-7°,  $n_D^20$  1.4320,  $d_4^{\circ}$  0.7808. Stirring 58 g. I, 201 g. 90%  $\text{MeOH}$ , 1 ml.  $\text{H}_2\text{SO}_4$ , and 4 g.  $\text{Hg}$  sulfate 2 hrs. at 70°, followed by addn. of a total of 12 g.  $\text{Hg}$  sulfate in 2 g. portions at 2-hr. intervals, gave 43 g. 5-methyl-1,5-tetradecen-4-one,  $b_2$  120-31°,  $n_D^20$  1.4781,  $d_4^{\circ}$  0.8730; shorter reaction results in partial reaction; no MeO derivs. were detected. The dienone on hydrogenation over Pt oxide in EtOH gave 5-methyl-4-tetradecanone,  $b_2$  140-1°,  $n_D^20$  1.4400,  $d_4^{\circ}$  0.8278, and a small amt. of the cyclic deriv. described below. The dienone (32 g.) and 50 g.  $\text{H}_3\text{PO}_4$  (d. 1.77) after 3 hrs. at 80-8° gave 30 g. 1,3-dimethyl-2-acyl-1-cyclopenten-3-one,  $b_2$  134-5°,  $n_D^20$  1.4763,  $d_4^{\circ}$  0.8920; semicarbone, m. 134-5°; 2,4-dinitrophenylhydrazone, m. 99-100° (from EtOH). This ketone is resistant to hydrogenation over a Pt catalyst. Oxidation by  $\text{KMnO}_4$  gave  $\text{AcOH}$ , methylsuccinic acid, caprylic acid, and  $\text{C}_5\text{H}_11\text{COCH}_2\text{MeCH}_2\text{CO}_2\text{H}$ ,  $b_2$  152-4°,  $n_D^20$  1.4595, which forms a semicarbone, m. 124-5° (from 50% EtOH). *LXXXIX.* Transformations of 2-butene-1,4-diol. I. N. Nazarov, L. N. Terekhova, and I. V. Torgov. *Ibid.* 287-92.— $\text{MeOH}$  (700 g.) and 40 ml. of a soln. of 15 g.  $\text{H}_2\text{SO}_4$  and 20 g.  $\text{Hg}$  sulfate in 135 ml.  $\text{H}_2\text{O}$  were treated simultaneously with the rest of the above soln. and 300 g.  $(\text{HOCH}_2\text{C}_2)_2$  in 300 ml.  $\text{MeOH}$  at 30° over 4 hrs.; then 2 g.  $\text{Hg}$  sulfate was added and stirred 5 hrs. at 30°; neutralization gave 74% 1-methoxybutan-4-ol-3-one (I),  $b_2$  85°,  $d_4^{\circ}$  1.0930,  $n_D^20$  1.4390; semicarbone, m. 123-3.5° (from  $\text{MeOH}$ ). To 40 g.  $\text{MeOH}$  and the catalyst from 2 g.  $\text{HgO}$ , 2 ml.  $\text{MeOH}$ , and 0.5 ml.  $\text{BF}_3\text{Et}_2\text{O}$  was added 20 g.  $(\text{HOCH}_2\text{C}_2)_2$  in 20 ml.  $\text{MeOH}$  over 2.5 hrs. at 4°, and the mixt. heated 2 hrs. to 50°, and neutralized with  $\text{Na}_2\text{CO}_3$ , giving 18.5 g. I; a poorer yield is obtained with  $\text{Hg}$  sulfate in dry  $\text{MeOH}$ . Heating I (110 g.) and 0.9 g.  $p\text{-MeC}_6\text{H}_4\text{SO}_3\text{Na}$  to 85° at 50 mm.

(Cont)

## PROCESSES AND PROPERTIES INDEX

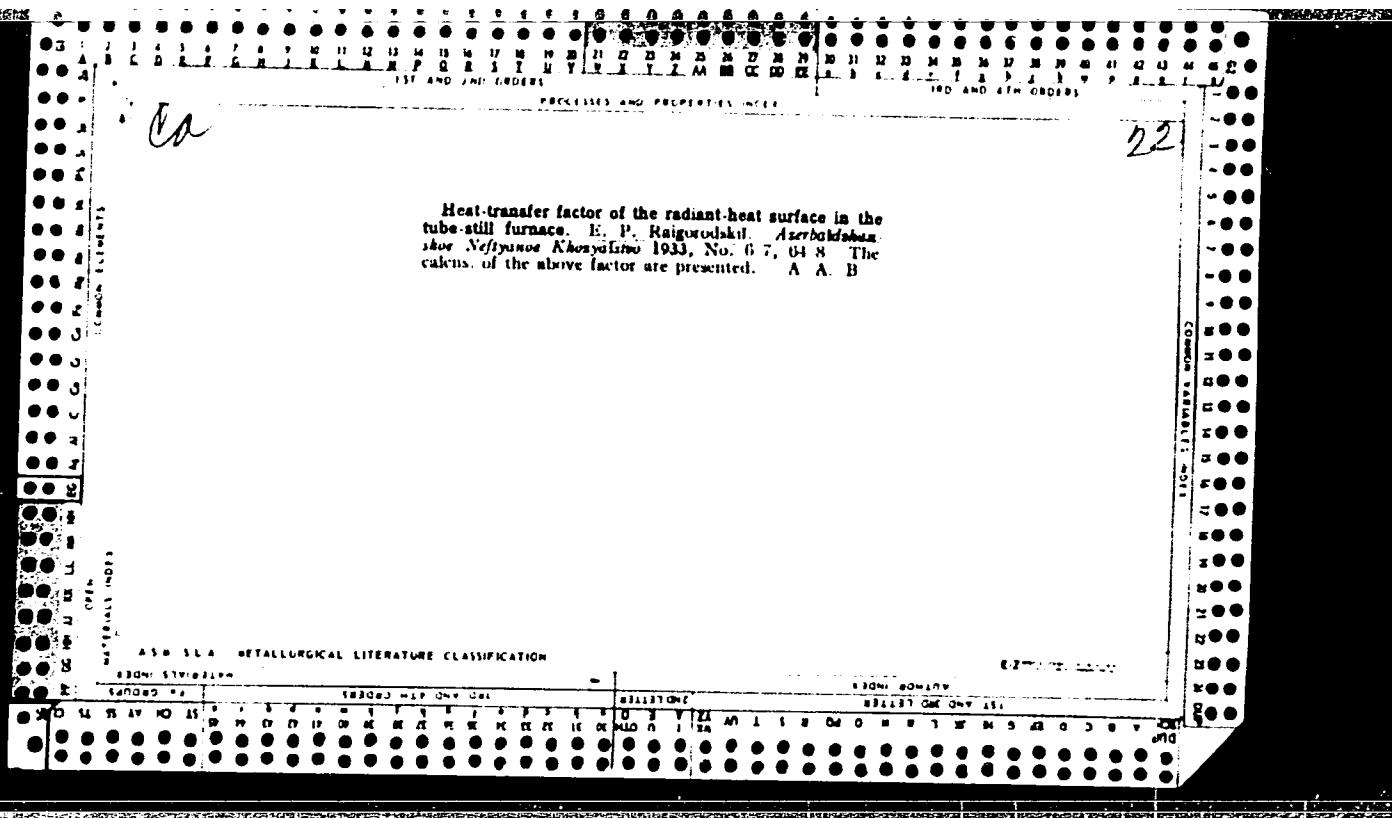
gave 2.5 g. *1-buten-4-ol-3-one*,  $b_{14}$  61-2°,  $n_D^2$  1.4528,  $d_4^{20}$  1.1113; 2,4-dinitrophenylhydrazone, m. 232-4° (from  $\text{C}_6\text{H}_5\text{NO}_2$ ); the keto alc. polymerizes on standing to a colorless solid; I (90 g.), 0.7 g. pyrogallol, and 0.76 g.  $p$ -Me $\text{C}_6\text{H}_4\text{SO}_3\text{H}$  heated to 50-80° at 60 mm. gave 40 g. distillate, which was immediately hydrogenated over Pt; fractionation of the combined products of several runs gave 0.5 g. *butan-1-ol-2-one*,  $b_{14}$  64-5°,  $n_D^2$  1.4200,  $d_4^{20}$  1.026; osazone, m. 117°. XC. Mechanism of hydration and cyclization of dienynes. 19. Hydration and cyclization of 5,6-diphenyl-1,5-hexadien-3-yne. I. N. Nazarov and I. L. Kotlyarevskii. *Ibid.* 293-8.—To the Grignard reagent of  $\text{CH}_3\text{CH}(\text{C}_6\text{H}_5)\text{CH}_3$  (from 100 g.  $\text{EtBr}$ , 24 g.  $\text{Mg}$ , and 70 g.  $\text{CH}_3\text{CH}(\text{C}_6\text{H}_5)\text{CH}_3$ ) in 300 ml.  $\text{Et}_2\text{O}$  was added with cooling 130 g.  $\text{PhCOCH}_2\text{Ph}$  in  $\text{Et}_2\text{O}$ ; stirring 2 hrs. at 35°, letting stand overnight, and treating with 10% HCl gave *phenylbenzyl(cinnylethynyl)carbinol*,  $b_{14}$  162°,  $n_D^2$  1.5912,  $d_4^{20}$  1.009, which slowly polymerizes on standing; hydrogenation over Pt gave *phenylbenzylbutylcarbinol*,  $b_{14}$  147-8°,  $n_D^2$  1.5559,  $d_4^{20}$  1.030; dehydration by  $\text{HgSO}_4$  at 130-40° and 10 mm. gave 5,6-diphenyl-1,5-hexadien-3-yne (I),  $b_{14}$  150-60°,  $n_D^2$  1.6757, also obtained by stirring with 60%  $\text{H}_2\text{SO}_4$  at 65-70°. Hydrogenation of the latter gave 5,6-diphenylhexane,  $b_{14}$  154°,  $n_D^2$  1.6370,  $d_4^{20}$  0.956. I (115 g. crude), 400 ml. 90% MeOH, 10 ml.  $\text{H}_2\text{SO}_4$ , and 30 g. Hg sulfate stirred 18 hrs. at 65-8°, cooled, diluted with water, neutralized, and the  $\text{C}_6\text{H}_5$  ext. b. of the mixt. rapidly distd. in small portions with  $p$ -

$\text{MeC}_6\text{H}_4\text{SO}_3\text{H}$  in *vacuo* gave 87 g. 5,6-diphenyl-1,5-hexadien-4-one (II),  $b_{14}$  170-9.5°,  $n_D^2$  1.6320; semicarbones could not be formed; on long standing the ketone solidifies and m. 70°; hydrogenation over Pt gave 5,6-diphenyl-4-hexanone,  $b_{14}$  117-9°,  $n_D^2$  1.5447,  $d_4^{20}$  1.011; semicarbone, m. 181.5° (from  $\text{EtOH}$ ); ozonization gave  $\text{BaOH}$  and  $\text{HCO}_2\text{H}$ . II (4 g.) and 4 ml.  $\text{H}_3\text{PO}_4$  (d. 1.82) after 15 min. at 70-80° give 3.8 g. 1,2-diphenyl-3-methyl-1-cyclopenten-5-one,  $b_{14}$  184°, m. 100° [semicarbone, m. 220° (from  $\text{EtOH}$ )], also formed on heating with  $p$ -Me $\text{C}_6\text{H}_4\text{SO}_3\text{H}$ ; hydrogenation over Pt gave 1,2-diphenyl-3-methyl-5-cyclopentanone, m. 97-8°; semicarbone, m. 208-9° (from  $\text{EtOH}$ ). Ozonation of the cyclopentene gave  $\text{BaOH}$  and  $\alpha$ -benzoylethrylic acid,  $b_{14}$  155-62°, m. 60-5° [semicarbone, m. 176-7°]. XC1. Chemistry of divinyl ketones. 16. Addition of hydrogen cyanide to 2,2-dimethyldivinyl ketone. I. N. Nazarov and M. V. Kuvarzina. *Ibid.* 293-304.—To 22 g.  $\text{Me}_2\text{CHCOCH}_2\text{CH}_3$ , 20 ml.  $\text{EtOH}$ , and 26 g.  $\text{AcOH}$  was added in 25 min. 28 g. KCN in 90  $\text{H}_2\text{O}$ , the mixt. filtered after 4 hrs. at 35-7°, cooled, and extd. with  $\text{Et}_2\text{O}$  to yield 21 g. 1-cyano-5-methyl-4-hexen-3-one (I),  $b_{14}$  103-4°,  $n_D^2$  1.4725,  $d_4^{20}$  0.978; semicarbone, m. 130° (from dil.  $\text{MeOH}$ ). Use of excess KCN failed to give further addn. Oxidation with  $\text{KMnO}_4$  gave  $\text{Me}_2\text{CO}$  and succinic acid; ozonization gave  $\text{HCO}_2\text{H}$  and succinic acid; hydrogena-

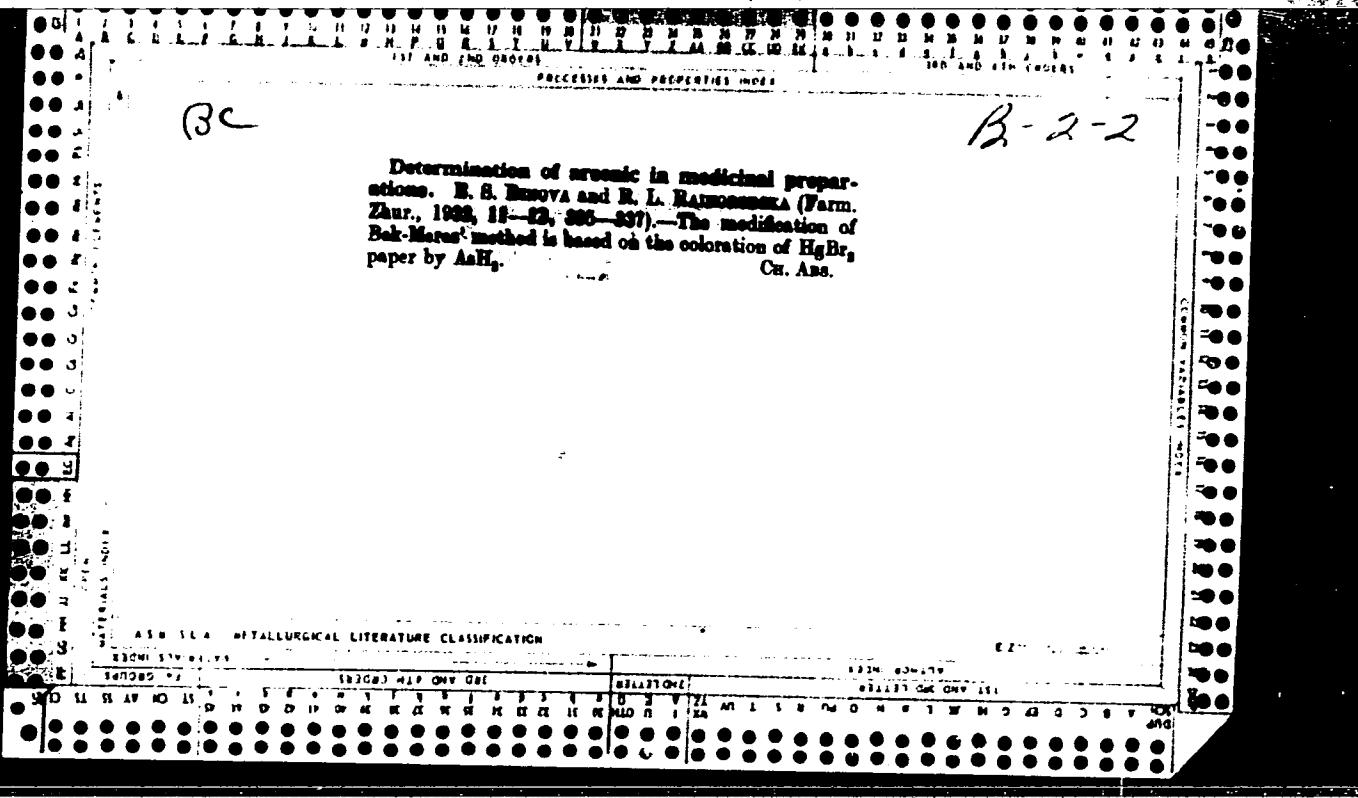
## ASTM DATA METALLURGICAL LITERATURE CLASSIFICATION

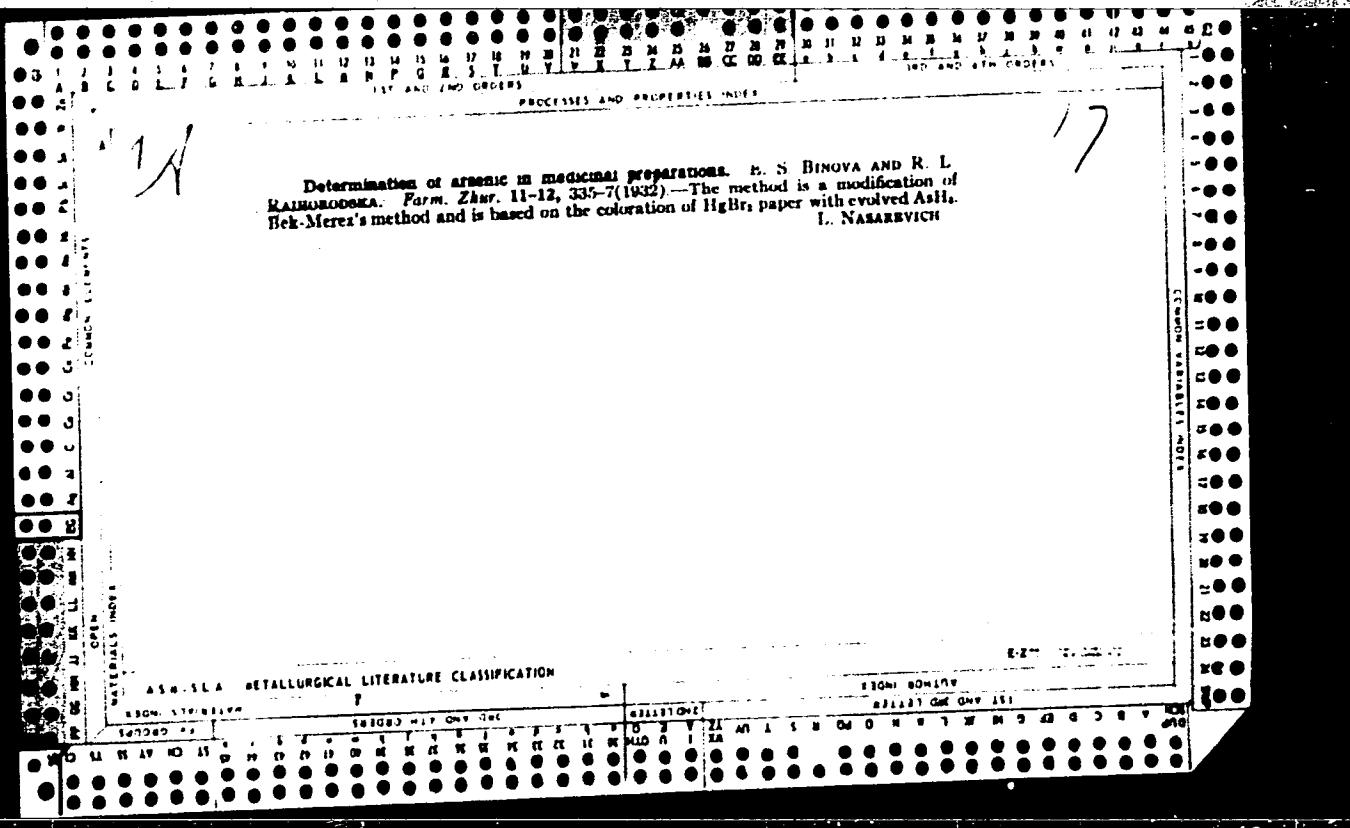
tion over Pt gave *l*-syno-5-methyl-3-hexanone, b.p. 133°;  $n_D^{20}$  1.4315,  $d_4^{20}$  0.9322 [semicarbazone, m. 139.5-40° (from 50% MeOH)]; hydrolysis of this satd. ketone by 30% H<sub>2</sub>SO<sub>4</sub> 6 hrs. at 70-5° gave  $\delta$ -isopropyllevulinic acid, m. 47° (from ligroin); semicarbazone, m. 147-8° (from 50% MeOH). Similar hydrolysis of the unsatd. ketone gave levulinic acid, sepd. by distn., and a poor yield of  $\text{Me}_2\text{C}(\text{CHCOCH}_2\text{CH}_2\text{CO}_2\text{H}$ , m. 73-4° (from water) [semicarbazone, m. 150°]; hydrogenation over Pt gave the above-described satd. acid, m. 47-8°. Use of more concd. acid gives only levulinic acid, due to cleavage of the unsatd. ketone. Addn. of 22 g. CH<sub>3</sub>CHCOCH<sub>2</sub>CClMe<sub>2</sub> in 22 ml. EtOH to 9 g. KCN in 15 ml. water raised the temp. to 42°, which was maintained 40 min.; stirring 3 hrs. and letting stand overnight gave 4.2 g.  $\text{Me}_2\text{C}(\text{CHCOCH}_2\text{CH}_2\text{CO}_2\text{H}$  and 10 g. I. Addn. of KCN to  $\text{MeOCH}_2\text{CH}_2\text{COCH}_2\text{CMe}_2$  in EtOH in the presence of AcOH gave only tars; the use of large aunts. of AcOH gave no reaction.

G. M. Kosolapoff



1. KUZNETSOV, N. P., Dozent; MIKHAILEV, YA. I., Dozent; KIRAKOVSKA, L. M.
2. USSR (Sov)
3. Influenza
4. Pathological changes in the nervous system in grippe, Medyer. zhur., 22, no. 1, 1952.
- 5.
- 6.
- 7.
- 8.
9. Monthly List of Russian Accessions, Library of Congress, April 1953, Uncl.





"APPROVED FOR RELEASE: 03/14/2001

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RAICU, P.; EREMIA, P.

Comparative research on the biology of the blossoming of the double hybrid maize Warwick 401, its simple hybrids, and consanguineous lines. Studii cerc biol veget 13 no.2:243-260 '61.  
(EEAI 10:11/12)

1. Comunicare prezentata de Al. Priadcencu, membru corespondent al Academici R.P.R.

(Corn(Maize)) (Hybridization)

KIVILO, Evald; RAIG, H., otv. red.

[Elementary course in hygiene for workers of dairy farms] Sanitaarminimumi kursus piimakarjafarmide töötajaile. Tartu, Vabariiklik sanitaarhariduse maja, 1962. 61 p. (MIRA 16:12)

BAIK, A.Ye.

Calculation of certain volumes as presented in the ancient  
Chinese treatise "Mathematics in nine books." Ist. mat.  
issl. no.14:467-472 '61. (MIRA 16:10)

(Mathematics, Chinese)

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9

Raik, A. E. The tenth book of Euclid's "Elements."  
Trudy Sem. MGU Istor. Mat. Istor. Mat. Issledov. no. 1,  
343-384 (1948). (Russian)

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Source: Mathematical Reviews, Vol 11 No, 3

APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001344020008-9"

Raik, A.Ye. (Baranov)

New reconstruction of certain problems from ancient Egyptian and  
Babylonian texts. Ist.-mat.issl. no.11:171-182 '58.

(MIRA 12:1)

(Mathematics, Babylonian)

(Mathematics, Egyptian)

KELMAN, E

P 4

16(1)

PHASE I BOOK EXPLOITATION SOV/1366

Istoriko -matematicheskiye issledovaniya, vyp. 11 (Research in  
Mathematical History, Nr 11) Moscow, Fizmatgiz, 1958. 792 p.  
3,000 copies printed.

Eds. (Title page): Rybkin, G.F. and Tushkevich, A.P.; Ed. (Inside  
book); Konoplyankin, A.A.; Tech. Ed.: Murashova, N. Ya.

PURPOSE: This book is intended for mathematicians and others  
interested in the history of mathematics, and may serve as the  
basis for a suitable university text on the history of mathematics,  
thereby filling the most serious gap in Soviet mathematical  
literature.

COVERAGE: This book contains reports made by members of the section  
on the history of mathematics at the Third All-Union Mathematical  
Congress which discussed problems of the history of mathematics and  
various articles on the significance of the history of mathematics

Card 1/8

Raik, A. Ye. (Saransk). New Reconstructions of Certain  
Problems from Ancient Egyptian and Babylonian Texts

171

RAIK, A.Ye.

Ivan Mikheevich Pervushin, the mathematician from the Urals. Ist.-  
mat. issl. no.6:535-572 '53. (MLRA 7:9)  
(Pervushin, Ivan Mikheevich, 1827-1900)

RAIK, A. Ye.

RAIK, A. Ye. APPROVED FOR RELEASE: 03/14/2001, In CIA-RDP86-00513R001344020008-9  
matem. issledovaniya, issue 1, Moscow, 1946, p. 342-34.

SC: u-3042, ll, March 53, (Letopis 'nykh Statej, No. 9, 1949)

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9

In memory of I.I.Raik. Vest.ven.1 derm. no.2:63 Mr-Ap '54. (MLRA 7:4)  
Raik, I.I., ? -1953)

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9"

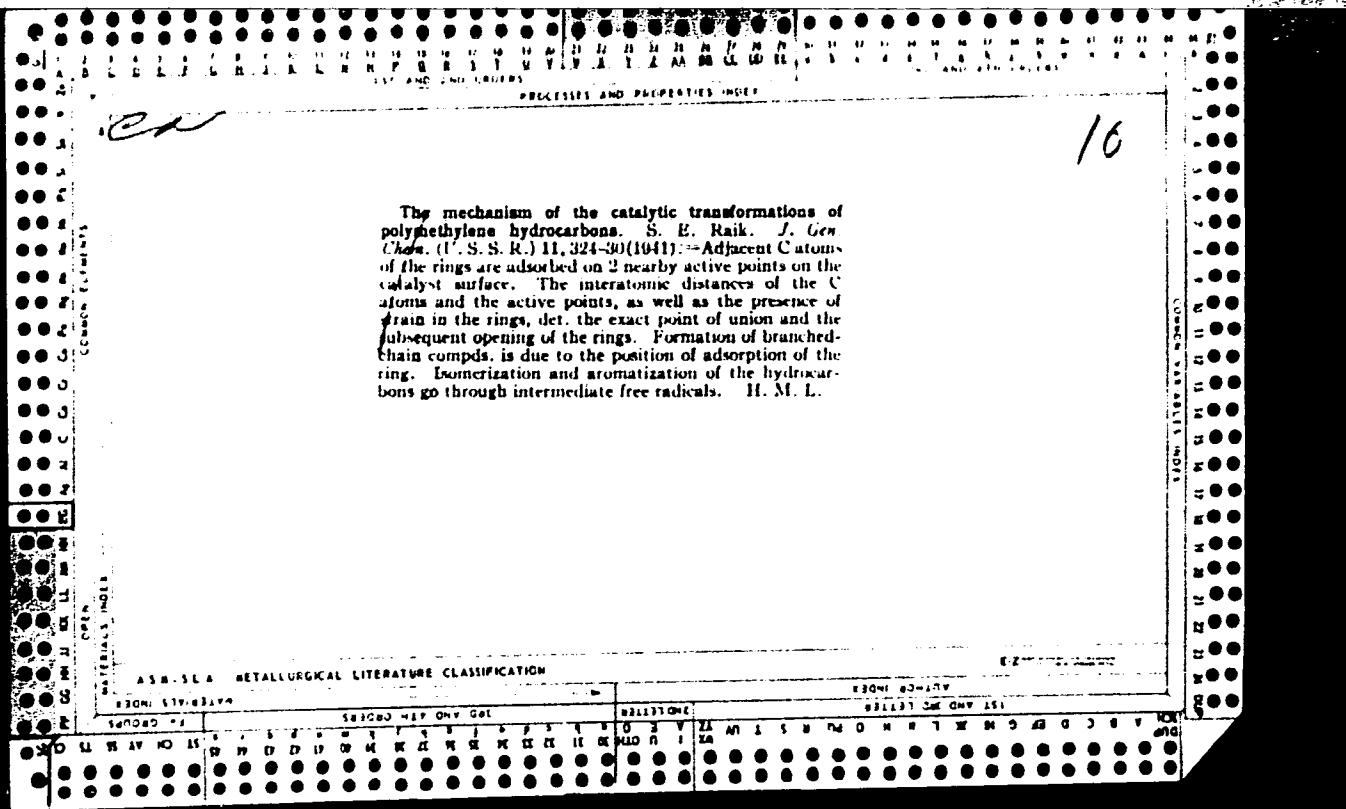
KHETAGUROV, G.I.; RAIK, I.O.

Results of four years of investigation on penacillin therapy of  
syphilis. Sovet. med. no.5:21-23 May 1951. (CLML 20:9)

1. Of Leningrad Skin-Venereological Institute of the Ministry of Public Health RSFSR (Director of Institute and Scientific Supervisor of Syphilological Department--Prof. S.Ye. Gorbovitskiy).

BAL'AKA, S.V.; RAIK, S.Ya.; DEGYAREVA, V., red.

[Pectic substances and their importance in the national economy] Pektinovye veshchestva i ikh znachenie v narodnom khoziaistve. Kishinev, Kartia Moldoveniasko, 1963.  
17 p.  
(MIRA 17:12)



CA

10

The dehydrocyclization of substituted pentanes and of  
gem-substituted hexanes. A. A. Balandin and S. E.  
Raik. *Compt. rend. acad. U.R.S.S.* 50, 101 (1947)  
(in English).—Three postulates are given in an explana-  
tion of the dehydrocyclization of substituted  $\alpha$ -pentanes  
(cf. *C.I.* 40, 1700) by way of a skeleton isomerization.  
These postulates, which occur simultaneously, are: (1)  
in such substituted pentanes and hexanes a 5-membered  
ring compd. is formed; (2) the 5-membered ring is iso-  
merized into a hexahydroaromatic 6-membered ring in such  
a way that the  $\alpha$ -C atom of the side chain enters into the  
neighboring C-C bond of the 5-membered ring; and (3)  
this entering of the new C takes place into those bonds of  
the 5-membered ring that are farthest from the newly  
formed bond of the 5-membered ring formed from the parent  
hydrocarbon. Thus, from 2,2,1 trimethylpentane, *p*-  
ethylene is formed, whereas from 2,2 dimethylcyclohexane  
*m*-xylene is formed. Of the 9 cases investigated, the re-  
sults are in quant. agreement with the above postulates in  
7 cases, with but 1 case being markedly divergent.  
Myron Q. Webb

RAYK, S. Ye.

Mechanism of catalytic opening of rings. S. E. Kulin.  
Problemy Kinetiki i Kataliza, Akad. Nauk S.S.R. 6, *FCL*  
Geterogennyi Kataliz, 259-61(1949).—Catalytic hydrogenation of cyclobutanes with side chains in the presence of Pt on an active C carrier, yielded open chain, branched hydrocarbons. Thus, the ring was broken preferentially in bonds remote from the point of attachment of the side chain. It is proposed that the catalyst attacks first one C atom, then the second neighboring C and causes stretching of the C—C bond. Considerations of steric hindrance and bond lengths indicate that the probability for a given C to become attached to the catalyst is related to the no. of H atoms attached to this C. The probability for the formation of the iso-compd. is 80%, since formation of the normal hydrocarbon would require that the tertiary C is attached before the ring opens. Similar considerations can be applied to the opening of pentamethylene ring; the yield of normal hydrocarbons from Me cyclopentene is 12% as compared to the predicted 11%. The actual yield of 2.4 dimethylpentene from 1,3-dimethylcyclopentane is 45%, while 50% is predicted. Several difficulties, however, are encountered in applying this mechanism to C<sub>4</sub> and C<sub>5</sub> rings, since it does not explain the greater difficulty to open a C<sub>4</sub> ring. The ability of the C atoms to rotate in C<sub>4</sub>, which is absent in C<sub>3</sub>, C<sub>4</sub>, and C<sub>5</sub> rings, may serve as explanation.

Andrew Dravnick

RAIK, S. Ye.

Verbatim: - "The extension of the theory of multipoles to the reaction of the dehydrocyclization of substituted pentanes and partial hexanes," Vestnik Mosk. un-ta, 1948, No. 12, p. 112-20, - Bibliog: 11 items

SC: U-4355, 14 August 53, (Letopis 'Khurnal 'nykh Statey, No. 15, 1949.)

CA

Use of the statistical method for the determination of the results of some contact reactions. S. E. Raik (Moscow State Univ.). *Vestn. Moskov. Univ.*, 6, No. 10, Ser. Fiz.-Mat. i Estestven. Nauk No. 6, 69-78 (1951). — (1) The no. of events sufficient for the probability that one single event will be applicable, within a small stated uncertainty, to a large no. of events, is estd. with the aid of the principle of practical assurance. The problem is to det. the no.  $n$  of elementary acts of a given catalytic reaction necessary for the statistical probability to differ from the probability derived on the basis of the mol. structure, by a magnitude  $\epsilon$ , not to exceed the error. For reactions where the products can be distd. to within 0.01%,  $\epsilon$  is of the order  $10^{-6}$ . With the aid

of Laplace's functions,  $\epsilon = (4.48/\epsilon)^{1/2}pq$ , where  $p$  and  $q = 1 - p$  are the elementary probabilities of the alternative outcomes of a ring-opening reaction. As an example,  $p$  and  $q$  correspond to: methylcyclobutane  $\rightarrow$  pentane ( $p = 0.14$ ) + 2-methylbutane ( $q = 0.86$ ); methylcyclopentane  $\rightarrow$  hexane + 3-methylpentane (0.33) + 2-methylpentane (0.67); 1,1-dimethylcyclopentane  $\rightarrow$  3,3-dimethylpentane (0.25) + 2,2-dimethylpentane (0.75); 1,2-dimethylcyclopentane  $\rightarrow$  3-methylhexane (0.25) + 2,3-dimethylpentane (0.75); 1,3-dimethylcyclopentane  $\rightarrow$  (3-methylhexane + 2-methylhexane) (0.60) + 2,4-dimethylpentane (0.40). In all these instances,  $n$  is of the order of  $10^6$  mols., which corresponds to  $\sim 10^{-12}$  g. of substance undergoing reaction. Consequently, the application of *a priori* probabilities under the usual conditions of catalytic expts. is fully justified. (2) Probabilities of cyclization of hexanes are estd. on the assumption that the hexane is adsorbed by one C atom, and cyclization occurs if the hexane folds in such a way that the 6th C atom comes into close contact with the C atom adsorbed. On account of free rotation around C-C bonds, there is a definite finite probability  $P$  that this will happen. The probability that the 1st 3 C atoms lie in one plane is taken to be a certainty. The probabilities that the 4th and the 6th C atoms will take positions favorable to ring closing are equal, and are shown to be  $1/2$  each, i.e. the probability that both take such positions is  $1/4$ . For the 6th C atom,

over

the probability is  $1/n$ , and hence the total  $P = 1/n$ . That is the probability that a mol. will possess a configuration favorable for the closing of a 6-membered ring at the moment of adsorption of one of its C atoms. The fraction  $P = 1/n$ , multiplied by the probability of adsorption of a suitable C atom, gives the probability  $\rho$  of cyclization. (3) The no. of contacts  $c$  between a mol. susceptible of cyclization, and the catalyst, which leaves a stated fraction  $\beta$  of the reactant uncyclized, can be expressed by  $\eta' = \beta$ , or by  $\eta' = \gamma \leq \beta$ , where  $c$  is now the smallest integer satisfying the inequality. These relations permit, with a known  $\eta'$  and a prescribed  $\beta$  (or  $\gamma$ ), the calcn. of  $c$  and hence, if an assumption is made with regard to the no. of active points per unit amt. of catalyst, also the min. amt. of catalyst  $g$  necessary for the cyclization of a given amt. of hydrocarbon. As an example, if for catalysts of the type of  $\text{Al}_2\text{O}_3 + 10\%$   $\text{ZnO}$ , or bauxite,  $\sim 2 \times 10^6$  active points/g. catalyst, in order to cyclize 0.1 ml. hexane ( $4.85 \times 10^{-6}$  mols.) down to  $\beta = 10^{-3}$ ,  $c = 754,15$  contacts and, hence, 21430 g. catalyst is necessary. This amt. decreases if the prescribed  $\beta$  is made smaller; e.g., for  $\beta = 10^{-4}$ ,  $c = 2008$  g. The same dependence of the fraction reacted on the probability is expressed by the formulas,  $\eta' = \beta$  and  $K_\rho = \log [1/(1 - \alpha)]$ , where  $\alpha = 1 - \beta$ . The exptl. yields of cyclization, from data of Kazanskii and Plata (C.A. 33, 9294<sup>a</sup>) on Pt, and of Hoog, Verheus, and Zuiderveld (C.A. 33, 9293<sup>b</sup>) on  $\text{Cr}_2\text{O}_3$ , can be represented equally well by either of these formulas. The factor dets. yields of aromatization is the probability of cyclization of the given hydrocarbon. N. Thon

MORDKOVICH, M.S.; RAIK, S.Ya.; ARASIMOVICH, V.V.

Losses of pectin substances in the production of tomato paste.  
Kons. i ov. prom. 18 no.11:19-21 N '63. (MIRA 16:12)

1. Moldavskiy nauchno-issledovatel'skiy institut pishchevoy  
promyshlennosti (for Mordkovich). 2. Institut fiziologii i  
biokhimii rasteniy AN Moldavskoy SSR (for Raik, Arasimovich).

RAIK, S.Ye.; KAZANSKIY, B.A.

Production of ethyl ester of cyclobutane-1, 1 dicarboxylic acid. Vest.Mosk. un. no.3:125-128 Mr '53. (MLRA 6:6)

1. Laboratoriya organicheskogo kataliza.

(Ethyl esters)

Raika, E.

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9"

The pathogenesis of urticarial pruritus. III. Quantitative evaluation of the effects of various drugs on morphine-induced pruritus. E. Raika, S. Korossy, and Marianne Gózony (Stevens' Hosp., Budapest). *Dermatologica* 112, 81-107(1960)(in German)(English summary); cf. *C.A.* 50, 1178b.—Of 78 drugs, of various types, tested (3008 tests) 44 were found, in the majority of cases, to prevent the occurrence of exptl., morphine-induced, pruritus. In no case was 100% inhibition obtained. Henry B. Hastie

3  
Med

RAIKH, I.Ya., inzhener; FAYNGERSH, Ya.D., inzhener.

Mechanical method of making openings in masonry walls. Mekh.stroi.  
11 no.11:28-29 N '54. (MIRA 7:12)  
(Masonry) (Drilling and boring)

11/1  
Determination of the spectral energy distribution of light emitted by an incandescent lamp with voltage stabilizer when the intensity was subject to continuous change produced by the opening and closing of the slit of the spectrograph by means of a constant speed motor.

778.3 : 535.33

531  
**Photometry of Spectra of Variable Intensity and of Spectral Lines of Low Intensity.**  
J. D. RAIKHBAUM. *Bull. Acad. Sci. U.R.S.S., Ser. Phys.*, 9, 765-768, 1945.—In SLAVIN's "spectral energy" method—complete vaporization of the sample in an electric arc—it is necessary to understand the laws governing the integration by the photographic plate of the varying energy of emission. An account is given of the results of experiments in the photometry of continuous spectra from an incandescent lamp with voltage stabilizer when the intensity was subject to continuous change produced by the opening and closing of the slit of the spectrograph by means of a constant speed motor. *Brit. Abs. C.*

*m* //

<sup>10</sup>Spectrochemical Analysis of Alloys of Gold, Silver, Copper, and Tellurium.  
V. D. Raiklinam, L. D. Raiklinam, and M. G. Vinogradov (Zavod  
*Labs.*, 1940, 9, 104-108; *Brit. Chem. Abstr.*, 1944, 1C, 5). (In Russian.)  
Au-Ag alloys are evaporated in an arc, and the total amounts of the radioisotopes Au 232-7 and Ag 2300-4 Å compared, using Betini's method (Joupol  
*Anal.*, 1931, 106, 580). These amounts can be compared by using a logarithmic sector or by microphotometry.

ASA SLA METALLURGICAL LITERATURE CLASSIFICATION

<sup>111</sup> Spectral analysis of alloys of gold, silver, copper and tellurium. Ya. D. Rakhlinbaum, L. D. Rakhlinbaum and M. G. Veskobrodnikov. *Zaradzhayushch. Lab.*, No. 10(1941). Reptl. details for spectral data of Ag, Cu and Te in Au nuggets of 3-5 mg. B. Z. Kamach

7

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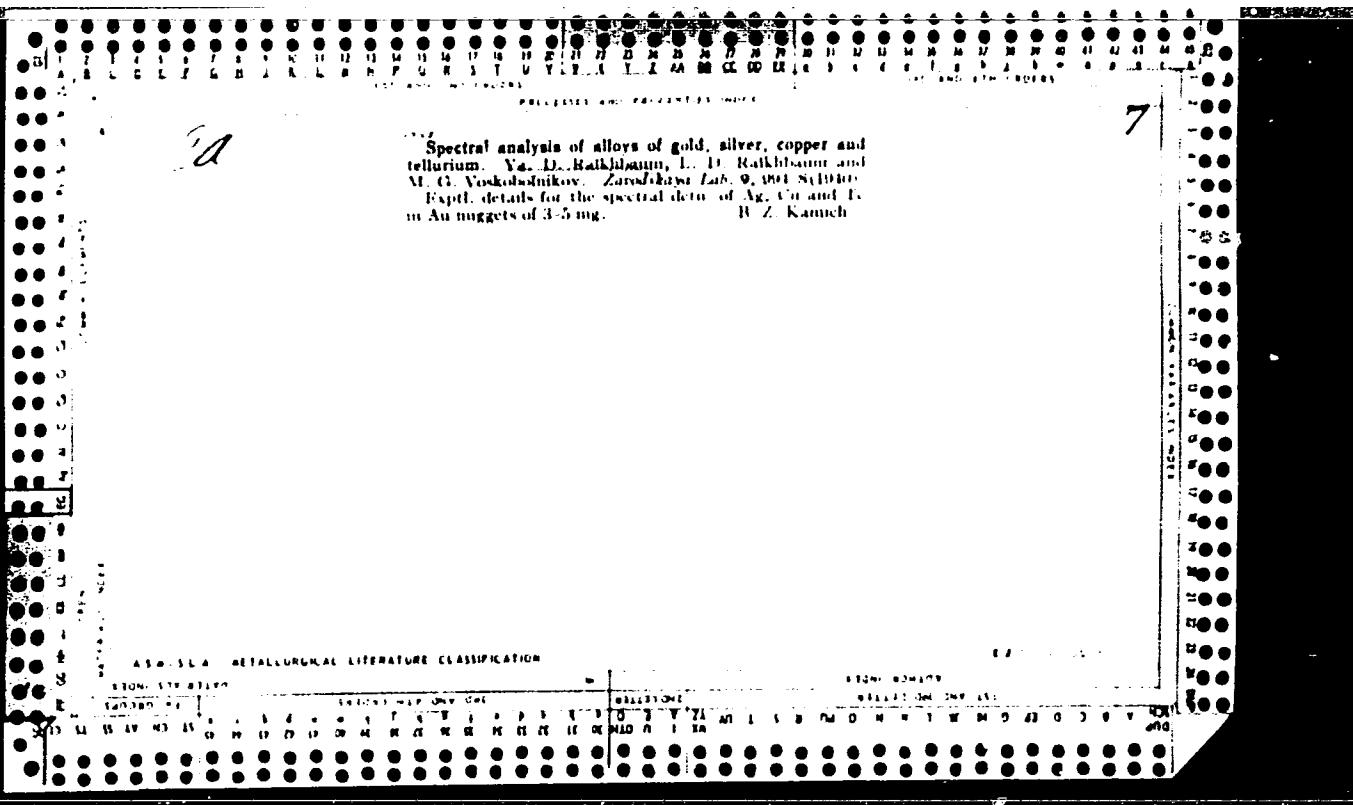
CIA-RDP86-00513R001344020008-9"

11  
M

**Spectrochemical Analysis of Alloys of Gold, Silver, Copper, and Tellurium.**  
Y. D. Raikhlbaum, L. D. Raikhlbaum, and M. G. Vaykobsonikov. *Zurnal  
Fizika*, 1940, **9**, 101-108; *Brit. Chem. Abstr.*, 1941, (C), 51. (In Russian)  
Au-Ag alloys are evaporated in an arc, and the total amounts of the radia-  
tions Au 23527 and Ag 23001 A. compared, using Betens's method (*Compt.  
rend.*, 1934, **198**, 500). These amounts can be compared by using a log-  
arithmic sector or by microphotometry.

ASA-SEA METALLURGICAL LITERATURE CLASSIFICATION

EZ



Determination of tungsten in minerals by the comparison of spectral energies. Yu. D. Ralikhman. Zavodskaya Lab. 8, 601-5 (1939).—W was detd. from the relative values of the spectral energies corresponding to 2 spectral lines of different elements. Co was added to the soln in const. concn, and the selected pairs of lines were photometered to det. the relative values of the energies emitted by W and Co during complete vaporization in an arc of d. c. The nature of the chem. bond of W had no effect on the results. The varying thermal stabilities of the different W compds. govern only the rate of vaporization of the W atoms. Vaporization of W and Co is hindered by presence of large amounts of NaCl but it becomes more intensive as the temp. of the arc rises and the NaCl is vaporized.

ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION

Local spectral analysis of Au surfaces. Ya. D. Raikhan. *Zavodskaya Lab.*, 10, 108-70(1941); *Chem. Zentr.* 1943, II, 1118; cf. *C.A.*, 37, 22977. — Local spectral analysis offers a means of supplementing the mineralogical and quant. spectral analysis of Au ores. It is possible to det. the qual. compn. of films of Au plate. Only with samples of less than 0.5 mg. were unsatisfactory results obtained. In these cases the samples were dissolved in acid and the solns. analyzed. M. G. Moore

7

## AS-A-12A METALLURGICAL LITERATURE CLASSIFICATION

TA

Sensitivity & Spectrometry

771.534.56

187  
Photographic Action of Light of Variable Intensity. YA D. RAIKHBAUM. J.  
Tech. Phys. U.S.S.R., 15, 485-8, 1945.--Photographic plates are illuminated  
with light of a gradually increasing or gradually decreasing intensity. The  
optical density of the image at a constant exposure is, for an increasing intensity,  
smaller than for a decreasing one and greater than for a constant intensity  
and variable time. When the rate of variation of light intensity rises, density  
rises also. These results must be borne in mind when using density in spectro-  
graphic analysis. *Brit. J. Phot. Abs.*

F-A

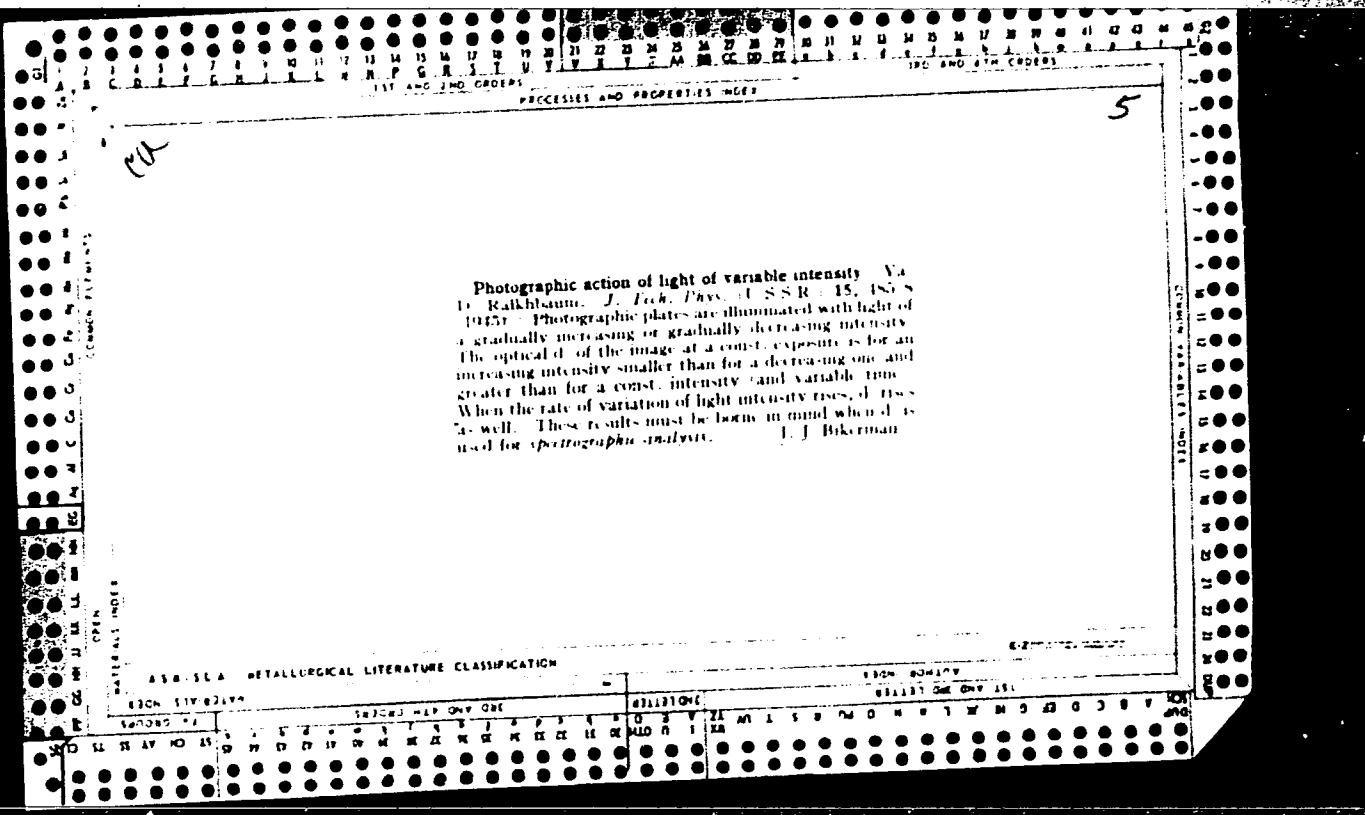
Preparation of Light-Sensitive  
M. T. and Spence

1145

77.021.1 : 544.62

Spectrographic Determination of Silver in Photographic Emulsions. Ya. D. RAIKHEAUM and Ya. M. DYMISHITS. Izvest. Akad. Nauk. S.S.R., Ser. Fiz., 1948, 12, 477-480.—Samples of the order of 0.04-0.1 sq. cm. of the emulsion are calcined and thoroughly mixed with a known amount (100-200 mg.) of a mixture of 25 mg. of anhydrous sodium sulphate, 25 mg. of anhydrous potassium sulphate, 50 mg. of graphite and 0.05 per cent of tin as standard. The determination is done in an a.c. arc under 10 amp., by the line pair Ag. 3280.68-Sn 3262.33A. The log calibration curve is rectilinear between 0.0002 and 0.02 per cent silver. The probable error of a single determination is 8.4 per cent. The method was used to determine for various types of emulsions the photometric constants and the mean mass of the developed grains.

Chem. Abs.



Quantitative Spectral Analysis of Tin, Lead, and Bismuth Alloys. Ya. D. Atakhbaum (*Zurnal Lab.*, 1939, **8**, 1101-1105; *Chem. Zentral.*, 1942, **113**, (11), 1042; *C. R. Acad.*, 1943, **37**, 5332). —[In Russian.] The mathematical basis of spectral analysis is explained and various curves are given which were obtained by experiments with the Hilger spectrograph. Results accurate to within about 1% were obtained even with high concentrations of Sn, Pb, and Bi.

REF ID: A6510

ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION

E-2-12-14-1

Spectrographic determination of silver in photographic emulsions Ya. D. Rakhbaum and Ya. M. Dvinskis  
*Izv. Akad. Nauk SSSR*, Ser. Fiz., 12, 177 (1948).  
Samples of the order of 0.01-0.1 sq. cm. of the emulsion are calcined and thoroughly mixed with a known amount (100-200 mg.) of a mixt. of Na<sub>2</sub>SO<sub>3</sub> 25, K<sub>2</sub>SO<sub>3</sub> 25, graphite 50, and Sn 0.05%, as standard. The determin. is done in an arc, 4000-4500 A, under 10 amp., by the line pair Ag 3280 & 3880 (3223.3 Å). The log-log calibration curve is rectilinear between 0.0002 and 0.02% Ag. The probable error of a single determin. is 8.4%. The method was used to determine the photometric constants and various types of emulsions, the photometric constants and the mean mass of the developed grains. N. Thom.

RAIKHER, G. S.

Gruzooborot transporta v novoi piatiletke. Freight turnover in the new five-year plan (Zhel-dor. transport, 1946, no. 2-3 p. 33-41). "Additional details on freight traffic by region and commodity. Supplements Kovalev's pamphlet.

DLC: HE7.Z5

Gruzovye perevozki. The freight transport. (In Levin, B.I. Osnovnye voprosy piatiletnego planavosstanovleniya i razvitiia zheleznodorozhnogo transporta. Moskva, 1947, p. 111-136). DLC: HE3137.L4

Osvobodit' zheleznye dorogi ot korotkoprobezhnykh perevozok. To free the railroads from short distance shipments. (Zhel-dor. transport, 1945, no. 10 11, p. 70-73).

DLC: HE7.Z5

Ratsionalizatsiia zhelezhodorozh nykh perevozok v tret'em piatiletii. Rationalization of railroad shipments in the third five-year plan (Problemy ekonomiki, 1940, no. 2, p. 88-97)

DLC: HB9.P75

Za bor'bu s neratsional'nyimi korotkoprobezhnymi perevozkami. Against inefficient short distance shipments (Sots. transport, 1940, no. 3, p. 8-11) "1935 and 1938 data on proportion and amount of such shipments.

DLC: HE7.S6

SO: Soviet Transportation and Communication A Bibliography, Library of Congress Reference Department, Washington 1952, Unclassified

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CIA-RDP86-00513R001344020008-9"

*22*

## CHARACTER AND PROPERTY INDEX

100 AND 4TH DECADE

**Characteristics of  $\alpha,\beta$ -unsaturated ketones. II.** V. I. Krasov and I. I. Rakhier. *J. Gen. Chem. (U. S. S. R.)* 13, No. 13 (1943) (English summary); cf. *C. A.* 35, 30389. It was shown that the displacement of ketones by aldehydes from  $\alpha,\beta$ -unsatd. ketones is connected with the preliminary hydrolytic cleavage of the latter into its components. BaII, treated with mesityl oxide in the presence of dil. NaOH, yielded small amounts of benzylideneacetone and benzylidenemesityl oxide, the latter being isolated as the tetra bromide, m. 118° (from R(ONa)). The reactions were conducted by prolonged standing at about 10°, with a longer period noticeably increasing the yield of benzylideneacetone. Me<sub>2</sub>CrO was shown to be inert to condensation on treatment with piperidine acetate on prolonged heating on a steam bath; the same result was obtained when MeEtCO was used. However, 25 g. BaII, 30 g. MeCO, 5 g. piperidine and 5 g. AcONa yielded, after 20 hrs. heating, 71% benzylideneacetone, m. 41.5°, while similar reaction with mesityl oxide gave crude benzylidenemesityl oxide, b.p. 172-3°, identified as the tetrabromide. Thus, piperidine acetate may be recommended for aldehyde-ketone condensations.  
G. M. Kosolapoff

10

## ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION

ECONOMIC SECTION

ECONOMIC SECTION

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ALL INFORMATION CONTAINED  
HEREIN IS UNCLASSIFIED

1. [REDACTED] (Top Secret) (SAC) (CIA)  
[REDACTED]

1. Anti-aircraft units. 2. Fire study. 3. Armor II, L.D., [REDACTED]

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9"

RAIKHER, M.E., professor; KAMINSKIY, I.N., inzhener; NIKOL'SKIY, V.S.,  
redaktor; SUROVA, V.A., redaktor; ANDREYEV, G.G., tekhnicheskij  
redaktor.

[Complex time study in coal mines and pits] Kompleksnyi khronometrazh  
na ugol'nykh shakhtakh i kar'erakh. Moskva, Ugletekhizdat, 1954.  
203 p.  
(Time study) (MIRA 8:5)

RAIKHER, M.E.; KAMINSKIY, I.N.

[Complex timekeeping in coal mines and pits]. Kompleksnyi khronometrazh na ugel'nykh shakhtakh i kar'erakh. Moskva, Ugletekhnizdat, 1954. 204 p. (MLRA 8:3D)

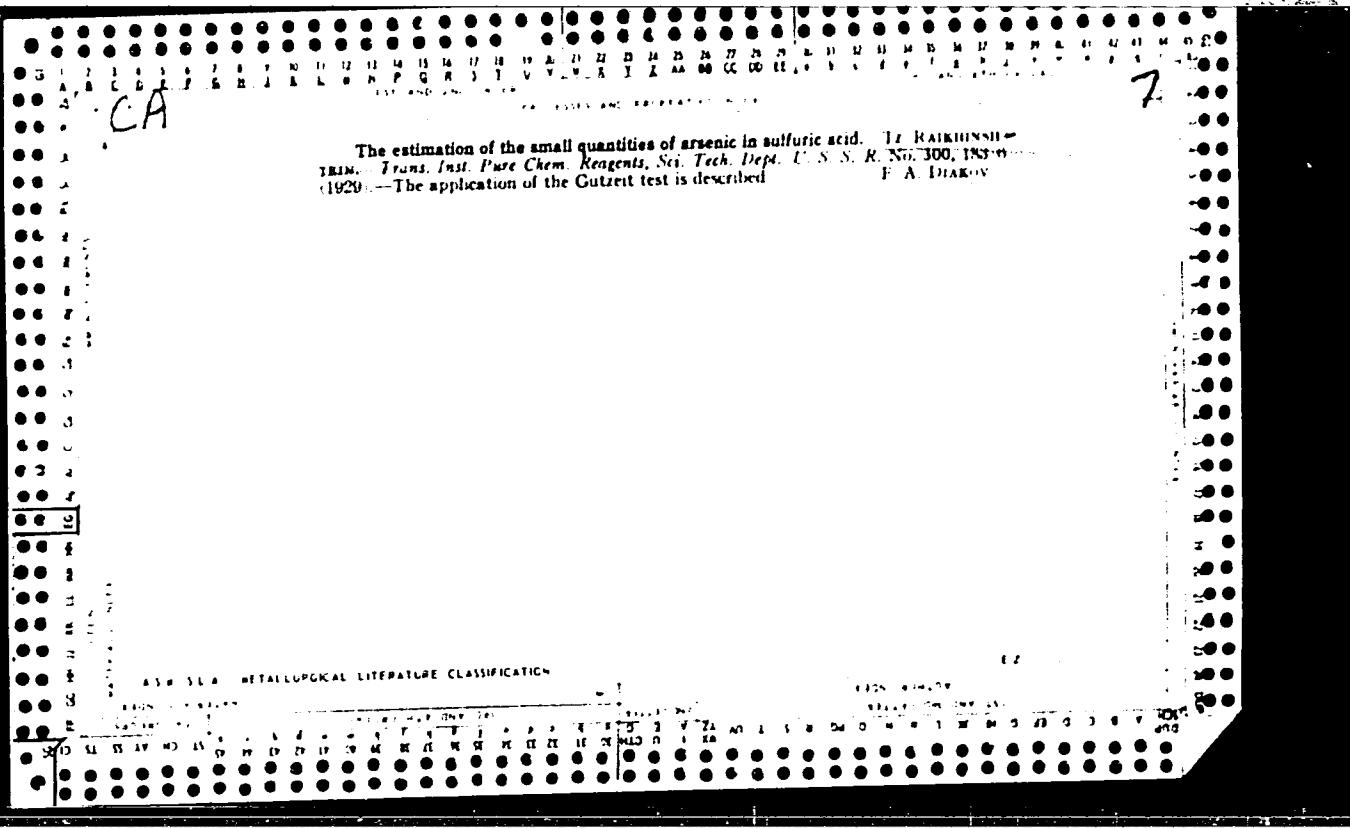
Determination of molybdenum and lead by Fajans method. T. Rakhmishstein and N. N. Konchalov. *J. Russ. Chem. (U.S.S.R.)* 3, 531 (1933); cf. *C.A.* 27, 2300. --A comparison between the methods of Fajans, Wiley, Sacher, Chandler and Alexander and Wiley, Ambrose and Bowers, for the detn. of Mo and Pb, showed that Fajans' method is the most rapid and accurate.  
S. I. Maderovsky

AMSLA METALLURGICAL LITERATURE CLASSIFICATION

Determination of molybdenum and lead by Fajans  
method. I. Rakhlinstein and S. N. Korobov.  
Gen. Chem. U.S.S.R., 1931, 14, 27.  
280. —A comparison between the methods of Fajans,  
Wiley, Sucher, Chandler and Alexander and Wiley,  
Bruce and Bowers, for the detn. of Mo and Pb, showed that  
Fajans' method is the most rapid and accurate.

S. I. Midorsky

AMERICAN METALLURGICAL LITERATURE CLASSIFICATION

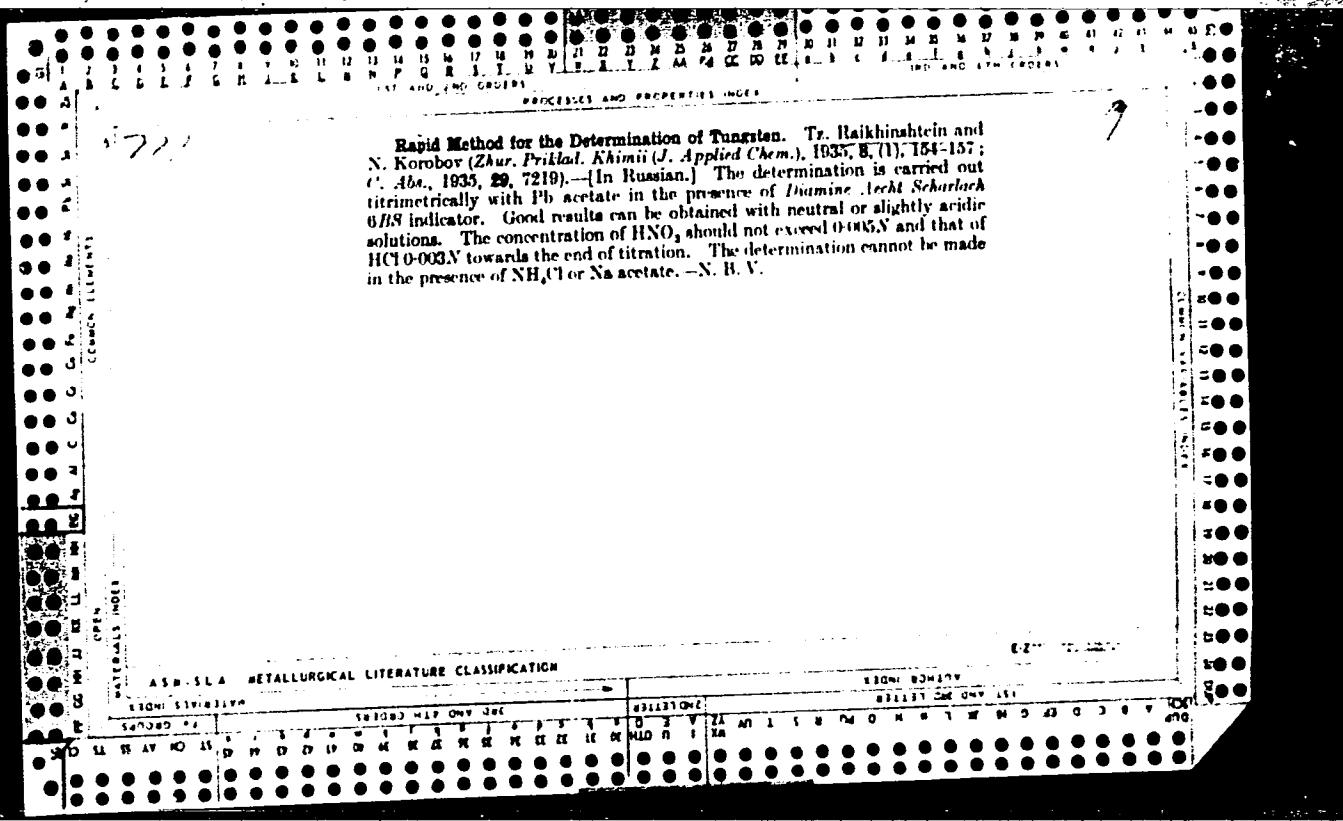


1ST AND 2ND GROUPS  
PROCESSES AND PROPERTIES INDEX

Determination of molybdenum and lead by the method of Fajans. I. Tz. RAI-KHIMSKII AND N. KOROBOV. *J. Gen. Chem. (U. S. S. R.)* 2, 601-5(1932) —  
 (C. A. 15, 2225; 17, 3633, etc.) developed a method for the volumetric detn. of Ag halides with the application of adsorption indicators, which forms the basis of the present method for the detn. of Pb and Mo.  $(\text{AcO})_2\text{Pb}$  in  $\text{H}_2\text{O}$  gives with a little alizarin red (I) a violet soln., and with an excess a violet ppt. of the Pb salt of I; on addn. of  $(\text{NH}_4)_2\text{MoO}_4$  to the soln. the ppt. of  $\text{PbMoO}_4$  formed completely absorbs, at an adequate concn. of I, the Pb salt of I with violet color until the equiv. amt. of  $(\text{NH}_4)_2\text{MoO}_4$  is added. With the slightest excess of  $(\text{NH}_4)_2\text{MoO}_4$  the anion  $\text{MoO}_4^{2-}$  displaces the anion of I, and the color of the ppt. is sharply changed to rose with orange tint. Any further addn. of  $(\text{NH}_4)_2\text{MoO}_4$  does not affect the color of the ppt. In the expts. 0.5 cc. 1% I was added to the soln. of  $\text{Pb}(\text{NO}_3)_2$  or  $\text{Pb}(\text{OAc})_2$ , and the soln. was brought to a boil and titrated with a standard soln. of  $(\text{NH}_4)_2\text{MoO}_4$ . For the detn. of Mo, an excess of standard soln. of Pb salt was added to the soln. of a molybdate, the soln. was brought to a boil, 0.5 cc. 1% I was added, and the excess of Pb titrated back with a standard molybdate soln. The standard solns. of  $\text{Pb}(\text{NO}_3)_2$  and  $\text{Pb}(\text{OAc})_2$  were prep'd. by adding aq.  $\text{NH}_3$  to the ppt. of  $\text{Pb}(\text{OH})_2$  just forming and dissolving the ppt. with AcOH. The solns. used were 0.1235 N, 0.1133 N and 0.0151 N with concns. of AcOH of 0.1-0.4 N. The vol. of all titrated solns. was 40 cc. The titer of Pb solns. was detd. as  $\text{PbSO}_4$  and checked by electrolysis, and that of  $(\text{NH}_4)_2\text{MoO}_4$  by evapg. to dryness and gently igniting in a Pt dish. Good results were obtained at different concns. of the titrated solns., but with the concns. of a molybdate below 0.01 N the amt. of I should not exceed 0.5 cc. 1% soln. The method can be used for the detn. of oxides of Pb by dissolving them in  $\text{AcOH}$  or  $\text{HNO}_3$  and neutralizing with  $\text{NH}_3$ , in which case the concn. must be below 1 N for  $\text{NH}_4\text{OAc}$  or 2 N for  $\text{NH}_4\text{NO}_3$ . Titration of Pb and Mo by this method in the presence of the acetates and nitrates of heavy metals and other elements which form difficultly sol. salts with  $(\text{NH}_4)_2\text{MoO}_4$  is impossible.

CHAR BLANC

## ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION



Oxidation reduction indicators. I. New indicators for  
the titration of tin and antimony by bromates. II.  
G. Rakhmashin. *J. Applied Chem. U.S.S.R.*, 6,  
1179 (in German). 14,500 R.R.P. Suppl. to *Izdat. in  
USSR*. HCl solution, 10 wt.-%, with bromopurpurin III or  
III-A. HCl solution, 10 wt.-%, with bromopurpurin IV  
bromate in 14.5% HCl soln. with bromopurpurin IV  
bromate in 14.5% HCl soln. with bromopurpurin IV  
H. M. Proc. 6.

EXCERPTS AND REFERENCES INDEX

Potentiometric determination of sulfates. I. I.  
Zhukov and Tz. G. Rakhimdzieva. *J. Gen. Chem.*  
(U. S. S. R.) 4, 802-8 (1934).—As the foundation for  
this detn. Kolthoff's method based on the use of Pb  
tri-ferrocyanide as an indicator electrode was used.  
Portions of  $\text{Na}_2\text{SO}_4$ , soln., were treated with alc., some  
 $\text{K}_3[\text{Fe}(\text{CN})_6]$ , and some  $\text{K}_3[\text{Fe}(\text{CN})_6]$ , and titrated with  
 $\text{Pb}(\text{NCH}_3)_4$ , soln. V. D. Karpenko

ASA SEA METALLURGICAL LITERATURE CLASSIFICATION

ARISTOV, Grigoriy Andrianovich; RAKHLIN, I., redaktor; SHIVELIEVA, A.,  
redaktor; IGHAT'YEVA, A., tekhnicheskiy redaktor

[Infinite universe] Vselennaya beskonechna. [Moskva] Mo-  
skovskii rabochii, 1955. 110 p. (MIRA 9:3)  
(Cosmogony)

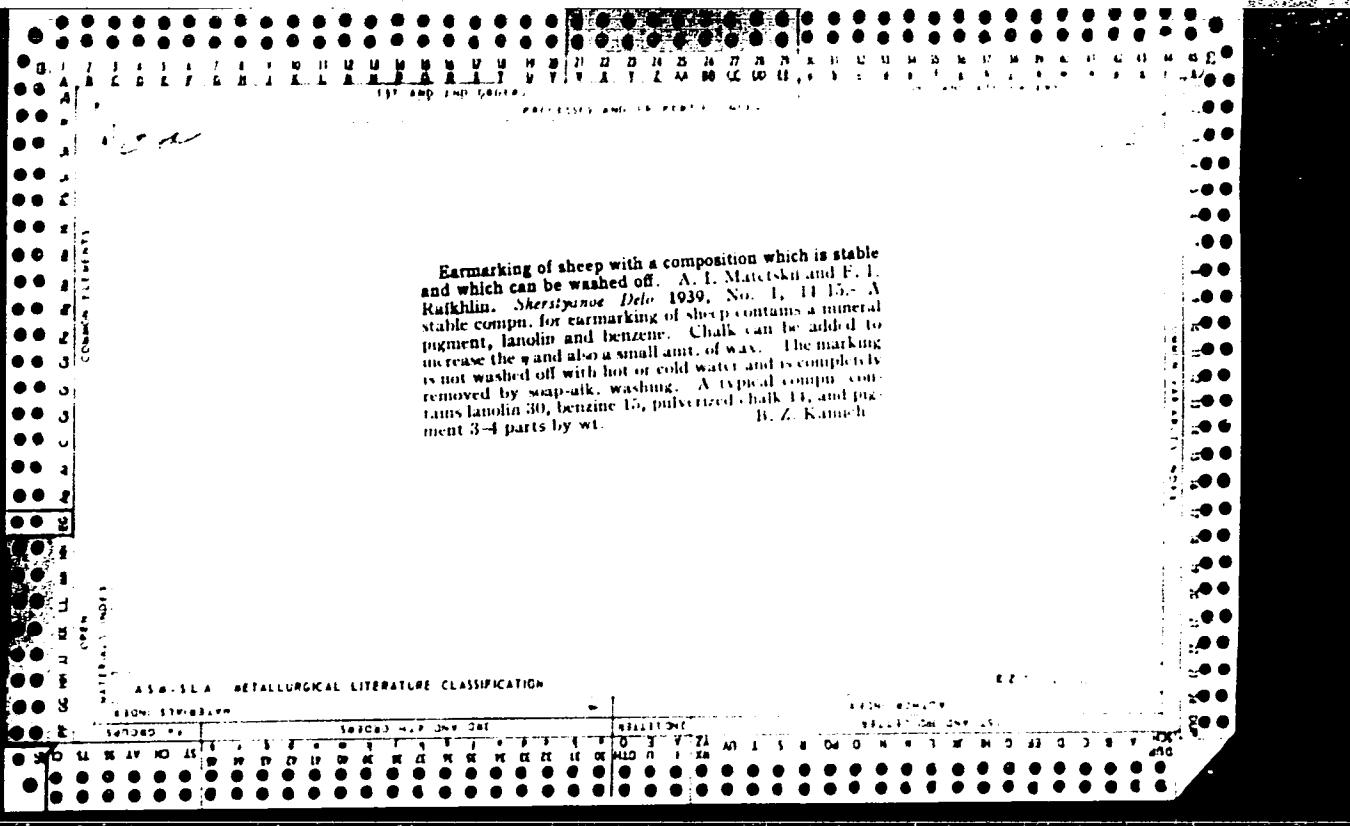
PAREKHO, Pavel Petrovich; RAKHLIN, I.Ye., redaktor; YARMAKVA, Ye.a.,  
tekhnicheskiy redaktor

[In the world of stars] V mire zvezd. Moskva, Gos.izd-vo tekhniko-  
teoret. lit-ry, 1957. 95 p. (Populiarnye lektsii po astronomii,  
no.5) (MLRA 10:1C)  
(Astronomy)

*Ca*

Use of cationic soaps. A. I. Matetskii and F. I. Rakshin. *Tekstil. Prom.* 1941, No. 5, 35-6; *Chem. Zentr.* 1943, I, No. 4, 463.—Cationic soaps were prep'd. from cetyl and octodecyl ales. and mixts. of high-mol. ales. produced from cottonseed oil and seal oil. The following compds. were used as bases: pyridine, pyridine bases (b. p. 142-53°), trimethylamine, diethylamine. Expts. in the washing of wool and dyeing of cotton- and half-wool material showed cationic soaps are valuable products, and their use in the dyeing and finishing of fiber is desirable.  
Sonya G. Machelson

ASME U.S. METALLURGICAL LITERATURE CLASSIFICATION



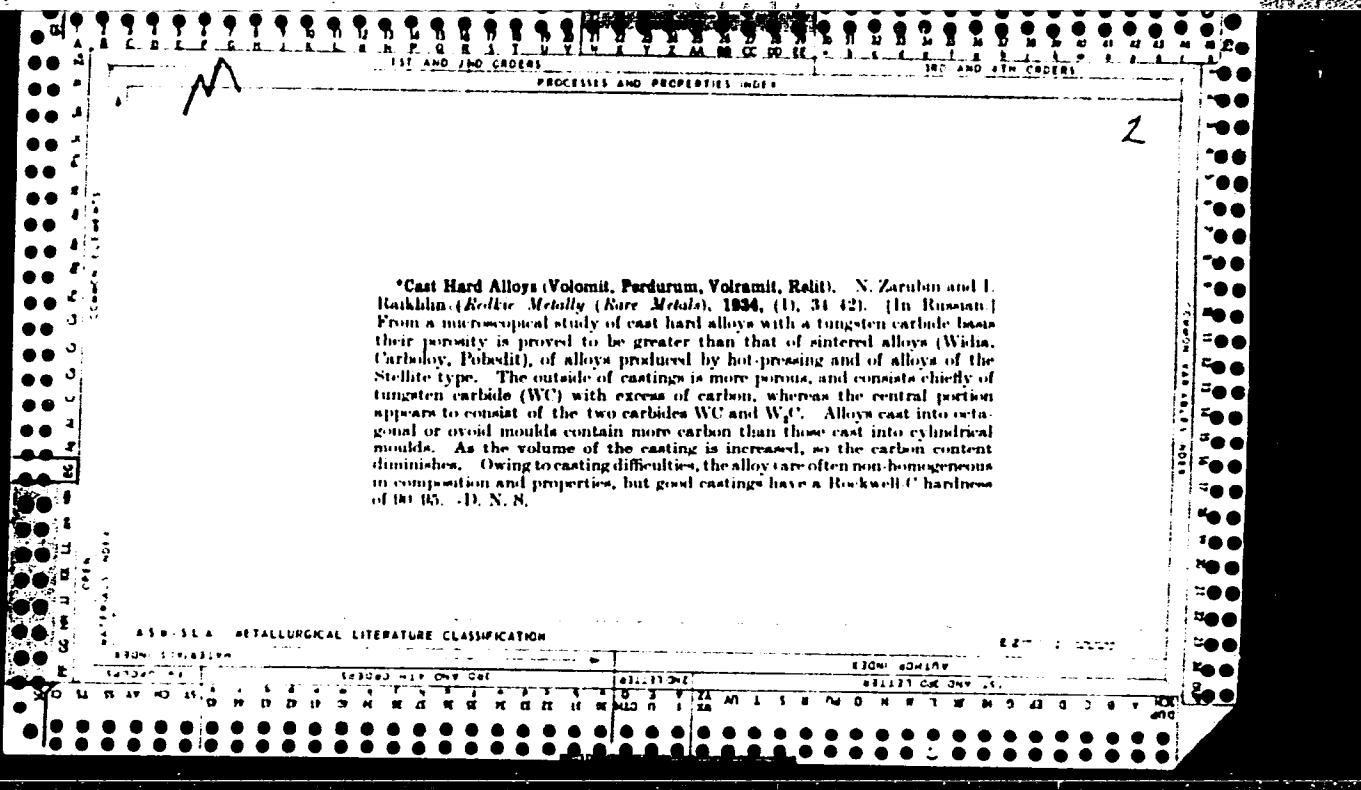
CA

22

**Stabilization of mineral-oil emulsions.** I. M. Rakhlin,  
*Tekstil. Prom.*, 7, No. 2, 34-5(1947). In an effort to  
eliminate olein from wool-scouring emulsions, emulsions  
of solar oil and spindle oil, together with Kontakt (sul-  
fonated naphthenic acid) and Nekal (butylnaphthalene-  
sulfonate) detergents were tested for stability. An emul-  
sion of 20-25% spindle oil, 8% olein, 4% NH<sub>3</sub> and 68-63%  
H<sub>2</sub>O was stable after 240 hrs. Substitution of solar oil in  
this formula reduced this time to 24 hrs. An emulsion  
contg 3.5% Nekal gave 240-hr. stability with 20-30%  
spindle oil and 77-67% H<sub>2</sub>O. Substitution of solar oil  
decreased time of stability to 24 hrs. Increasing Nekal  
content to 7-10% reduced stability to 3 hrs. An emulsion  
consisting of 4% Kontakt, 20% spindle oil, 1% Na<sub>2</sub>CO<sub>3</sub>,  
and 75% H<sub>2</sub>O was recommended for use.  
M. S.

1000

ASCE LIBRARY METALLURGICAL LITERATURE CLASSIFICATION									
SIGHT INDEX					ECON. INDEX				
SEARCHED	INDEXED	FILED	REF'D	SEARCHED	INDEXED	FILED	REF'D	SEARCHED	INDEXED
SEARCHED	INDEXED	FILED	REF'D	SEARCHED	INDEXED	FILED	REF'D	SEARCHED	INDEXED



RAIKHLIN, N.T. (Moskva)

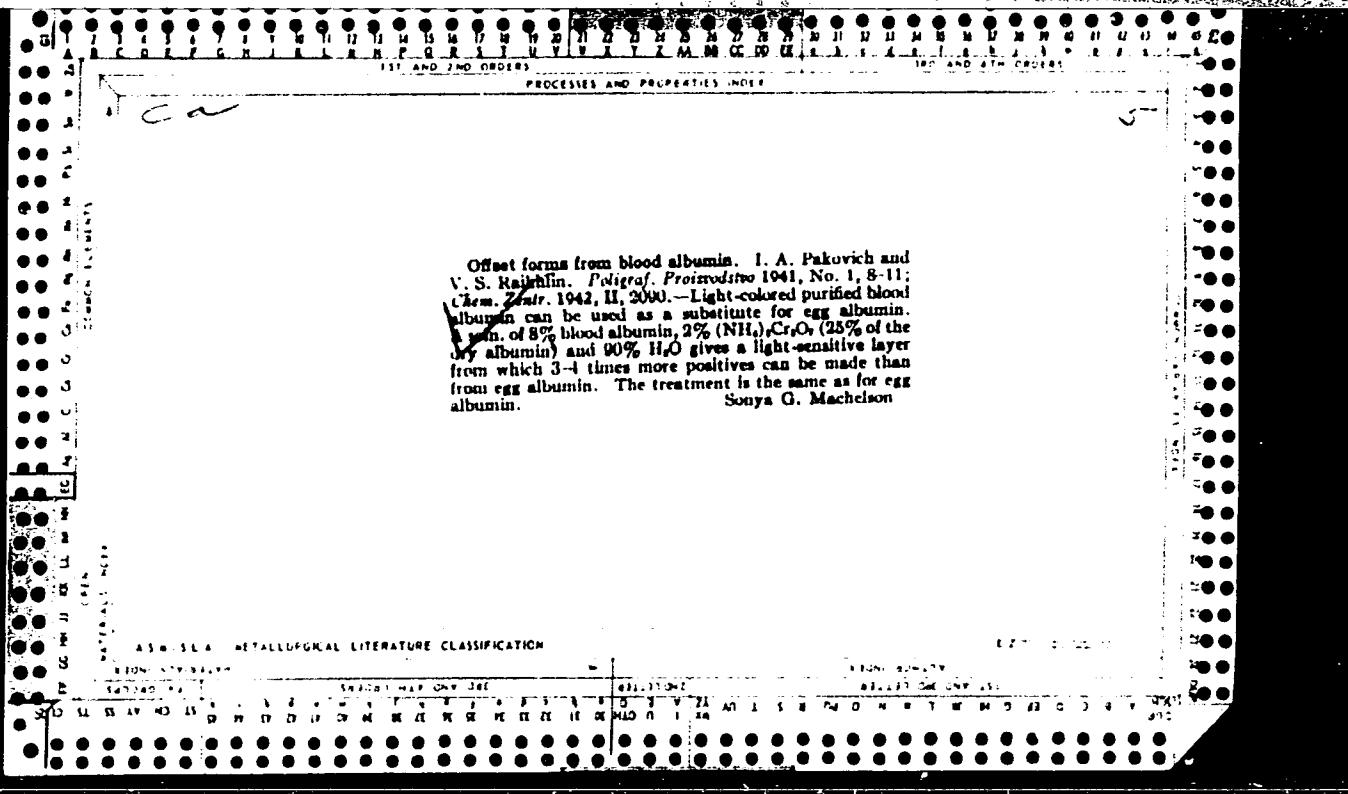
Compensatory adaptation modifications in the lungs in pneumoconioses.  
Arkh.pnt. 18 no.3:18-23 '56  
(MIRA 11:10)

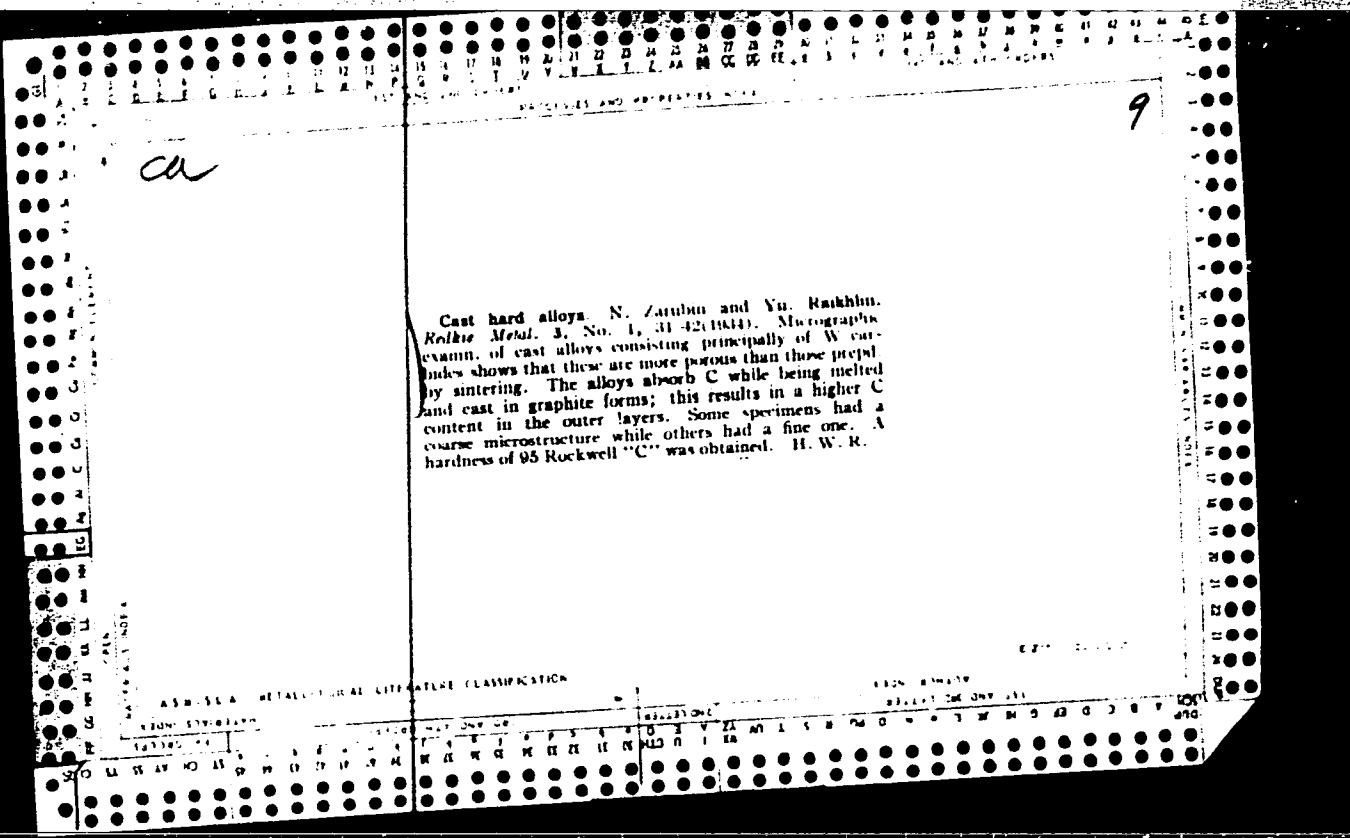
1. Iz kafedry patologicheskoy anatomi (zav. - chlen-korrespondent  
AMN SSSR prof. A.I. Strukov) I Moskovskogo ordena Lenina meditsinskogo  
instituta imeni I.M. Sechenova.  
(PNEUMOCONIOSES, physiol.  
compensatory adjustment modifications of lungs (Rus))

*CH* 7

Rapid method for determining tungsten. D. Raskin, J. S. Kostyuk. *J. Applied Chem.* (U.S.S.R.) 8, 283-7 (1958). The determination is carried out with phenacetate (ultrometrically) in the presence of "diamine azo茜素" indicator. Good results can be obtained with neutral or slightly acidic solutions. The concn. of HNO<sub>3</sub> should not exceed 0.005 N and that of HCl 0.03 N toward the end of titration. Na and NH<sub>4</sub> nitrate are permissible in higher concns. The determination cannot be made in the presence of NH<sub>4</sub>Cl or CH<sub>3</sub>COONH<sub>4</sub>. This method may also be used for deterg. W (in W solns.). Nine references.  
A. A. Raskin, 3

434 SLA METALLURGICAL LITERATURE CLASSIFICATION





## MEDICAL AND INDUSTRIAL EMPLOYMENT.

Dr. E. I., Ed., Professor

...olodye redkih metallov (Toxicology of Rare Metals);  
...nia 1953. 335 p. 1500 copies printed.

A. A. Khanidullin; Tech. Ed.: Yu. S. Bel'chikova.

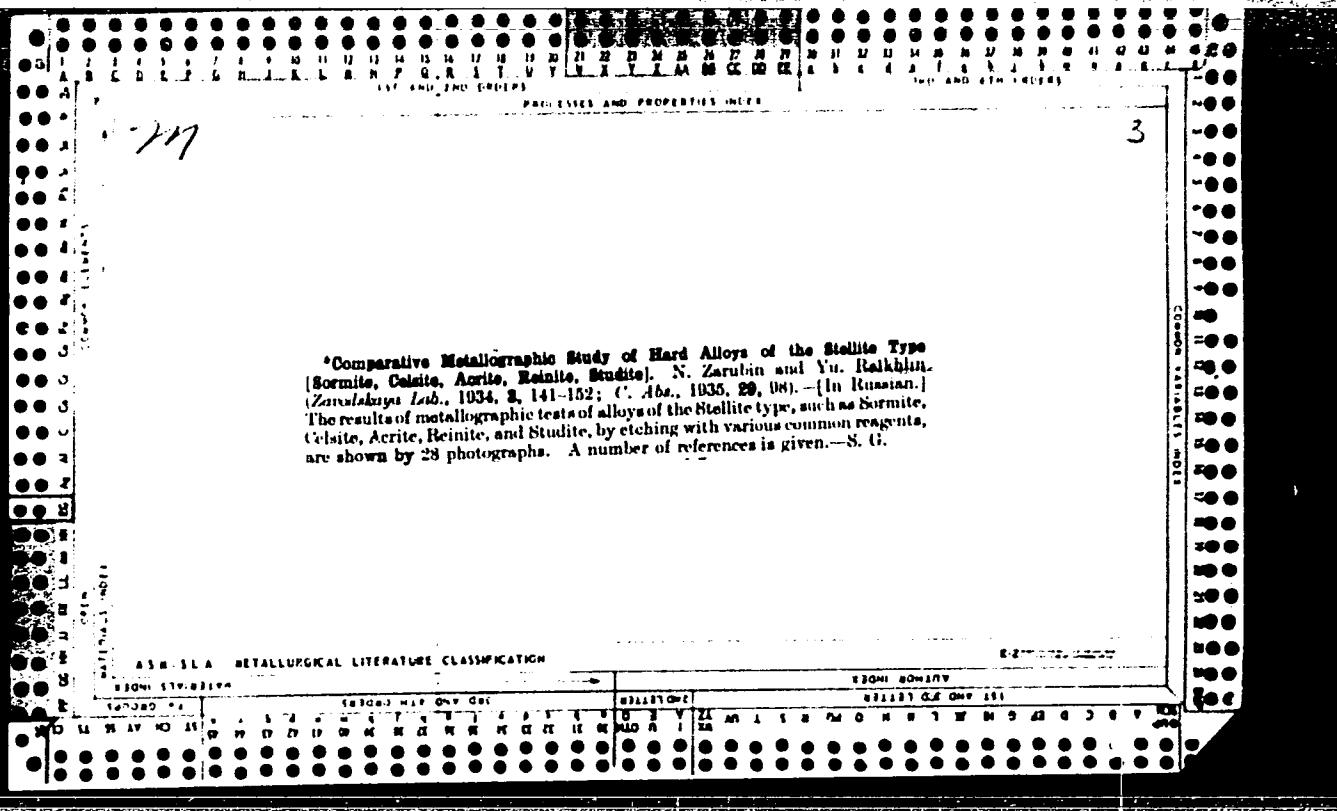
... 1953. An introduction on the toxic effects of rare

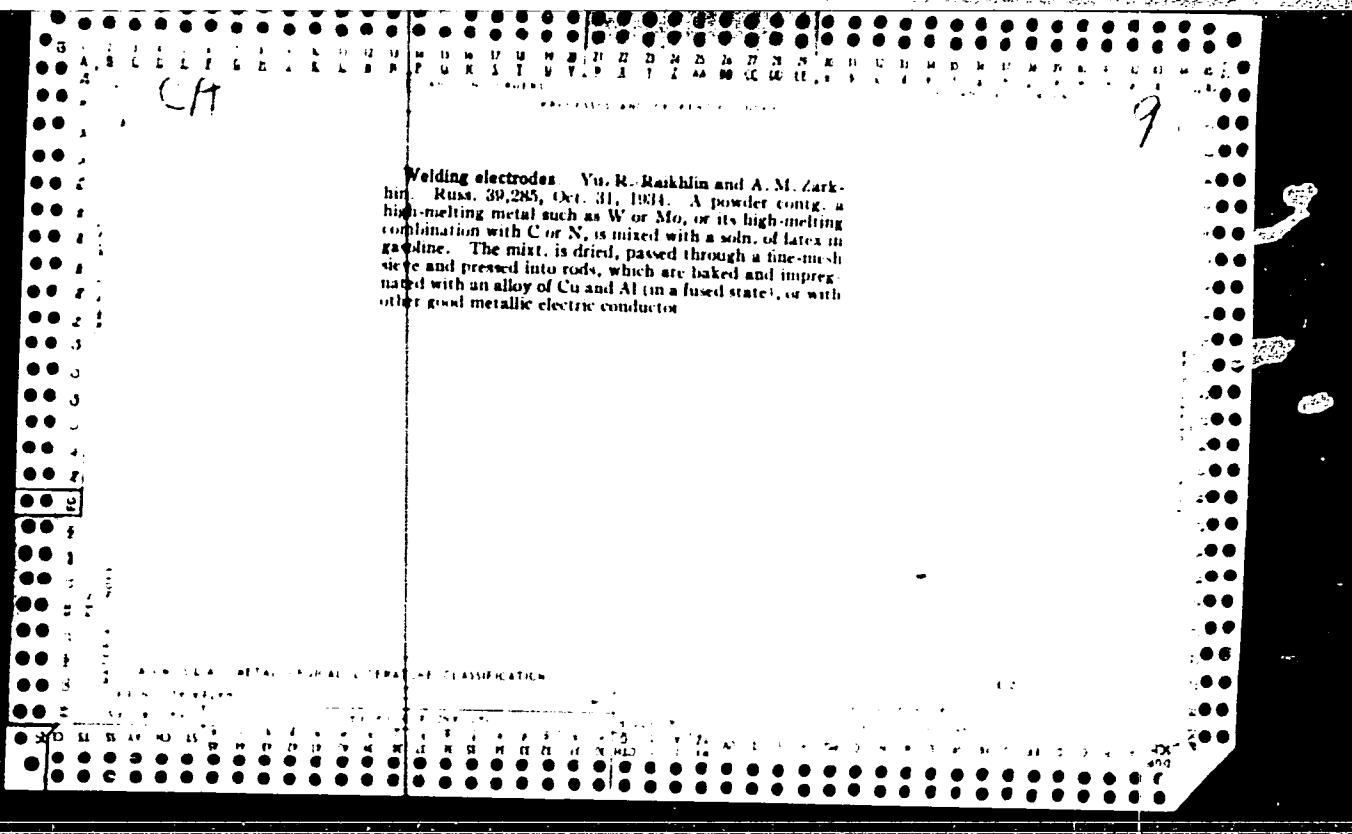
metals, absorption and analytical application of rare  
metals in industry, the clinical picture and  
therapy of rare-metal poisoning, is also given. There are 37  
appendices.

... The Experimental Studies of ... Effect on an Organism of  
Rare, Dispersed, and Other Metals Used in Industry ...  
Their Compounds.

10. Nickel. Yu. Yu. Mogilevskaya	151
11. Cobalt. Z. S. Kaplun (Second ed)	164
12. Uranium compounds. G. I. Rumyantsev	176
13. Zinc and Zinc oxide. I. Ya. Mogilevskaya	187
14. Rare earths. O. Ya. Mogilevskaya and N. I. Raikhlin	195

Comparative metallographic study of hard alloys of the Stellite type. N. Zarubin and Yu. Ralkhin. Zavodskaya Lab. 8, 141-63.—The results of metallographic tests of alloys of the Stellite type, such as Sormite, Celcite, Acrite, Reinitite, Percite, Studite, by etching with various common reagents are shown by 24 photographs. Over 24 references.





CA

9

**Welding electrodes.** Yu. R. Ranklin and A. M. Zarkhin. Russ. 38,715, Sept. 30, 1934. Electrodes are prep'd. from powdered hard metals, such as W, coated (electrolytically or by substitution) with a metal layer that is a good conductor, such as Cu ppt'd. from a soln. of its salts

ADDITIONAL DETAILS AND LITERATURE CLASSIFICATION

*4/16/01* *Class*

Method for making metallceramic products. Yu. N.  
RALKIN AND A. M. ZARKHIN. Russ. 56,975. April 30,  
1910. 406. 17. Such products are made from powdered  
metals to which may be added diamond dust. The molded  
powder is saturated with molten copper under pressure.  
M. Ho

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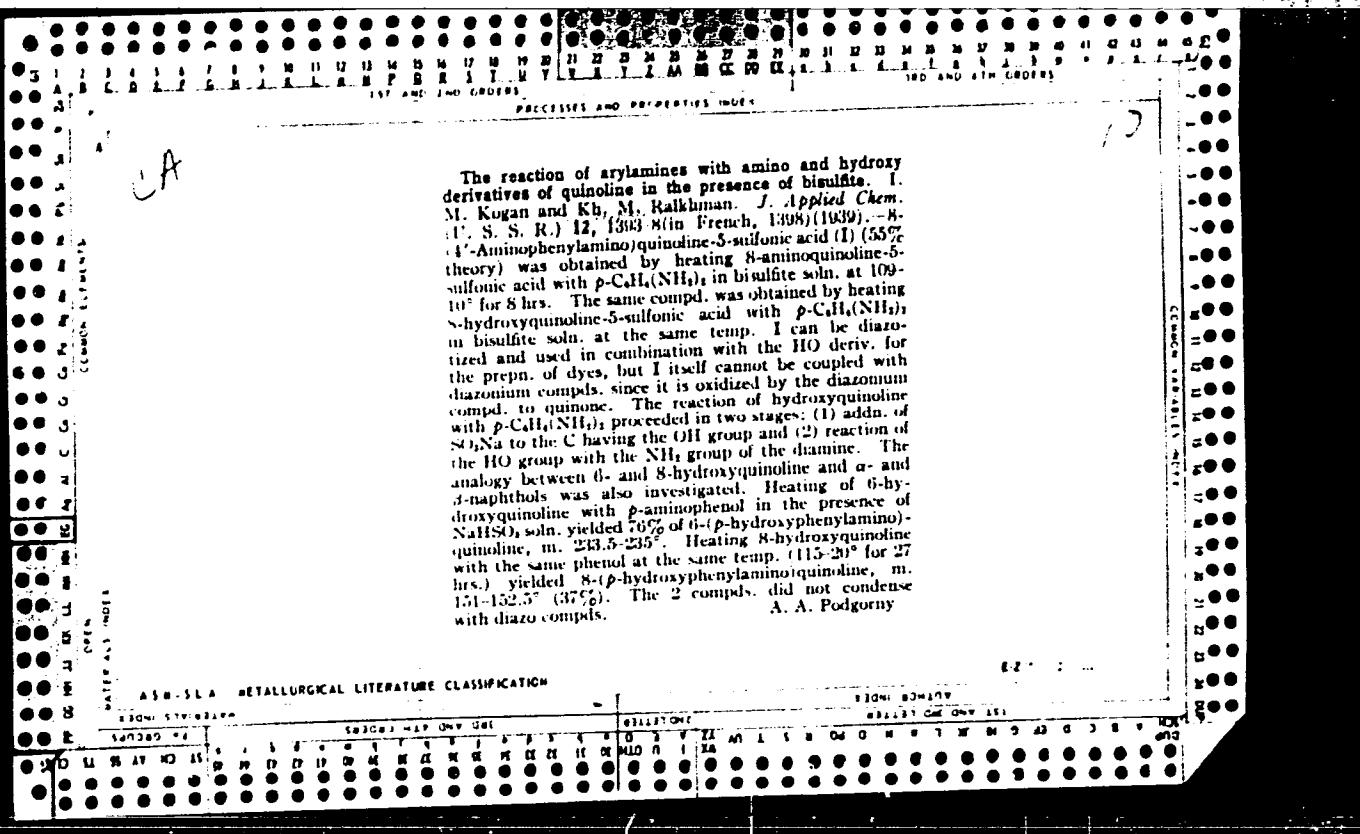
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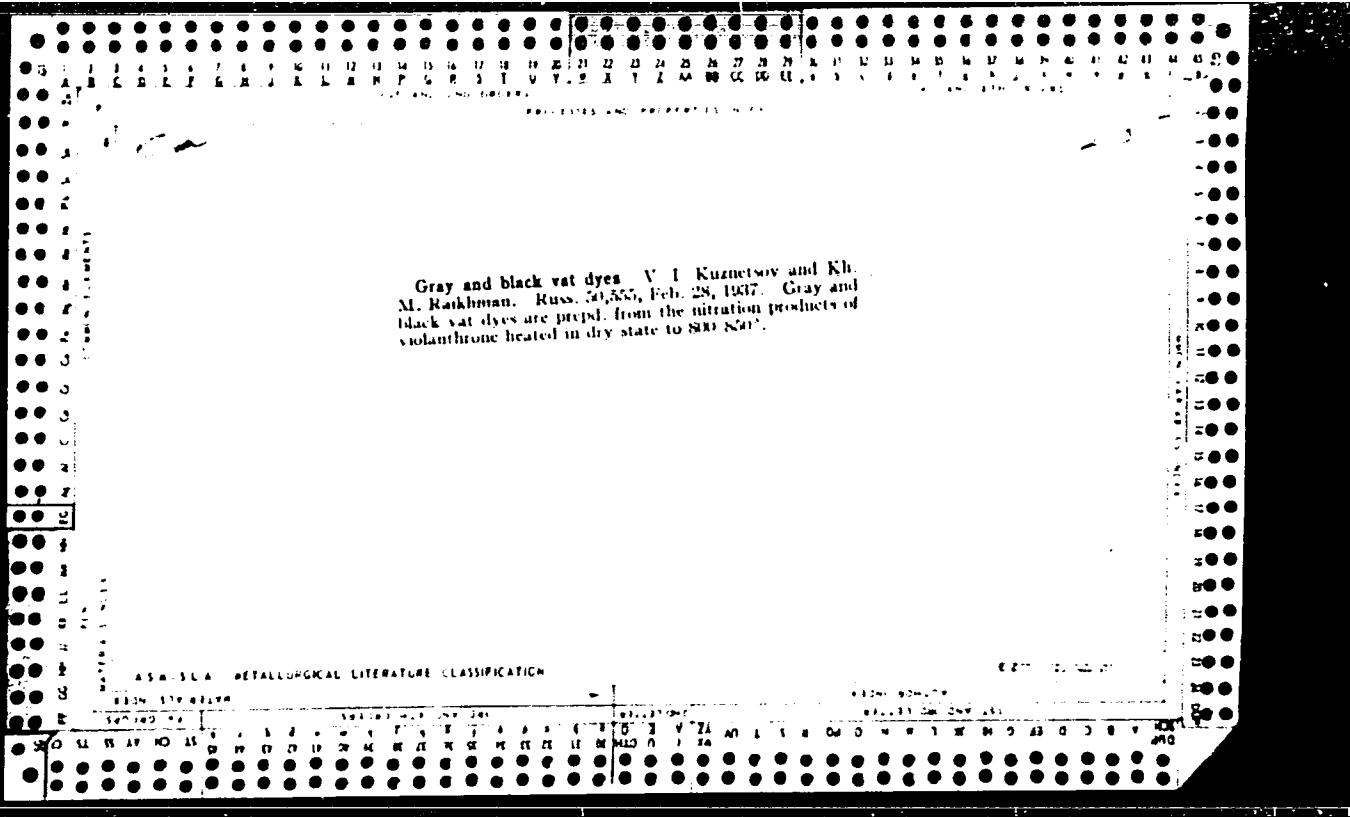
RAIKHMAN, A.Z., inzh.

Controlling steam turbine disks. Energetik 5 no.9:17-19 S '57.  
(MIRA 10:10)  
(Steam turbines)

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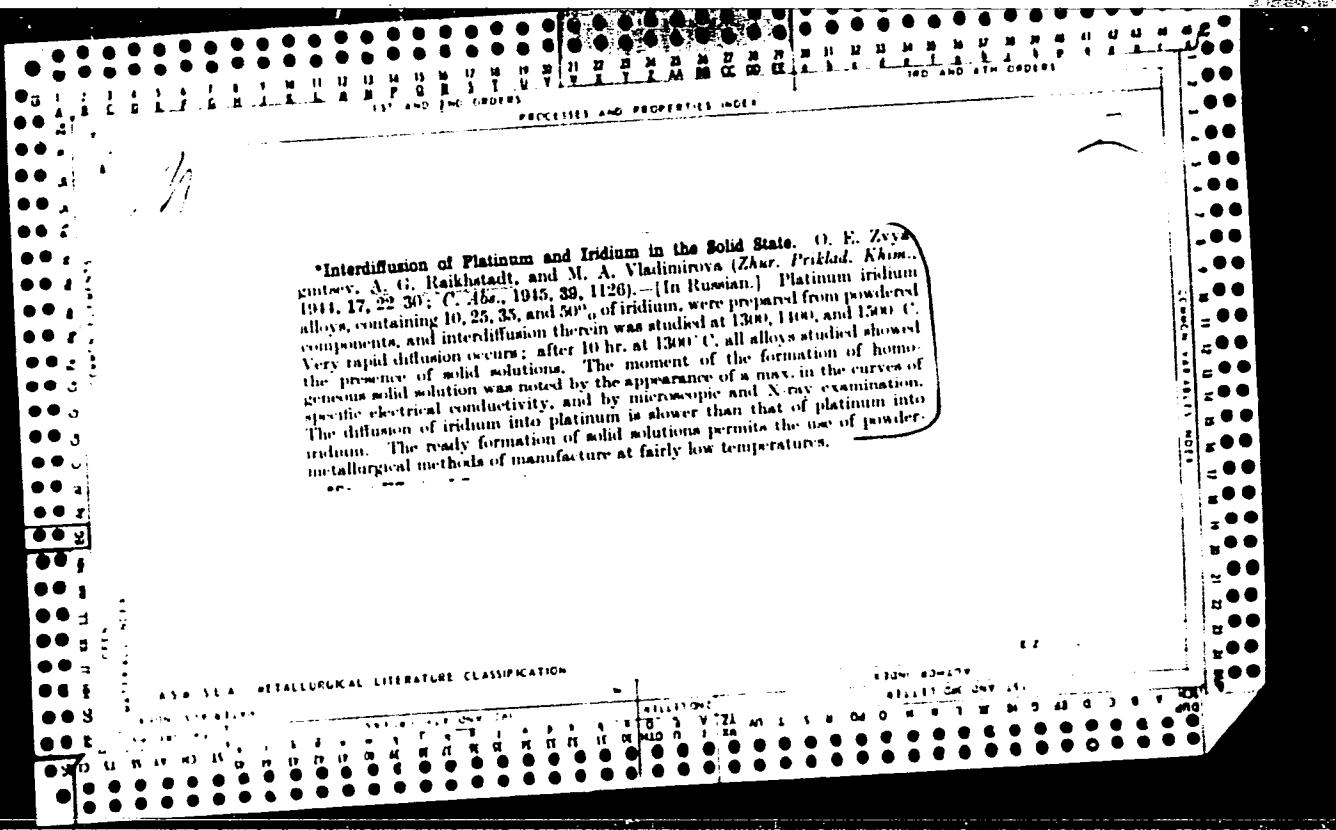
CIA-RDP86-00513R001344020008-9"

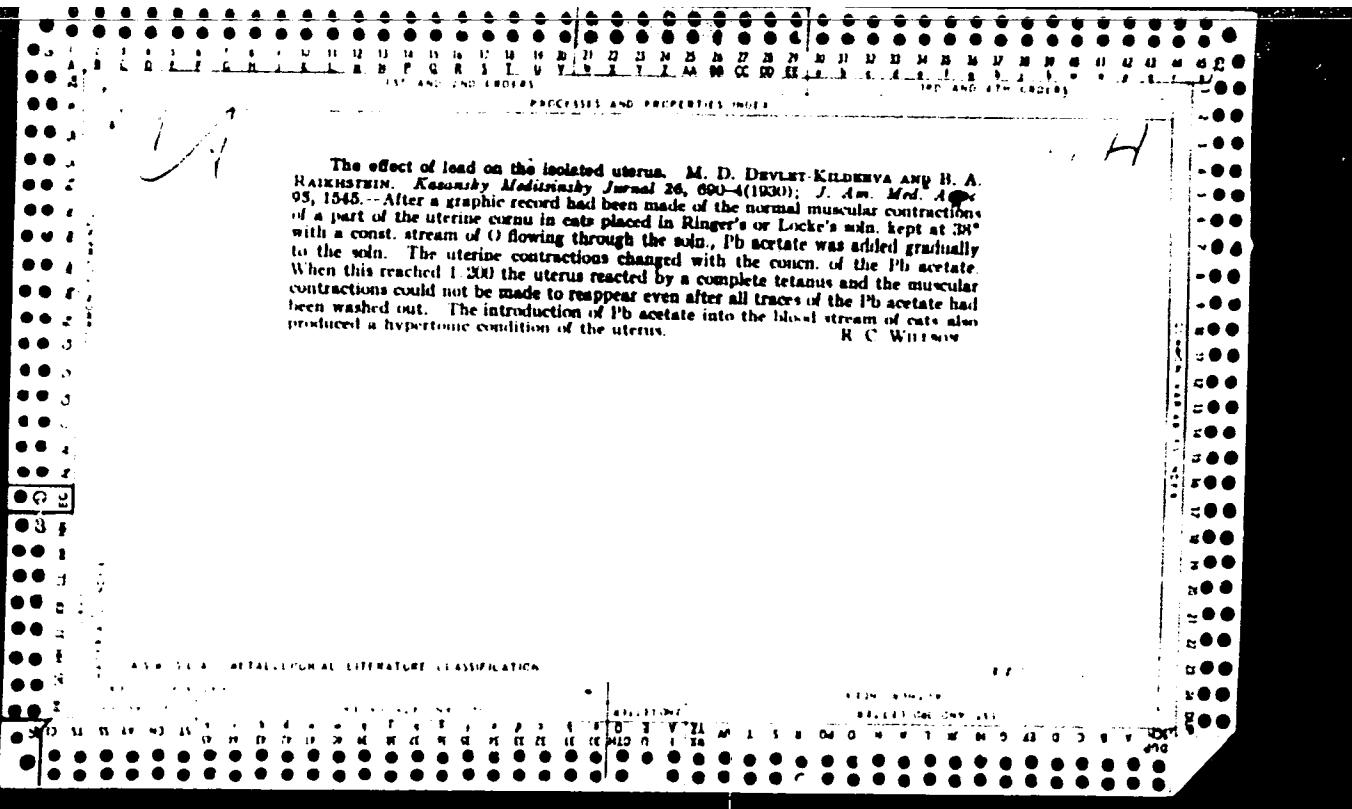
RAIKHRUD, A. Ya.

Quantitative regularities in the regulation of synthesis in the  
virus cell system. Vop. virus. 8 no.3:325-329 My-Je'63.

(MIRA 16:10)

1. Institut virusologii imeni D.I.Ivanovskogo AMN SSSR, Moskva.  
(VIRUS RESEARCH) CELL METABOLISM)





RECORDED, S. I.

Pa-2T49

USSR/Physical Chemistry - Apparatus Mar 1947  
Elasticity Measurements

"An Apparatus for Determination of Dissociation Pressure," S I Raikhshtein, and I A Kazarnovskiy,  
4 pp

"Zhurn Fiz Khim" Vol XXI, No 3

Diagrams and operating data of subject equipment  
for elasticity measurements in dissociation of  
hard substances

2T49

**inorganic peroxides.** XI. Higher oxides of potassium  
I. A. Kazarinovskii and S. I. Radikhishvili. Karpoval'ka  
Phys. Chem., Moscow). *J. Phys. Chem. U.S.S.R.* 24,  
245-255 (1971) (in Russian); cf. *C.A.* 73, 3743c. The  
pressures above  $K_2O_2$  (prepn., described) are 0.35, 1.15  
and  $1.65 \pm 0.1$  mm. Hg at 300°, 300°, and 370°, resp.  
The pressure remains const. on removing O until the  
comptg.  $K_2b_3$  is reached, when it drops to about 0.05 mm.  
Thus,  $K_3O$  does not exist. The dissociation is reversible,  
and  $K_2O_2$  can be obtained by heating  $K_2O_3$  (prepn., described)  
in O.  $K_2O_3$  and  $K_2b_3$  have  $d_{15}^{25}$  2.158  $\pm$  0.001 and  
 $2.190 \pm 0.001$ . From the mol. refraction of  $K_2O_3$ , the  
fraction of the univ. dent. anion  $O_2^-$  is calculated to be 0.0. The  
heat of formation of  $K_2O_3$  is calcd. to be 117,000 cal/l. by  
using thermochem. data of Forer and the above pres-  
sure data. J. J. Bikerman

J. J. Bierman

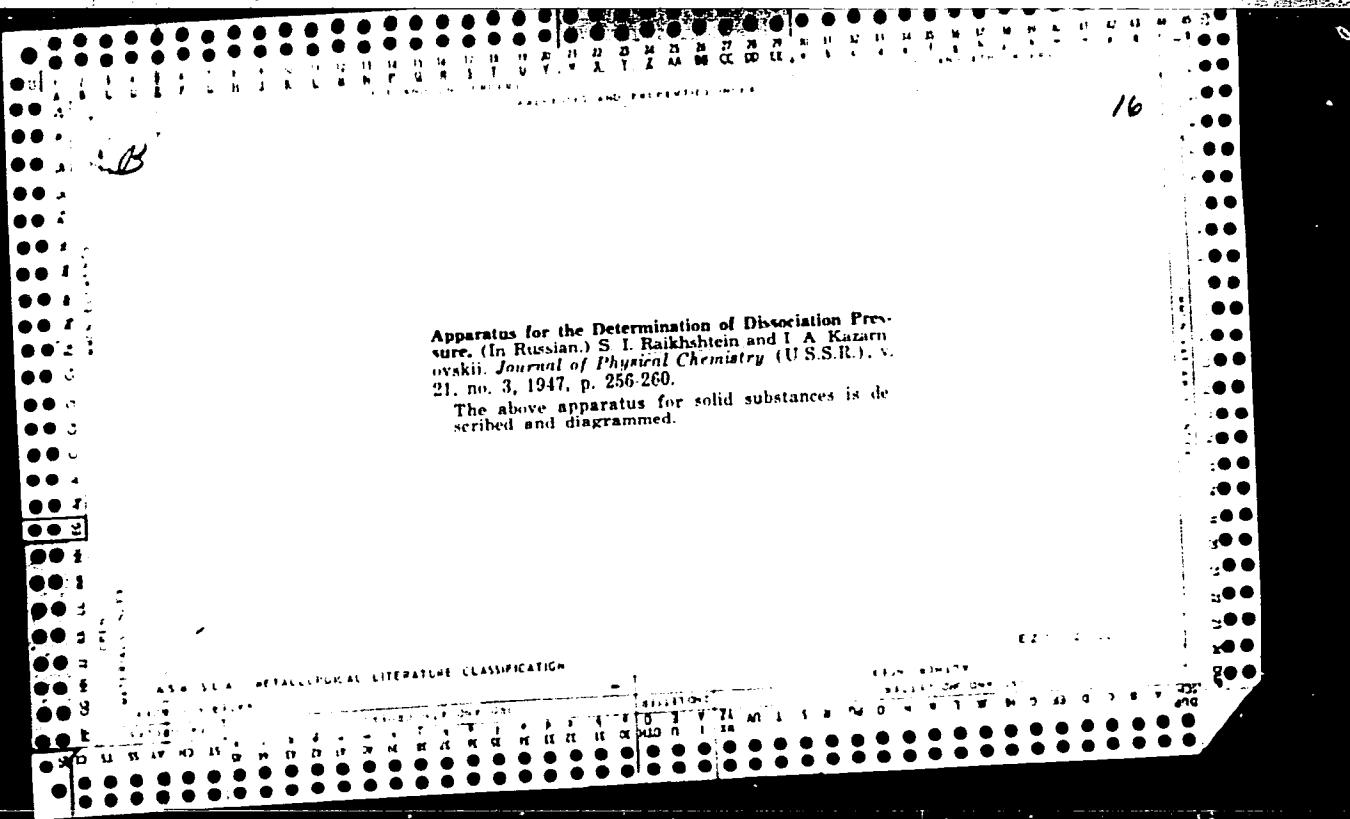
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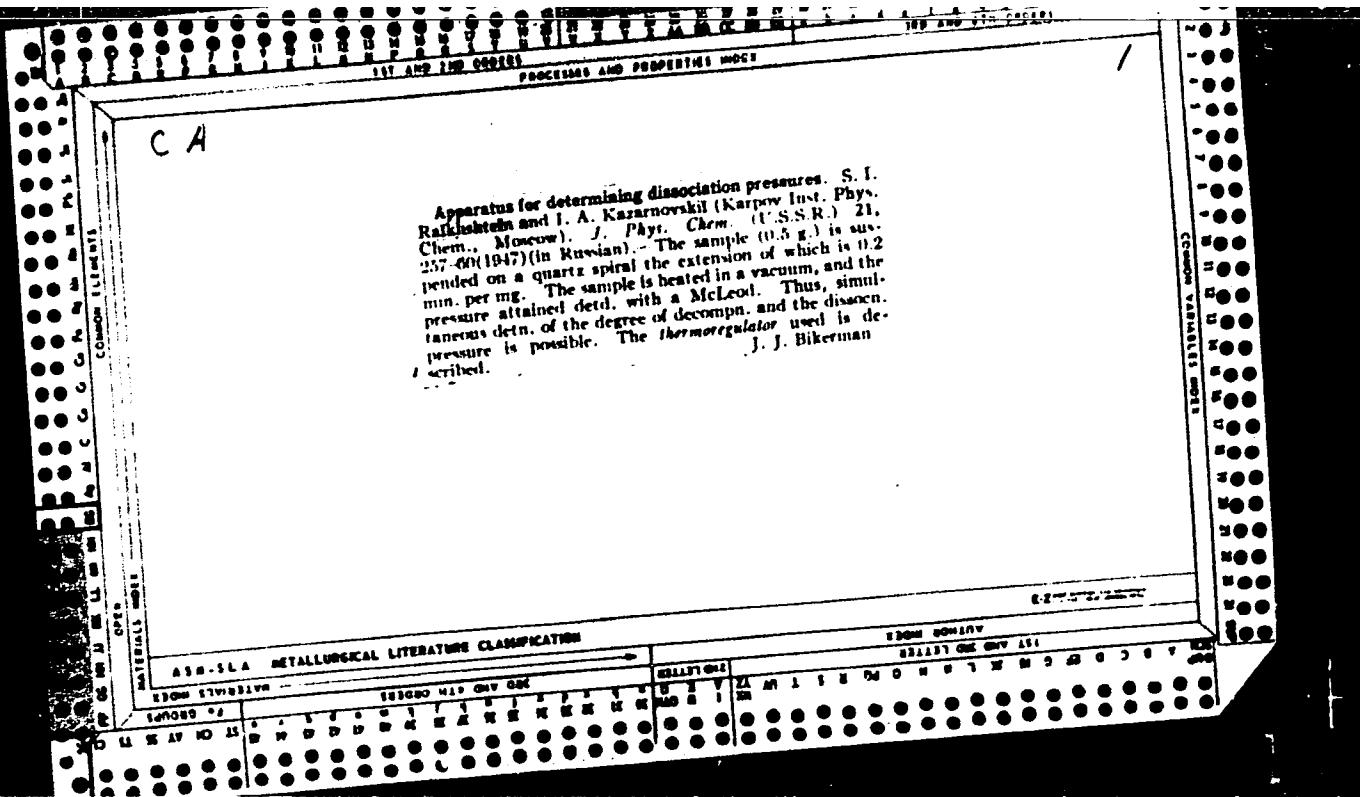
Preliminary communications and discussion. Is there a trioxide of potassium? S. I. Barkhatkina and I. A. Kazarnovskii. *J. Phys. Chem.* U.S.S.R. 11, 133 (1938).—The existence of  $K_3O_3$  was investigated by means of its dissociation point and that of its products of thermal disintegration up to  $K_2O_2$ . It was found that even the first measurements of the thermal disintegration kinetics at 360° and 0.1 mm. pressure pointed to the fact that  $K_3O_3$  is not an individual compnd but is a mixt. of  $K_2O_2$  with  $K_2O$ . The curve is smooth and no bend is found at the point corresponding to  $K_2O_2$ .

W. R. Bratt

## AS-65A - METALLURICAL LITERATURE CLASSIFICATION



16



M. IKHTSABE, A.G.

Khimicheskie laboratorii po  
issledovaniyu uglei (Chemical laboratories for coal  
research). Moscow, Gletezhizdat, 1952. 158 p.

See: Monthly List of Russian Accessions, Vol. 6, No. 1, April 1953

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9

MAKHOV, N. I.

MAKHOV, N. I. "The Janzeyursk Seismic Expedition." Vestnik Akad. Nauk S.S.R., Leninigrad,  
Extra Number, 1932, pp. 177-184.

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020008-9"

RATIN, N. V.

Ratin, N. V. "On the Possibility of Observing Mohorovicic Phase During the Caucasian Earthquakes." Trudy Seismologicheskogo Instituta, No. 14, 1930, p. 1-10.

IVANOV, P., inzh.; KERVANBASHILY, St., inzh.; ARSOV, IA., inzh.; RAIKOV,  
K., inzh.

A new foundry binder based on bitumen. *Neftinostroene 13 no.4:*  
23-27 Ap '64.

RAIKOV, Raiko, inzh.

Distribution of power reserves in various power systems in  
Bulgaria. Elektroenergiia 14 no.1:3-5 Ja '63.

RAIKOV, F.

"Let us mechanize the fight against the grape Paraconsporalec", p 131  
(ROZTRATYNA ZEME FILIP. Vol 6 #4, Apr. 1951, Bulgaria)

SO: Monthly List of RUSSIAN Accessions, /Library of Congress, August 1953, Uncl.  
East European Vol. #2

2.1 A 11

Кирилов (А.) & Райков (Б. В.). Действие на праховидните фунгициди върху  
клатиността на зеленчукоците семена при максимално напрашване.  
[The effect of fungicidal dusts on the germination of vegetable seed when  
maximum dusting is employed.] -Reprinted from Семепроизводство [Seed  
Production], iv, 1-2, 6 pp., 1945.

In comparative tests conducted over a period of ten years in Bulgaria, the highest  
germination rates of vegetable seeds treated with fungicidal dusts were as follows:  
red cabbage with porzol [R.I.M., xvii, p. 460] 87.75 per cent., tillantin R. [ibid.,  
xxvi, p. 187] 85.75, control 79.5; white cabbage with copper carbonate [ibid.,  
xxvi, p. 324] 82.5, tillantin 78.75, control 60.5; dill [*Puucedanum graveolens*] with  
copper carbonate 76, ceresan 71.25, control 67.5; pepper with tillantin 98.25, con-  
trol 95.75; radish with copper carbonate 84.25, tillantin 82.25, control 44.25;  
lettuce with porzol 88, copper carbonate 81.75, control 81.5; eggplant with tillantin  
73.25, porzol 69.75, control 45. In one test granosan and ceresan completely con-  
trolled *Alternaria radicina* [ibid., xxv, p. 378] on heavily infested carrot seed.

RAIKOV, G., inzh.; ZLATEV, Zl.

Installation of cables at the mine hoisting machine with friction plates. Min delo 16 no.11:24-27 '61.

1. "Mashiproekt"(for Raikov) 2. Gl. mekhanik na Durzhavno minno predpriatie "Burgazski medni mini"(for Zlatev)

(Mining machinery) (Hoisting machinery)

RAIKOV, G.  
TECHNOLOGY

Novelties on the prespinning machines for worsted spinning. p. 19.

LEKA PROMISLJENOST. TEKSTIL. (Ministerstvo na lekata promishlenost) Sofiia.

Vol. ?, no. 6, 1958.

SO: Monthly List of East European Accessions (EEAI) LC

Vol. 8, No. 3  
Uncl. March 1959

RAIKOV, G.

Regulating the mechanisms of the flat carding machine.

P. 21, (Lika Promishlenost) Vol. 6, no. 2, 1957, Sofia, Bulgaria

SO: Monthly Index of East European Acessions (EEAI) Vol. 6, No. 11 November 1957

RADEV, R., inzh.; BALEV, V., inzh.; RAJKOV, Il., inzh.

Determining the depth of dressing of the Chukurovo coals.  
Min delo 17 no.11:12-14 '62.

1. Bulgarska akademia na naukite.

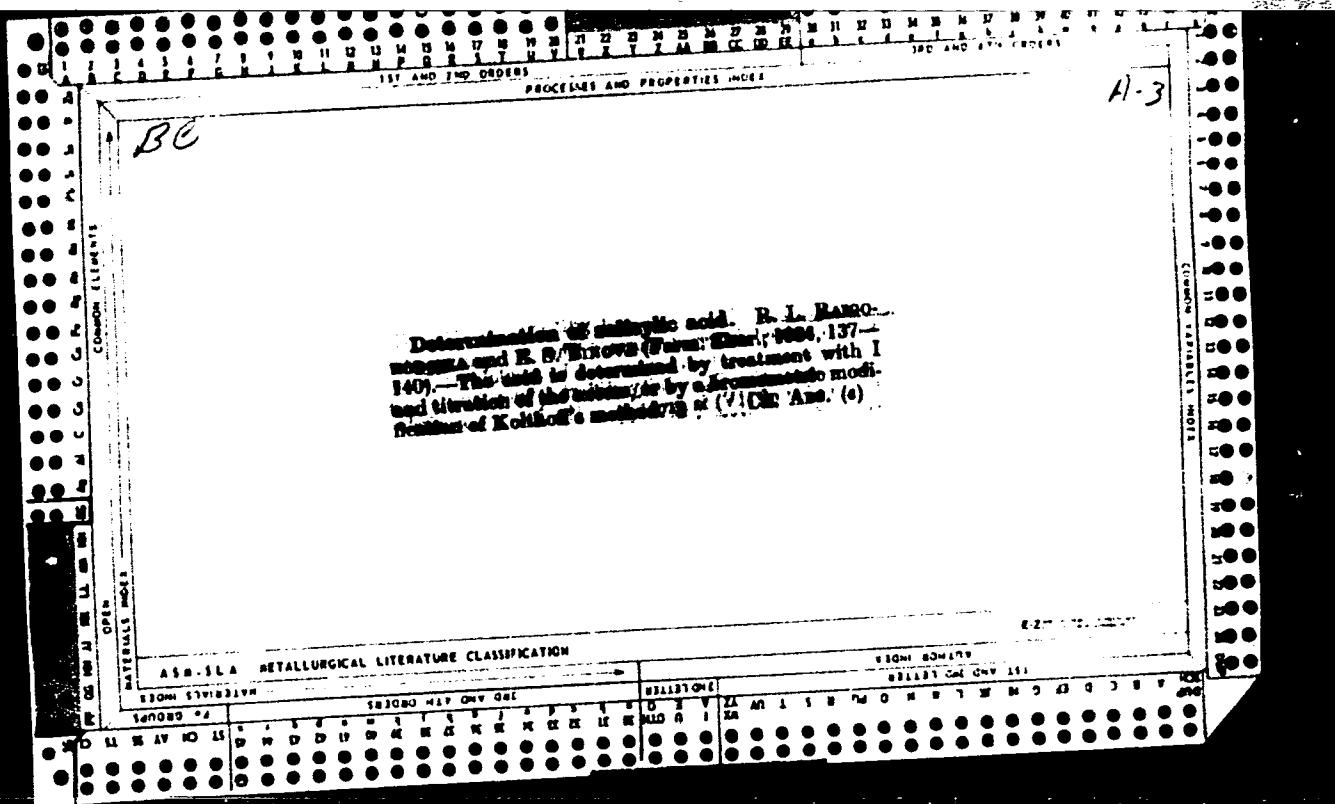
RAIKOV, IVAN

Ot Buzludzha do vrukh Botev. Sofiia, Profizdat, 1957. 7sp. (From Buzludzha  
Mountain to Botev Peak. illus., fold map, bibl.)

SO: Monthly Index of East European Accessions (EEAI) Vol. 6, No.11 November 1957

19  
Bond clays of northern White Russia and their industrial  
utilization. I. P. Kaligurovetskii and L. I. Kalika.  
*Struktol Material'* 1035, No. 8, 42-4. E.E.S.

AS4-114 METALLURGICAL LITERATURE CLASSIFICATION



Determination of iodine in iodides. R. L. RAOGORODSKA and E. S. BINOVA (Farm. Shur., 1935, No. 1, 23-25).—Free I is titrated with  $\text{Na}_2\text{S}_2\text{O}_3$  and total I' with  $\text{AgNO}_3$ .  
Ca. Ann. (e)

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## ASS-11A METALLURGICAL LITERATURE CLASSIFICATION

3306 82010.8

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R001344020008-9"

Determination of salicylic acid. R. I. Raigorskaya and E. S. Vinova. *Farm. Zhur.* 1934, 137-40. Salicylic acid can be detd. by treatment with  $I_2$  and titration of the excess, or by a bromometric modification of Koltzoff's method, which is very accurate. L. Nasarevich

APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R001344020008-9"

CS

7

Detection and determination of salicylic and benzoic acids and their salts. V. S. Il'ina and R. I. Ruzgorskaya. Izv. Akad. Nauk. 1934, no. 4. Detection is effected by sublimation in a Hg sublimate. Benzoic acid volatilizes at 250-260° and the salicylic acid at 70-85°, to form their characteristic crystal sublimes. Salts must be converted into the acids. For quantitative work the methods of Messinger (J. Prakt. Chem., 61, 20, 1890) or Baumgärtel (ibid., 2, 250) and of Kesthelyi (ibid., 15, 280) are recommended.

I. N. G.

**Microchemical determination of alkaloids of ipecacuanha.** E. S. Binova and R. L. Raigorsk'a, *Trans. U.S.S.R. Inst. Exptl. Pharm.*, 1, 151 (1938). The qual. reagent used was a mixt. of  $\text{NH}_4$  molybdate and  $\text{H}_2\text{SO}_4$ . This gives a green color with the alkaloid. For the detn.,  $\text{MgO}$ , tragacanth and chloroform were used. One hundred mg. of powd. root can be analyzed by this method. R. Levine

R. Levine

**APPROVED FOR RELEASE: 03/20/2001**

CIA-RDP86-00513R001344020008-9"

CA

Determination of Iodine with Iodides R. E. Babu  
govinda and K. S. Bhava *Zeta Zeta* 1915, No. 1,  
20 A. Free I is titrated with Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, the iodides are  
then titrated with AgNO<sub>3</sub> and the free I equiv. is sub-  
tracted from total iodides. 1 Na atom each

7

CA  
1st AND 2nd Cols (1)  
PROCESSES AND PROPERTIES INDEX  
3rd AND 4th Cols (2)

CONCEN. ELEMENTS	ACETYLENE DERIVATIVES																																																							
	LXXXVI. Heterocyclic compounds. 7. Synthesis of 4-vinylenyl-4-hydroxypiperidines by condensation of vinylacetyles with 4-piperidones. I. N. Nazarov, V. Ya. Ralgorodskaya, and V. A. Rudenko. <i>Izvest. Akad. Nauk S.S.R., Otdel. Khim. Nauk</i> 1949, No. 1, 68-75; cf. <i>C.A.</i> 43, 2094. - CH <sub>3</sub> :CHC:CH (7 g.) and 14.1 g. 2,5-dimethyl-4-piperidone in 90 ml. Et <sub>2</sub> O added to 10 g. powd. KOH in 30 ml. Et <sub>2</sub> O with stirring at -5°, stirred 8 hrs. in the cold, and let stand overnight, gave 12.3 g. 2,5-dimethyl-4-vinylenyl-4-hydroxypiperidine (I), b.p. 135-6°. 1 hydrogenated with Pt catalyst in EtOH gave 2,5-dimethyl-4-butyl-4-hydroxypiperidine, needles, m.p. 106-7° (sublimed for purification). The latter, b.p. 85-6° (liquid form), was obtained also in 7.3 g. yield from Bu <sub>2</sub> MgCl (30 g. BuCl) and 20 g. 2,5-dimethyl-4-piperidone, after heating in Et <sub>2</sub> O, then in C <sub>6</sub> H <sub>6</sub> to 70°, prior to decompr. with 15% HCl and ice. CH <sub>3</sub> :CHC:CH (6 g.) and 12.6 g. 2,5,6-trimethyl-4-piperidone, treated as above, gave 13.2 g. 2,5,6-trimethyl-4-vinylenyl-4-hydroxypiperidine, b.p. 140-3°, m.p. 123.5-4.5° (from C <sub>6</sub> H <sub>6</sub> ); 4-Bu analog, m.p. 135-6° (from C <sub>6</sub> H <sub>6</sub> ). 2,5-Dimethyl-6-ethyl-4-piperidone (11.3 g.) gave 10 g. 2,5-dimethyl-6-ethyl-4-vinylenyl-4-hydroxypiperidine, b.p. 143-5°; 4-Bu analog, b.p. 142-4°, n <sub>D</sub> <sup>20</sup> 1.4825, d <sub>40</sub> <sup>20</sup> 0.9301. Similarly, 11.3 g. 2-methyl-4-ketodecahydroquinoline gave 9.8 g. 2-methyl-4-hydroxy-4-vinylenyldecahydroquinoline, b.p. 140-1°; 4-Bu analog, m.p. 136.8° (from C <sub>6</sub> H <sub>6</sub> ). Analogously, 23 g. 1,2,5,6-tetramethyl-4-piperidone gave 14 g. 1,2,5,6- <p style="text-align: right;">10</p> <p style="text-align: right;">tetramethyl-4-vinylenyl-4-hydroxypiperidine, b.p. 122-2.5°, n<sub>D</sub><sup>20</sup> 1.8138, d<sub>40</sub><sup>20</sup> 0.9658; 4-Bu analog, b.p. 104-1°, n<sub>D</sub><sup>20</sup> 1.4894, d<sub>40</sub><sup>20</sup> 0.9418 [also prep'd. in 25% yield from Bu<sub>2</sub>MgCl and the corresponding piperidone; in this case, the product b.p. 113-15°, crystall., and m.p. 58-9° (from petr. ether); apparently the Grignard syntheses yield isomers of the compds. obtained by the CH<sub>3</sub>:CHC:CH condensation]. LXXXVII. Mechanism of dehydration and cyclization of diynes. 17. Hydration and cyclization of 5-propyl-1,5-octadien-3-yne. I. N. Nazarov and I. I. Zuretskaya. <i>Ibid.</i> 178-83; cf. <i>C.A.</i> 43, 115.—CH<sub>3</sub>:CHC:CH (159 g.) in 200 ml. dry Et<sub>2</sub>O and 230 g. Pr<sub>2</sub>CO added to 155 g. powd. KOH in 400 ml. Et<sub>2</sub>O in the cold over 4 hrs., and stirred 8 hrs. in the cold and 4 hrs. at room temp. gave, after the usual aq. treatment, 251 g. (77%) dipropyl(vinylethynyl)carbinol, b.p. 73-6°, n<sub>D</sub><sup>20</sup> 1.4780. This (103 g.) and 100 g. 50% H<sub>2</sub>SO<sub>4</sub>, stirred 5 hrs. at 60-5° gave 81% 5-propyl-1,5-octadien-3-yne (I), b.p. 65-7°, n<sub>D</sub><sup>20</sup> 1.4900. I (6 g.) hydrogenated in AcOH with Pt oxide gave 5-propylactane, b.p. 174-5°, b.p. 63-5°, n<sub>D</sub><sup>20</sup> 1.4190, d<sub>40</sub><sup>20</sup> 0.7447. I (77 g.), 300 g. 90% MeOH, 1.5 ml. H<sub>2</sub>SO<sub>4</sub>, and 4 g. Hg sulfate stirred 1 hr. at 65°, then treated with a total of 10 g. Hg sulfate in 2 portions spaced 2.5 hrs. (total time 6 hrs.) gave, after concn. and neutralization, 87 g. 5-propyl-1,5-octadien-4-one (contaminated with some 5-propyl-2-methoxy-5-octen-4-one), b.p. 90-8°; the impurity was removed by 1 hr. at 60-100° and 75 mm. with 0.5 g. <i>p</i>-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H (3 (faux))</p> <p style="text-align: center;">ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION</p> <table border="1"> <tr> <td rowspan="2">MATERIALS INDEX</td> <td colspan="8">1930-53 MIP ONE DAY</td> <td colspan="8">1931-34 MIP ONE DAY</td> </tr> <tr> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td> </tr> <tr> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td> </tr> </table>								MATERIALS INDEX	1930-53 MIP ONE DAY								1931-34 MIP ONE DAY								1	2	3	4	5	6	7	8	1	2	3	4	5	6	7	8	1	2	3	4	5	6	7	8	1	2	3	4	5	6	7
MATERIALS INDEX	1930-53 MIP ONE DAY									1931-34 MIP ONE DAY																																														
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*LXXXVII.*  $b_2$  103-0°,  $n_D^20$  1.4770,  $d_4^{\circ}$  0.8900, which does not give a semicarbazone or dinitrophenylhydrazone, nor undergo hydration under the above conditions. Hydrogenation over Pt oxide in EtOH gave 5-propyl-4-octanone,  $b_2$  89-92°,  $b_2$  201-4°,  $n_D^20$  1.4300,  $d_4^{\circ}$  0.8300, which yields a 2,4-dinitrophenylhydrazone, m. 61° (from EtOH), but not an oxime or semicarbazone. The structure of the dienone was shown by ozonization, which gave  $\text{HCO}_2\text{H}$ ,  $\text{AcO}_2\text{H}$ ,  $\text{EtCO}_2\text{H}$ , and  $\text{PrCO}_2\text{H}$ . The dienone (25 g.) and 45 g.  $\text{H}_3\text{PO}_4$  (d. 1.77) after 2.5 hrs. at 80-5° gave 20 g. 1-propyl-2-ethyl-3-methyl-1-cyclopenten-5-one,  $b_2$  103-5°,  $n_D^20$  1.4770,  $d_4^{\circ}$  0.9116; semicarbazone, m. 185-8° (from EtOH); 2,4-dinitrophenylhydrazone, m. 137-8°. The same product is obtained with undiminished yield on treatment with  $\text{H}_3\text{PO}_4$  11 hrs. at 10-14°. Hydrogenation over Pt oxide gave in poor yield the corresponding cyclopentanone deriv.; semicarbazone, m. 173-4°. Treatment of (28 g.) 5-propyl-1,5-octadien-3-yne with 50 g.  $\text{H}_3\text{PO}_4$  at 80-5° for 8 hrs. gave the above cyclopentenone directly but in very poor yield (1 g.), the rest of the dienone being unchanged. The structure of the cyclopentenone is shown by ozonization, which yields  $\text{PrCO}_2\text{H}$  and  $\text{EtCO}_2\text{H}$ ;  $\text{CH}_3\text{MeCH}_2\text{CO}_2\text{H}$ ,  $b_2$  130-9°,  $n_D^20$  1.4680, and by  $\text{KMnO}_4$  oxidation, which yields the same products; the above keto acid yields a 2,4-dinitrophenylhydrazone, m. 115-18°. *LXXXVIII.* The mechanism of hydration and cyclization of diynes. 18. Hydration and cyclization of 5-methyl-1,5-tetradecadien-3-yne. *Ibid.* 184-0. Adm. of 344 g.  $\text{C}_5\text{H}_11\text{COMe}$ , 150 ml.  $\text{CH}_3\text{CHC}_2\text{CH}_3$ , and 200 ml. dry Et<sub>2</sub>O to 130 g. powd. KOH in 500 ml. Et<sub>2</sub>O with stirring and cooling over 4 hrs., followed by 3 hrs. stirring in the cold and 4 hrs. at room temp. gave, after the usual aq. treatment and neutralization, 322 g. mixed methylnonyl- (vinylethynyl)carbinol and 5-methyl-1,5-tetradecadien-3-yne (I),  $b_2$  103-30°; repeated fractionation gave 160 g. of the former,  $b_2$  120-2°,  $b_2$  131-3°,  $n_D^20$  1.4725,  $d_4^{\circ}$  0.8580, which on hydrogenation over Pt oxide in EtOH gave 5-methyl-5-tetradecanol,  $b_2$  135-8°,  $n_D^20$  1.4440,  $d_4^{\circ}$  0.8303, while heating the carbisol (80 g.) with 80 g. 65%  $\text{H}_2\text{SO}_4$  4 hrs. to 65° gave 79% pure I,  $b_2$  100-2°,  $n_D^20$  1.4850,  $d_4^{\circ}$  0.8183. Hydrogenation of I over Pt oxide in AcOH gave 5-methyldecahexene,  $b_2$  125-7°,  $n_D^20$  1.4320,  $d_4^{\circ}$  0.7808. Stirring 58 g. I, 201 g. 90% MeOH, 1 ml.  $\text{H}_2\text{SO}_4$ , and 4 g. Hg sulfate 2 hrs. at 70°, followed by addn. of a total of 12 g. Hg sulfate in 2-g. portions at 2-hr. intervals, gave 43 g. 5-methyl-1,5-tetradecen-4-one,  $b_2$  120-33°,  $n_D^20$  1.4781,  $d_4^{\circ}$  0.8730; shorter reaction results in partial reaction; no MeO derivs. were detected. The dienone on hydrogenation over Pt oxide in EtOH gave 5-methyl-4-tetradecanone,  $b_2$  140-1°,  $n_D^20$  1.4400,  $d_4^{\circ}$  0.8278, and a small amt. of the cyclic deriv. described below. The dienone (32 g.) and 50 g.  $\text{H}_3\text{PO}_4$  (d. 1.77) after 3 hrs. at 80-8° gave 30 g. 1,3-dimethyl-2-acyl-1-cyclopenten-3-one,  $b_2$  134-5°,  $n_D^20$  1.4763,  $d_4^{\circ}$  0.8920; semicarbazone, m. 134-5°; 2,4-dinitrophenylhydrazone, m. 99-100° (from EtOH). This ketone is resistant to hydrogenation over a Pt catalyst. Oxidation by  $\text{KMnO}_4$  gave  $\text{AcO}_2\text{H}$ , methylsuccinic acid, caprylic acid, and  $\text{C}_5\text{H}_11\text{COCH}_2\text{MeCH}_2\text{CO}_2\text{H}$ ,  $b_2$  152-4°,  $n_D^20$  1.4595, which forms a semicarbazone, m. 124-5° (from 50% EtOH). *LXXXIX.* Transformations of 2-butyne-1,4-diol. I. N. Nazarov, L. N. Terekhova, and I. V. Torgov. *Ibid.* 287-92.—MeOH (700 g.) and 40 ml. of a soln. of 15 g.  $\text{H}_2\text{SO}_4$  and 20 g. Hg sulfate in 135 ml.  $\text{H}_2\text{O}$  were treated simultaneously with the rest of the above soln. and 300 g. ( $\text{HOCH}_2\text{C}_2\text{H}_5$ ) in 300 ml. MeOH at 30° over 4 hrs.; then 2 g. Hg sulfate was added and stirred 5 hrs. at 30°; neutralization gave 74% 1-methoxybutan-4-ol-3-one (I),  $b_2$  85°,  $d_4^{\circ}$  1.0930,  $n_D^20$  1.4390; semicarbazone, m. 123-3.5° (from MeOH). To 40 g. MeOH and the catalyst from 2 g.  $\text{HgO}$ , 2 ml. MeOH, and 0.5 ml.  $\text{BF}_3\text{Et}_2\text{O}$  was added 20 g. ( $\text{HOCH}_2\text{C}_2\text{H}_5$ ) in 20 ml. MeOH over 2.5 hrs. at 4°, and the mixt. heated 2 hrs. to 50°, and neutralized with  $\text{NaOAc}$ , giving 18.5 g. I; a poorer yield is obtained with Hg sulfate in dry MeOH. Heating I (110 g.) and 0.9 g.  $\text{p-MeC}_6\text{H}_4\text{SO}_3\text{Na}$  to 85° at 50 mm.

(Cont)

## PROCESSES AND PROPERTIES INDEX

gave 2.5 g. *1-buten-4-ol-3-one*,  $b_{14}$  61-2°,  $n_D^2$  1.4529,  $d_4^{20}$  1.1113; 2,4-dinitrophenylhydrazone, m. 232-4° (from  $\text{C}_6\text{H}_6$ ); the keto alc. polymerizes on standing to a colorless solid; I (90 g.), 0.7 g. pyrogallol, and 0.76 g.  $p$ -Me $\text{C}_6\text{H}_4\text{SO}_3\text{H}$  heated to 50-80° at 60 mm. gave 40 g. distillate, which was immediately hydrogenated over Pt; fractionation of the combined products of several runs gave 0.5 g. *butan-1-ol-2-one*,  $b_{14}$  64-5°,  $n_D^2$  1.4200,  $d_4^{20}$  1.026; osazone, m. 117°. XC. Mechanism of hydration and cyclization of dienynes. 19. Hydration and cyclization of 5,6-diphenyl-1,5-hexadien-3-yne. I. N. Nazarov and I. L. Kotlyarevskii. *Ibid.* 293-8.—To the Grignard reagent of  $\text{CH}_3\text{CH}(\text{C}_6\text{H}_5)\text{CH}_3$  (from 100 g.  $\text{EtBr}$ , 24 g. Mg, and 70 g.  $\text{CH}_3\text{CH}(\text{C}_6\text{H}_5)\text{CH}_3$ ) in 300 ml.  $\text{Et}_2\text{O}$  was added with cooling 130 g.  $\text{PhCOCH}_2\text{Ph}$  in  $\text{Et}_2\text{O}$ ; stirring 2 hrs. at 35°, letting stand overnight, and treating with 10% HCl gave *phenylbenzyl(cinnylethynyl)carbinol*,  $b_{14}$  162°,  $n_D^2$  1.5912,  $d_4^{20}$  1.009, which slowly polymerizes on standing; hydrogenation over Pt gave *phenylbenzylbutylcarbinol*,  $b_{14}$  147-8°,  $n_D^2$  1.5559,  $d_4^{20}$  1.030; dehydration by  $\text{HgSO}_4$  at 130-40° and 10 mm. gave 5,6-diphenyl-1,5-hexadien-3-yne (I),  $b_{14}$  150-60°,  $n_D^2$  1.6757, also obtained by stirring with 60%  $\text{H}_2\text{SO}_4$  at 65-70°. Hydrogenation of the latter gave 5,6-diphenylhexane,  $b_{14}$  154°,  $n_D^2$  1.6370,  $d_4^{20}$  0.956. I (115 g. crude), 400 ml. 90% MeOH, 10 ml.  $\text{H}_2\text{SO}_4$ , and 30 g. Hg sulfate stirred 18 hrs. at 65-8°, cooled, diluted with water, neutralized, and the  $\text{C}_6\text{H}_6$  ext. of the mixt. rapidly distd. in small portions with  $p$ -

$\text{MeC}_6\text{H}_4\text{SO}_3\text{H}$  in *vacuo* gave 87 g. 5,6-diphenyl-1,5-hexadien-4-one (II),  $b_{14}$  170-9.5°,  $n_D^2$  1.6320; semicarbones could not be formed; on long standing the ketone solidifies and m. 70°; hydrogenation over Pt gave 5,6-diphenyl-4-hexanone,  $b_{14}$  117-9°,  $n_D^2$  1.5447,  $d_4^{20}$  1.011; semicarbone, m. 181.5° (from  $\text{EtOH}$ ); ozonization gave  $\text{BaOH}$  and  $\text{HCO}_2\text{H}$ . II (4 g.) and 4 ml.  $\text{H}_3\text{PO}_4$  (d. 1.82) after 15 min. at 70-80° gave 3.8 g. 1,2-diphenyl-3-methyl-1-cyclopenten-5-one,  $b_{14}$  184°, m. 100° [semicarbone, m. 220° (from  $\text{EtOH}$ )], also formed on heating with  $p$ -Me $\text{C}_6\text{H}_4\text{SO}_3\text{H}$ ; hydrogenation over Pt gave 1,2-diphenyl-3-methyl-5-cyclopentanone, m. 97-8°; semicarbone, m. 208-9° (from  $\text{EtOH}$ ). Ozonation of the cyclopentene gave  $\text{BaOH}$  and  $\alpha$ -benzoylethrylic acid,  $b_{14}$  155-62°, m. 60-5° [semicarbone, m. 176-7°]. XC1. Chemistry of divinyl ketones. 16. Addition of hydrogen cyanide to 2,2-dimethyldivinyl ketone. I. N. Nazarov and M. V. Kuvarzina. *Ibid.* 291-304.—To 22 g.  $\text{Me}_2\text{C}(\text{CH}_2)_2\text{CH}_2$ , 20 ml.  $\text{EtOH}$ , and 26 g.  $\text{AcOH}$  was added in 25 min. 28 g. KCN in 90  $\text{H}_2\text{O}$ , the mixt. filtered after 4 hrs. at 35-7°, cooled, and extd. with  $\text{Et}_2\text{O}$  to yield 21 g. 1-cyano-5-methyl-4-hexen-3-one (I),  $b_{14}$  103-4°,  $n_D^2$  1.4725,  $d_4^{20}$  0.978; semicarbone, m. 130° (from dil.  $\text{MeOH}$ ). Use of excess KCN failed to give further addn. Oxidation with  $\text{KMnO}_4$  gave  $\text{Me}_2\text{CO}$  and succinic acid; ozonization gave  $\text{HCO}_2\text{H}$  and succinic acid; hydrogena-

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tion over Pt gave *l*-syno-5-methyl-3-hexanone, b.p. 133°;  $n_D^{20}$  1.4315,  $d_4^{20}$  0.9322 [semicarbazone, m. 139.5-40° (from 50% MeOH)]; hydrolysis of this satd. ketone by 30% H<sub>2</sub>SO<sub>4</sub> 6 hrs. at 70-5° gave  $\delta$ -isopropyl levulinic acid, m. 47° (from ligroin); semicarbazone, m. 147-8° (from 50% MeOH). Similar hydrolysis of the unsatd. ketone gave levulinic acid, sepd. by distn., and a poor yield of  $\text{Me}_2\text{C}(\text{CHCOCH}_2\text{CH}_2\text{CO}_2\text{H}$ , m. 73-4° (from water) [semicarbazone, m. 150°]; hydrogenation over Pt gave the above-described satd. acid, m. 47-8°. Use of more concd. acid gives only levulinic acid, due to cleavage of the unsatd. ketone. Addn. of 22 g. CH<sub>3</sub>CHCOCH<sub>2</sub>CClMe<sub>2</sub> in 22 ml. EtOH to 9 g. KCN in 15 ml. water raised the temp. to 42°, which was maintained 40 min.; stirring 3 hrs. and letting stand overnight gave 4.2 g.  $\text{Me}_2\text{C}(\text{CHCOCH}_2\text{CH}_2\text{CO}_2\text{H}$  and 10 g. I. Addn. of KCN to  $\text{MeOCH}_2\text{CH}_2\text{COCH}_2\text{CMe}_2$  in EtOH in the presence of AcOH gave only tars; the use of large aunts. of AcOH gave no reaction.

G. M. Kosolapoff

Heat-transfer factor of the radiant-heat surface in the tube-still furnace. E. P. Raigurovskii. *Azerbaidzhanskoe Neftegazovye Khodyashchee* 1933, No. 6, 7, 8-9. The calcs. of the above factor are presented. A. B.

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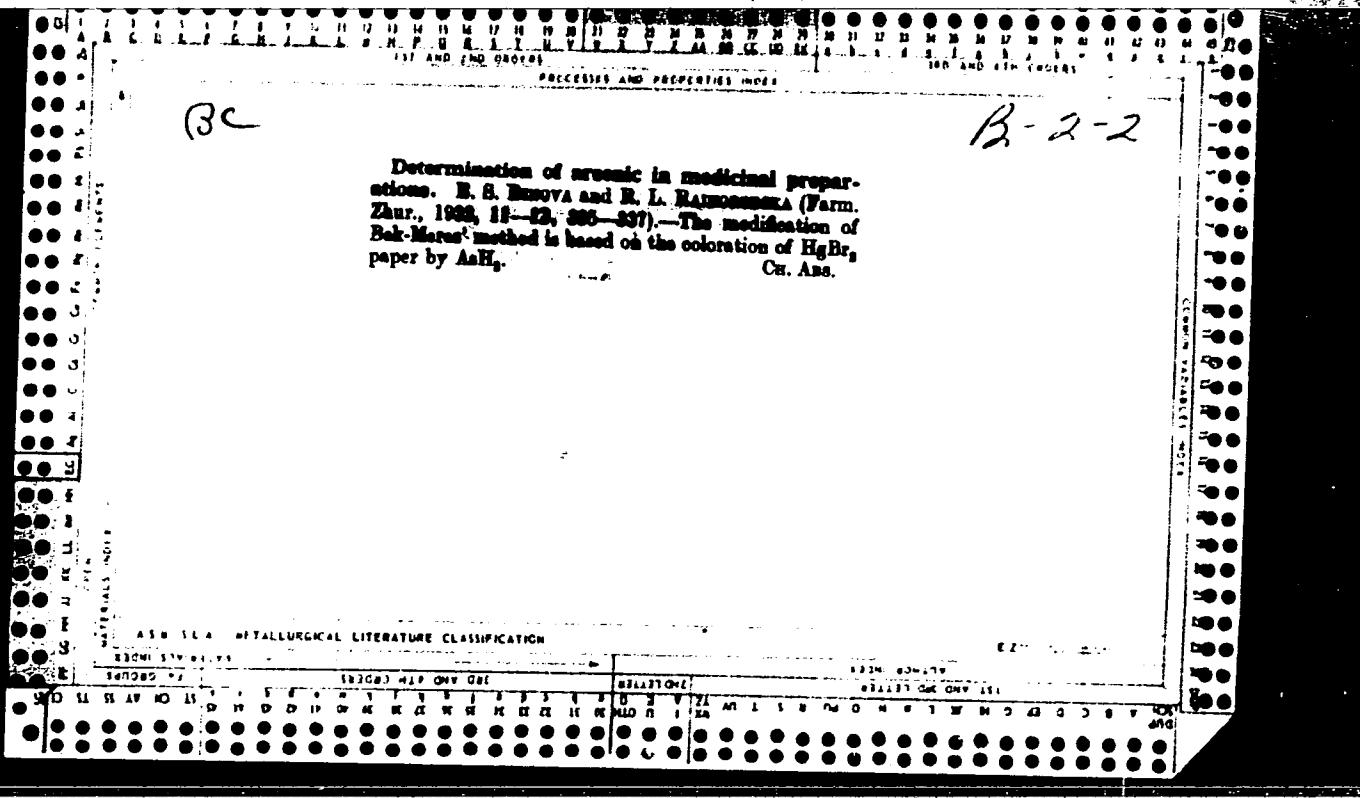
22

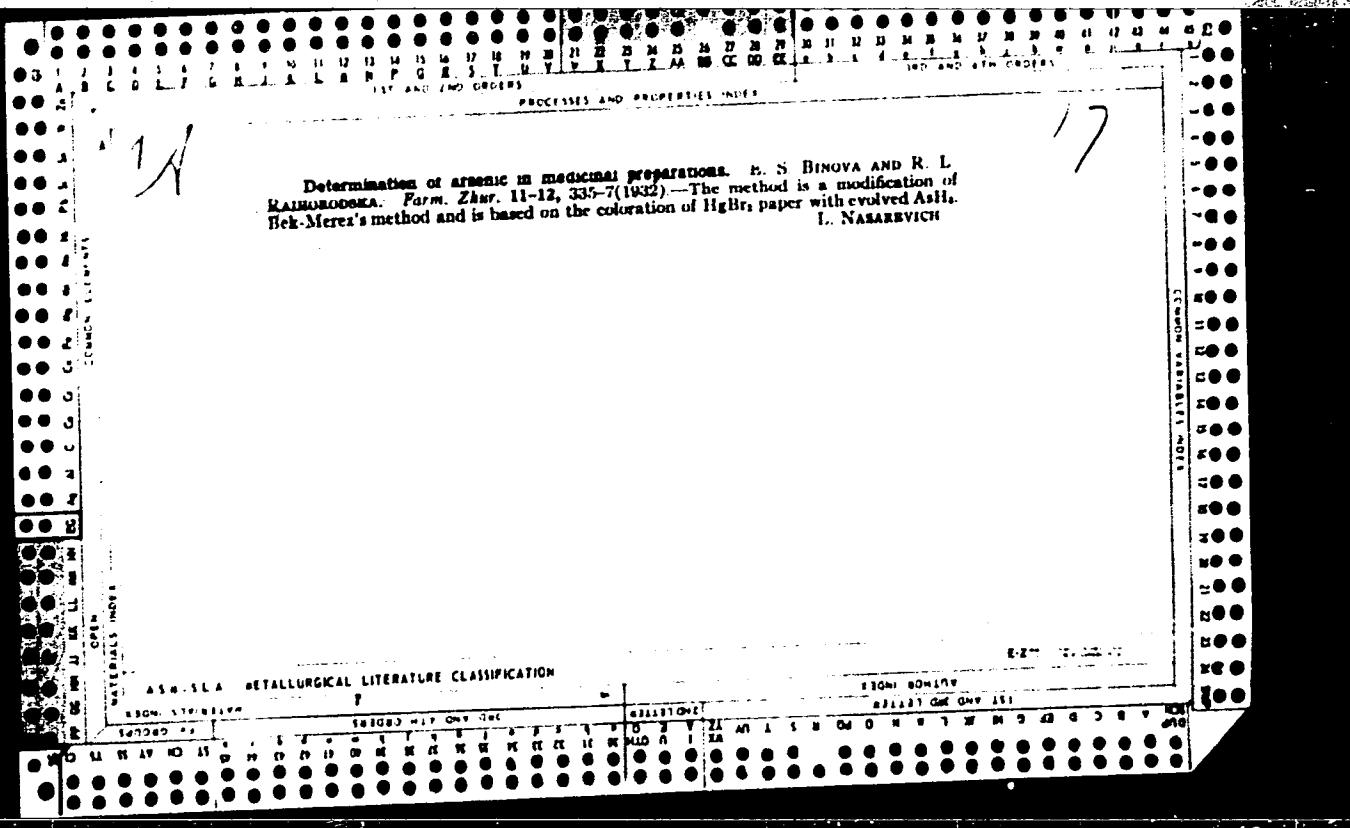
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3. Influenza
4. Pathological changes in the nervous system in grippe, Medyer. zhur., 22, no. 1, 1952.
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Chinese treatise "Mathematics in nine books." Ist. mat.  
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New reconstruction of certain problems from ancient Egyptian and  
Babylonian texts. Ist.-mat.issl. no.11:171-182 '58.

(MIRA 12:1)

(Mathematics, Babylonian)

(Mathematics, Egyptian)

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Istoriko -matematicheskiye issledovaniya, vyp. 11 (Research in  
Mathematical History, Nr 11) Moscow, Fizmatgiz, 1958. 792 p.  
3,000 copies printed.

Eds. (Title page): Rybkin, G.F. and Tushkevich, A.P.; Ed. (Inside  
book); Konoplyankin, A.A.; Tech. Ed.: Murashova, N. Ya.

PURPOSE: This book is intended for mathematicians and others  
interested in the history of mathematics, and may serve as the  
basis for a suitable university text on the history of mathematics,  
thereby filling the most serious gap in Soviet mathematical  
literature.

COVERAGE: This book contains reports made by members of the section  
on the history of mathematics at the Third All-Union Mathematical  
Congress which discussed problems of the history of mathematics and  
various articles on the significance of the history of mathematics

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Problems from Ancient Egyptian and Babylonian Texts

171

RAIK, A.Ye.

Ivan Mikheevich Pervushin, the mathematician from the Urals. Ist.-  
mat. issl. no.6:535-572 '53.  
(Pervushin, Ivan Mikheevich, 1827-1900) (MLRA 7:9)

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SC: u-3042, ll, March 53, (Letopis 'nykh Statej, No. 9, 1949)

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Raik, I.I., ? -1953)

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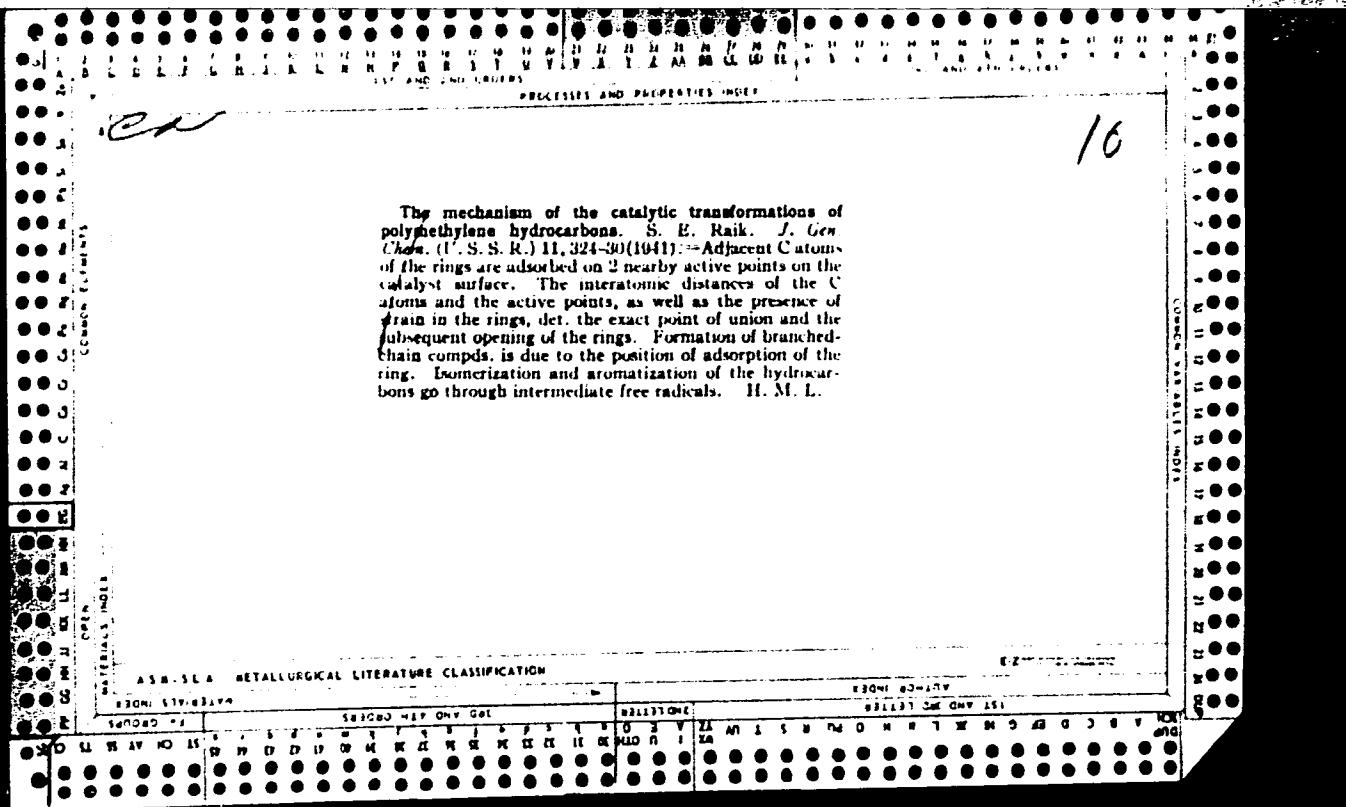
KHETAGUROV, G.I.; RAIK, I.O.

Results of four years of investigation on penacillin therapy of  
syphilis. Sovet. med. no.5:21-23 May 1951. (CLML 20:9)

1. Of Leningrad Skin-Venereological Institute of the Ministry of Public Health RSFSR (Director of Institute and Scientific Supervisor of Syphilological Department--Prof. S.Ye. Gorbovitskiy).

BAL'AKA, S.V.; RAIK, S.Ya.; DEGYAREVA, V., red.

[Pectic substances and their importance in the national economy] Pektinovye veshchestva i ikh znachenie v narodnom khoziaistve. Kishinev, Kartia Moldoveniasko, 1963.  
17 p.  
(MIRA 17:12)



CA

10

The dehydrocyclization of substituted pentanes and of  
gem-substituted hexanes. A. A. Balandin and S. E.  
Raik. *Compt. rend. acad. U.R.S.S.* 50, 101 (1947)  
(in English).—Three postulates are given in an explana-  
tion of the dehydrocyclization of substituted  $\alpha$ -pentanes  
(cf. *C.I.* 40, 1700) by way of a skeleton isomerization.  
These postulates, which occur simultaneously, are: (1)  
in such substituted pentanes and hexanes a 5-membered  
ring compd. is formed; (2) the 5-membered ring is iso-  
merized into a hexahydroaromatic 6-membered ring in such  
a way that the  $\alpha$ -C atom of the side chain enters into the  
neighboring C-C bond of the 5-membered ring; and (3)  
this entering of the new C takes place into those bonds of  
the 5-membered ring that are farthest from the newly  
formed bond of the 5-membered ring formed from the parent  
hydrocarbon. Thus, from 2,2,1-trimethylpentane, *p*-  
xylene is formed, whereas from 2,2-dimethylcyclohexane  
*m*-xylene is formed. Of the 9 cases investigated, the re-  
sults are in quant. agreement with the above postulates in  
7 cases, with but 1 case being markedly divergent.  
Myron Q. Webb

RAYK, S. Ye.

Mechanism of catalytic opening of rings. S. E. Kulin.  
Problemy Kinetiki i Kataliza, Akad. Nauk S.S.R. 6, *FCL*  
Geterogennyi Kataliz, 259-61(1949).—Catalytic hydrogenation of cyclobutanes with side chains in the presence of Pt on an active C carrier, yielded open chain, branched hydrocarbons. Thus, the ring was broken preferentially in bonds remote from the point of attachment of the side chain. It is proposed that the catalyst attacks first one C atom, then the second neighboring C and causes stretching of the C—C bond. Considerations of steric hindrance and bond lengths indicate that the probability for a given C to become attached to the catalyst is related to the no. of H atoms attached to this C. The probability for the formation of the iso-compd. is 80%, since formation of the normal hydrocarbon would require that the tertiary C is attached before the ring opens. Similar considerations can be applied to the opening of pentamethylene ring; the yield of normal hydrocarbons from Me cyclopentene is 12% as compared to the predicted 11%. The actual yield of 2.4 dimethylpentene from 1,3-dimethylcyclopentane is 45%, while 50% is predicted. Several difficulties, however, are encountered in applying this mechanism to C<sub>4</sub> and C<sub>5</sub> rings, since it does not explain the greater difficulty to open a C<sub>4</sub> ring. The ability of the C atoms to rotate in C<sub>4</sub>, which is absent in C<sub>3</sub>, C<sub>4</sub>, and C<sub>5</sub> rings, may serve as explanation.

Andrew Dravnick

RAIK, S. Ye.

Verbatim: - "The extension of the theory of multipoles to the reaction of the dehydrocyclization of substituted pentanes and partial hexanes," Vestnik Mosk. un-ta, 1948, No. 12, p. 112-20, - Bibliog: 11 items

SC: U-4355, 14 August 53, (Letopis 'Khurnal 'nykh Statey, No. 15, 1949.)

CA

Use of the statistical method for the determination of the results of some contact reactions. S. E. Raik (Moscow State Univ.). *Vestn. Moskov. Univ.*, 6, No. 10, Ser. Fiz.-Mat. i Estestven. Nauk No. 6, 69-78(1951). — (1) The no. of events sufficient for the probability that one single event will be applicable, within a small stated uncertainty, to a large no. of events, is estd. with the aid of the principle of practical assurance. The problem is to det. the no.  $n$  of elementary acts of a given catalytic reaction necessary for the statistical probability to differ from the probability derived on the basis of the mol. structure, by a magnitude  $\epsilon$ , not to exceed the error. For reactions where the products can be detd. to within 0.01%,  $\epsilon$  is of the order  $10^{-6}$ . With the aid

of Laplace's functions,  $\epsilon = (4.48/\epsilon)^{1/2}pq$ , where  $p$  and  $q = 1 - p$  are the elementary probabilities of the alternative outcomes of a ring-opening reaction. As an example,  $p$  and  $q$  correspond to: methylcyclobutane  $\rightarrow$  pentane ( $p = 0.14$ ) + 2-methylbutane ( $q = 0.86$ ); methylcyclopentane  $\rightarrow$  hexane + 3-methylpentane (0.33) + 2-methylpentane (0.67); 1,1-dimethylcyclopentane  $\rightarrow$  3,3-dimethylpentane (0.25) + 2,2-dimethylpentane (0.75); 1,2-dimethylcyclopentane  $\rightarrow$  3-methylhexane (0.25) + 2,3-dimethylpentane (0.75); 1,3-dimethylcyclopentane  $\rightarrow$  (3-methylhexane + 2-methylhexane) (0.60) + 2,4-dimethylpentane (0.40). In all these instances,  $n$  is of the order of  $10^6$  mols., which corresponds to  $\sim 10^{-12}$  g. of substance undergoing reaction. Consequently, the application of *a priori* probabilities under the usual conditions of catalytic expts. is fully justified. (2) Probabilities of cyclization of hexanes are estd. on the assumption that the hexane is adsorbed by one C atom, and cyclization occurs if the hexane folds in such a way that the 6th C atom comes into close contact with the C atom adsorbed. On account of free rotation around C-C bonds, there is a definite finite probability  $P$  that this will happen. The probability that the 1st 3 C atoms lie in one plane is taken to be a certainty. The probabilities that the 4th and the 6th C atoms will take positions favorable to ring closing are equal, and are shown to be  $1/2$  each, i.e. the probability that both take such positions is  $1/4$ . For the 6th C atom,

over

the probability is  $1/n$ , and hence the total  $P = 1/n$ . That is the probability that a mol. will possess a configuration favorable for the closing of a 6-membered ring at the moment of adsorption of one of its C atoms. The fraction  $P = 1/n$ , multiplied by the probability of adsorption of a suitable C atom, gives the probability  $\rho$  of cyclization. (3) The no. of contacts  $c$  between a mol. susceptible of cyclization, and the catalyst, which leaves a stated fraction  $\beta$  of the reactant uncyclized, can be expressed by  $\eta' = \beta$ , or by  $\eta' = \gamma \leq \beta$ , where  $c$  is now the smallest integer satisfying the inequality. These relations permit, with a known  $\eta'$  and a prescribed  $\beta$  (or  $\gamma$ ), the calcn. of  $c$  and hence, if an assumption is made with regard to the no. of active points per unit amt. of catalyst, also the min. amt. of catalyst  $g$  necessary for the cyclization of a given amt. of hydrocarbon. As an example, if for catalysts of the type of  $\text{Al}_2\text{O}_3 + 10\%$   $\text{ZnO}$ , or bauxite,  $\sim 2 \times 10^6$  active points/g. catalyst, in order to cyclize 0.1 ml. hexane ( $4.85 \times 10^{-6}$  mols.) down to  $\beta = 10^{-3}$ ,  $c = 754,15$  contacts and, hence, 21430 g. catalyst is necessary. This amt. decreases if the prescribed  $\beta$  is made smaller; e.g., for  $\beta = 10^{-1}$ ,  $c = 2005$  g. The same dependence of the fraction reacted on the probability is expressed by the formulas,  $\eta' = \beta$  and  $K_p = \log [1/(1 - \alpha)]$ , where  $\alpha = 1 - \beta$ . The exptl. yields of cyclization, from data of Kazanskii and Plata (C.A. 33, 9294<sup>a</sup>) on Pt, and of Hoog, Verheus, and Zuiderveld (C.A. 33, 9293<sup>b</sup>) on  $\text{Cr}_2\text{O}_3$ , can be represented equally well by either of these formulas. The factor dets. yields of aromatization is the probability of cyclization of the given hydrocarbon. N. Thon

MORDKOVICH, M.S.; RAIK, S.Ya.; ARASIMOVICH, V.V.

Losses of pectin substances in the production of tomato paste.  
Kons. i ov. prom. 18 no.11:19-21 N '63. (MIRA 16:12)

1. Moldavskiy nauchno-issledovatel'skiy institut pishchevoy  
promyshlennosti (for Mordkovich). 2. Institut fiziologii i  
biokhimii rasteniy AN Moldavskoy SSR (for Raik, Arasimovich).

RAIK, S.Ye.; KAZANSKIY, B.A.

Production of ethyl ester of cyclobutane-1, 1 dicarboxylic acid. Vest. Mosk. un. no. 3:125-128 Mr '53. (MLRA 6:6)

1. Laboratoriya organicheskogo kataliza.

(Ethyl esters)

Raika, E.

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R001344020008-9"

The pathogenesis of urticarial pruritus. III. Quantitative evaluation of the effects of various drugs on morphine-induced pruritus. E. Raika, S. Korossy, and Marianne Gózony (Stevens' Hosp., Budapest). *Dermatologica* 112, 81-107(1960)(in German)(English summary); cf. *C.A.* 50, 1178b.—Of 78 drugs, of various types, tested (3008 tests) 44 were found, in the majority of cases, to prevent the occurrence of exptl., morphine-induced, pruritus. In no case was 100% inhibition obtained. Henry B. Hastie

3  
Med

RAIKH, I.Ya., inzhener; FAYNGERSH, Ya.D., inzhener.

Mechanical method of making openings in masonry walls. Mekh.stroi.  
11 no.11:28-29 N '54. (MIRA 7:12)  
(Masonry) (Drilling and boring)

11/1  
Determination of the spectral energy distribution of the light emitted by an incandescent lamp with voltage stabilizer when the intensity was subject to continuous change produced by the opening and closing of the slit of the spectrograph by means of a constant speed motor.

778.3 : 535.33

531  
**Photometry of Spectra of Variable Intensity and of Spectral Lines of Low Intensity.**  
J. D. RAIKHBAUM. *Bull. Acad. Sci. U.R.S.S., Ser. Phys.*, 9, 765-768, 1945.—In SLAVIN's "spectral energy" method—complete vaporization of the sample in an electric arc—it is necessary to understand the laws governing the integration by the photographic plate of the varying energy of emission. An account is given of the results of experiments in the photometry of continuous spectra from an incandescent lamp with voltage stabilizer when the intensity was subject to continuous change produced by the opening and closing of the slit of the spectrograph by means of a constant speed motor. *Brit. Abs. C.*

*m* //

<sup>10</sup>Spectrochemical Analysis of Alloys of Gold, Silver, Copper, and Tellurium.  
V. D. Raiklinam, L. D. Raiklinam, and M. G. Vinogradov (Zavod  
*Labs.*, 1940, 9, 104-108; *Brit. Chem. Abstr.*, 1944, 1C, 5). (In Russian.)  
Au-Ag alloys are evaporated in an arc, and the total amounts of the radioisotopes Au 2332.7 and Ag 2300.4 Å compared, using Betini's method (Joupol  
*Anal.*, 1931, 106, 590). These amounts can be compared by using a logarithmic sector or by microphotometry.

ASA SLA METALLURGICAL LITERATURE CLASSIFICATION

Spectral analysis of alloys of gold, silver, copper and tellurium. Ya. D. Rakhitbaum, L. D. Rakhitbaum and M. G. Voskobodnikov. *Zurnal chistoj i prikladnoj spektroskopii*, 9, No. 10 (1948).  
Kapitl. details for the spectral data of Ag, Cu and Te in Au nuggets of 3-5 mg. B. Z. Kamich.

7

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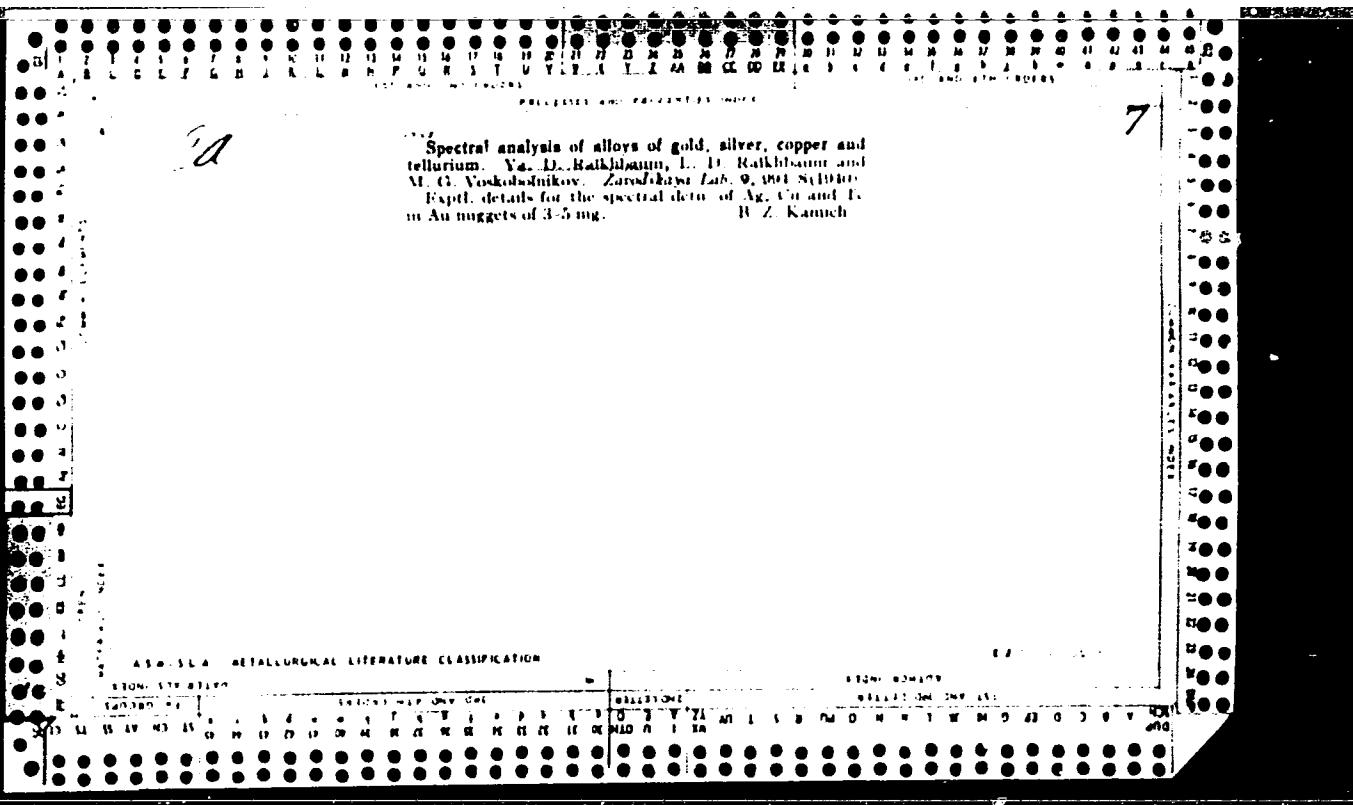
CIA-RDP86-00513R001344020008-9"

11  
M

**Spectrochemical Analysis of Alloys of Gold, Silver, Copper, and Tellurium.**  
Y. D. Raikhlbaum, L. D. Raikhlbaum, and M. G. Vaykobsonikov. *Zurnal  
Fizika*, 1940, **9**, 104-108; *Brit. Chem. Abstr.*, 1941, (C), 51. (In Russian)  
Au-Ag alloys are evaporated in an arc, and the total amounts of the radia-  
tions Au 23527 and Ag 23001 A. compared, using Betens's method (*Compt.  
rend.*, 1934, **198**, 500). These amounts can be compared by using a log-  
arithmic sector or by microphotometry.

ASA-SEA METALLURGICAL LITERATURE CLASSIFICATION

EZ



Determination of tungsten in minerals by the comparison of spectral energies. Yu. D. Ralikhman. Zavodskaya Lab. 8, 601-5 (1939).—W was detd. from the relative values of the spectral energies corresponding to 2 spectral lines of different elements. Co was added to the soln. in const. concn. and the selected pairs of lines were photometered to det. the relative values of the energies emitted by W and Co during complete vaporization in an arc of d. c. The nature of the chem. bond of W had no effect on the results. The varying thermal stabilities of the different W compds. govern only the rate of vaporization of the W atoms. Vaporization of W and Co is hindered by presence of large amounts of NaCl but it becomes more intensive as the temp. of the arc rises and the NaCl is vaporized.

Local spectral analysis of Au surfaces. Ya. D. Raikhan. *Zavodskaya Lab.*, 10, 108-70(1941); *Chem. Zentr.* 1943, II, 1118; cf. *C.A.*, 37, 2297. — Local spectral analysis offers a means of supplementing the mineralogical and quant. spectral analysis of Au ores. It is possible to det. the qual. compn. of films of Au plate. Only with samples of less than 0.5 mg. were unsatisfactory results obtained. In these cases the samples were dissolved in acid and the solns. analyzed. M. G. Moore

7

## AS-A-12A METALLURGICAL LITERATURE CLASSIFICATION

TA

Sensitivity & Spectrometry

771.534.56

187  
Photographic Action of Light of Variable Intensity. YA D. RAIKHBAUM. J.  
Tech. Phys. U.S.S.R., 15, 485-8, 1945.--Photographic plates are illuminated  
with light of a gradually increasing or gradually decreasing intensity. The  
optical density of the image at a constant exposure is, for an increasing intensity,  
smaller than for a decreasing one and greater than for a constant intensity  
and variable time. When the rate of variation of light intensity rises, density  
rises also. These results must be borne in mind when using density in spectro-  
graphic analysis. *Brit. J. Phot. Abs.*

F-A

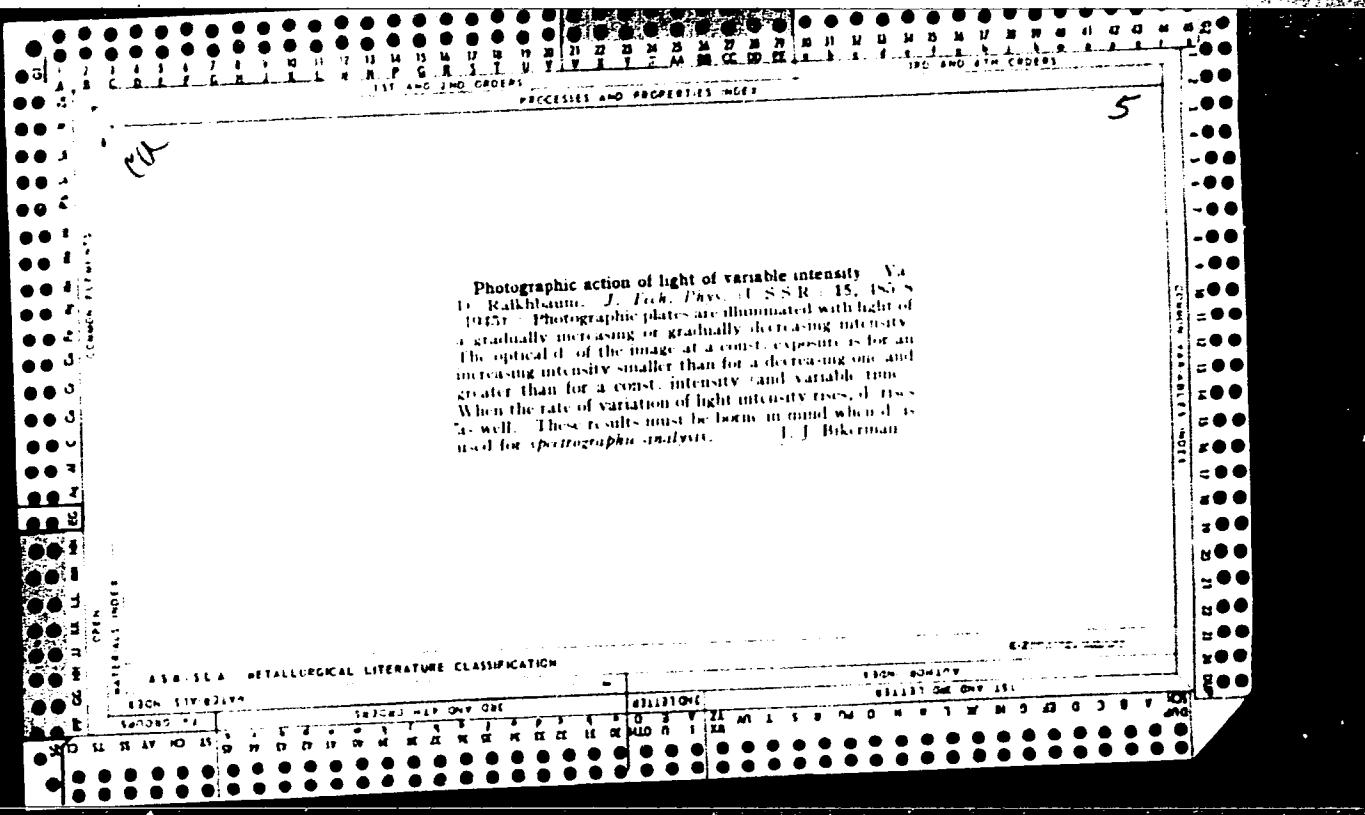
Preparation of Light-Sensitive  
M. T. and Spence

1145

77.021.1 : 544.62

Spectrographic Determination of Silver in Photographic Emulsions. Ya. D. RAIKHEAUM and Ya. M. DYMISHITS. Izvest. Akad. Nauk. S.S.R., Ser. Fiz., 1948, 12, 477-480.—Samples of the order of 0.04-0.1 sq. cm. of the emulsion are calcined and thoroughly mixed with a known amount (100-200 mg.) of a mixture of 25 mg. of anhydrous sodium sulphate, 25 mg. of anhydrous potassium sulphate, 50 mg. of graphite and 0.05 per cent of tin as standard. The determination is done in an a.c. arc under 10 amp., by the line pair Ag 3280.68-Sn 3262.33A. The log calibration curve is rectilinear between 0.0002 and 0.02 per cent silver. The probable error of a single determination is 8.4 per cent. The method was used to determine for various types of emulsions the photometric constants and the mean mass of the developed grains.

Chem. Abs.



Quantitative Spectral Analysis of Tin, Lead, and Bismuth Alloys. Ya. D. Atakhbaum (*Zurnal Lab.*, 1939, **8**, 1101-1105; *Chem. Zentral.*, 1942, **113**, (11), 1042; *C. R. Acad.*, 1943, **37**, 5332). —[In Russian.] The mathematical basis of spectral analysis is explained and various curves are given which were obtained by experiments with the Hilger spectrograph. Results accurate to within about 1% were obtained even with high concentrations of Sn, Pb, and Bi.

REF ID: A6510

ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION

E-2-12-1

Spectrographic determination of silver in photographic emulsions Ya. D. Rakhbaum and Ya. M. Dvinskaya  
*Izv. Akad. Nauk SSSR, Ser. Fiz.* 12, 177 (1948).

Samples of the order of 0.01-0.1 sq. cm. of the emulsion are calcined and thoroughly mixed with a known amount (100-200 mg.) of a mixt. of Na<sub>2</sub>SO<sub>3</sub>, 25, K<sub>2</sub>SO<sub>3</sub>, 25, graphite 50, and Sn 0.05%, as standard. The determin. is done in an arc, 4000-4500 A., under 10 amp., by the line pair Ag 3280 & Sn 3223 Å. The log-log calibration curve is rectilinear between 0.0002 and 0.02% Ag. The probable error of a single determin. is 8.4%. The method was used to determine the photometric constants and various types of emulsions, the photometric constants and the mean mass of the developed grains. N. Thom

RAIKHER, G. S.

Gruzooborot transporta v novoi piatiletke. Freight turnover in the new five-year plan (Zhel-dor. transport, 1946, no. 2-3 p. 33-41). "Additional details on freight traffic by region and commodity. Supplements Kovalev's pamphlet.

DLC: HE7.Z5

Gruzovye perevozki. The freight transport. (In Levin, B.I. Osnovnye voprosy piatiletnego planavosstanovleniya i razvitiia zheleznodorozhnogo transporta. Moskva, 1947, p. 111-136). DLC: HE3137.L4

Osvobodit' zheleznye dorogi ot korotkoprobezhnykh perevozok. To free the railroads from short distance shipments. (Zhel-dor. transport, 1945, no. 10 11, p. 70-73).

DLC: HE7.Z5

Ratsionalizatsiia zhelezhodorozh nykh perevozok v tret'em piatiletii. Rationalization of railroad shipments in the third five-year plan (Problemy ekonomiki, 1940, no. 2, p. 88-97)

DLC: HB9.P75

Za bor'bu s neratsional'nymi korotkoprobezhnymi perevozkami. Against inefficient short distance shipments (Sots. transport, 1940, no. 3, p. 8-11) "1935 and 1938 data on proportion and amount of such shipments.

DLC: HE7.S6

SO: Soviet Transportation and Communication A Bibliography, Library of Congress Reference Department, Washington 1952, Unclassified

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**CIA-RDP86-00513R001344020008-9**

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CIA-RDP86-00513R001344020008-9"

*22*

## CHARACTER AND PROPERTY INDEX

100 AND 4TH DECADE

**Characteristics of  $\alpha,\beta$ -unsaturated ketones. II.** V. I. Krasov and I. I. Rakhner. *J. Gen. Chem. (U. S. S. R.)* 13, No. 13 (1943) (English summary); cf. *C. A.* 35, 30389. It was shown that the displacement of ketones by aldehydes from  $\alpha,\beta$ -unsatd. ketones is connected with the preliminary hydrolytic cleavage of the latter into its components. BaII, treated with mesityl oxide in the presence of dil. NaOH, yielded small amounts of benzylideneacetone and benzylidenemesityl oxide, the latter being isolated as the tetrabromide, m. 118° (from R(ONa)). The reactions were conducted by prolonged standing at about 10°, with a longer period noticeably increasing the yield of benzylideneacetone. Me<sub>2</sub>CrO was shown to be inert to condensation on treatment with piperidine acetate on prolonged heating on a steam bath; the same result was obtained when MeEtCO was used. However, 25 g. BaII, 30 g. Me<sub>2</sub>CO, 5 g. piperidine and 3 g. AcONa yielded, after 20 hrs. heating, 71% benzylideneacetone, m. 41.5°, while similar reaction with mesityl oxide gave crude benzylidenemesityl oxide, b.p. 172-3°, identified as the tetrabromide. Thus, piperidine acetate may be recommended for aldehyde-ketone condensations.  
G. M. Kosolapoff

10

## ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION

ECONOMIC SECTION

ECONOMIC SECTION

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ALL INFORMATION CONTAINED  
HEREIN IS UNCLASSIFIED

1. [REDACTED] (Top Secret) (SAC) (CIA) (TECHINT) (INFO)  
[REDACTED]

1. Anti-aircraft units. 2. Fire study. 3. Armor II, L.D., [REDACTED]

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R001344020008-9"

RAIKHER, M.E., professor; KAMINSKIY, I.N., inzhener; NIKOL'SKIY, V.S.,  
redaktor; SUROVA, V.A., redaktor; ANDREYEV, G.G., tekhnicheskij  
redaktor.

[Complex time study in coal mines and pits] Kompleksnyi khronometrazh  
na ugol'nykh shakhtakh i kar'erakh. Moskva, Ugletekhizdat, 1954.  
203 p.  
(Time study) (MIRA 8:5)

RAIKHER, M.E.; KAMINSKIY, I.N.

[Complex timekeeping in coal mines and pits]. Kompleksnyi khronometrazh na ugel'nykh shakhtakh i kar'erakh. Moskva, Ugletekhnizdat, 1954. 204 p. (MLRA 8:3D)

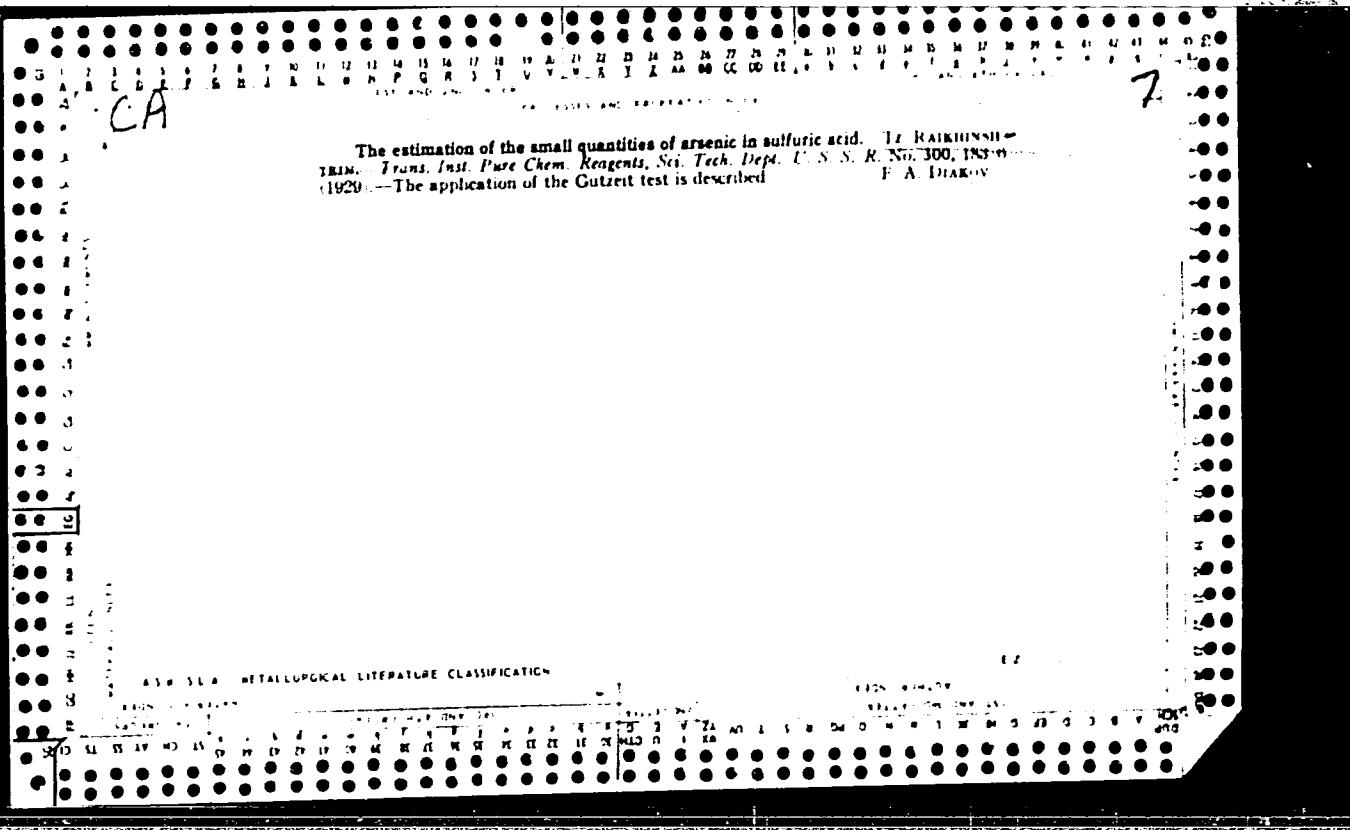
Determination of molybdenum and lead by Fajans method. T. Rakhmishstein and N. N. Konchalov. *J. Russ. Chem. (U.S.S.R.)* 3, 531 (1933); cf. *C.A.* 27, 2300. --A comparison between the methods of Fajans, Wiley, Sacher, Chandler and Alexander and Wiley, Ambrose and Bowers, for the detn. of Mo and Pb, showed that Fajans' method is the most rapid and accurate.  
S. I. Maderovsky

AMSLA METALLURGICAL LITERATURE CLASSIFICATION

Determination of molybdenum and lead by Fajans  
method. I. Rakhmishstein and N. N. Korobov.  
Gen. Chem. U.S.S.R., 1951, 31, 314-327.  
280. —A comparison between the methods of Fajans,  
Wiley, Sacher, Chandler and Alexander and Wiley,  
Brose and Bowers, for the detn. of Mo and Pb, showed that  
Fajans' method is the most rapid and accurate.

S. I. Midorsky

AMERICAN METALLURGICAL LITERATURE CLASSIFICATION

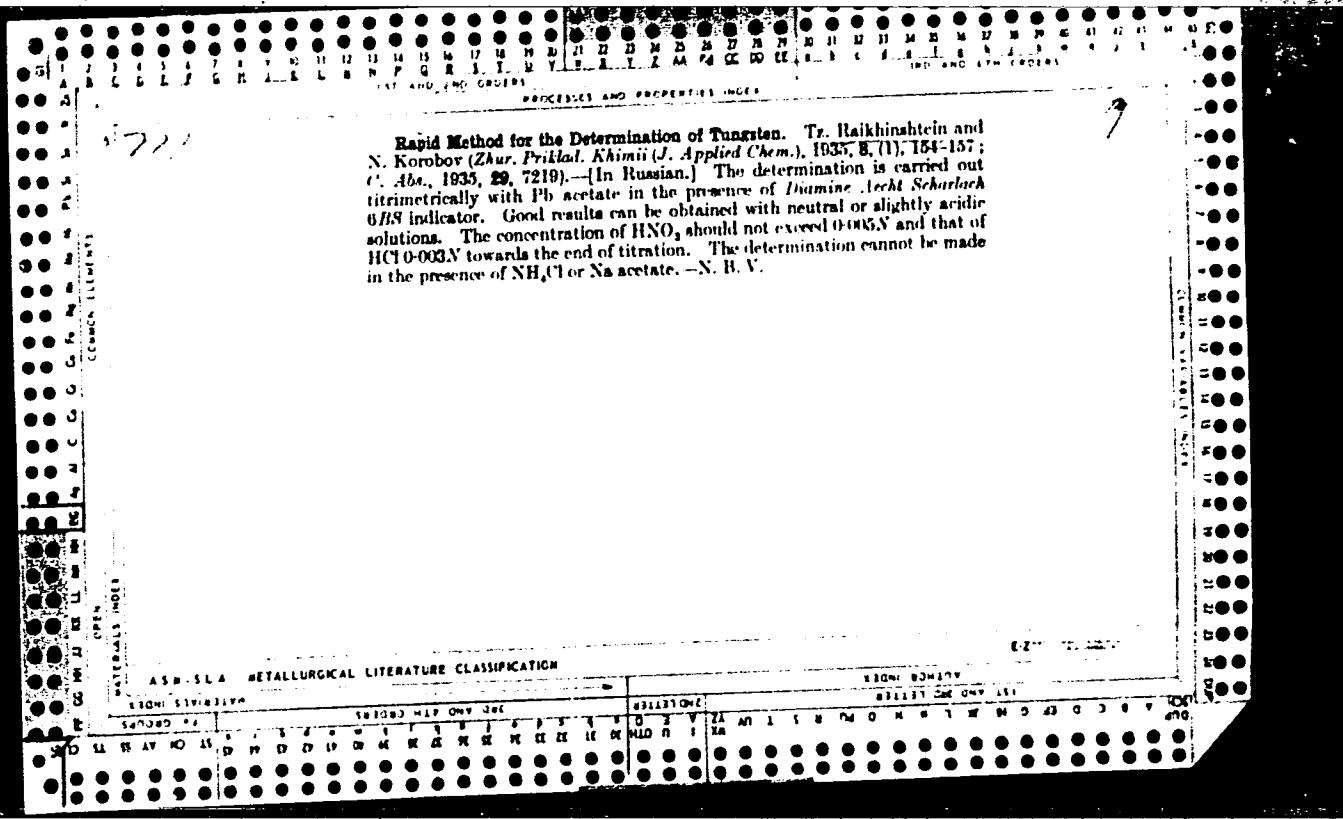


1ST AND 2ND GROUPS  
PROCESSES AND PROPERTIES INDEX

Determination of molybdenum and lead by the method of Fajans. I. Tz. RAI-KHIMSKII AND N. KOROBOV. *J. Gen. Chem. (U. S. S. R.)* 2, 601-5(1932) —  
 (C. A. 15, 222; 17, 3633, etc.) developed a method for the volumetric detn. of Ag halides with the application of adsorption indicators, which forms the basis of the present method for the detn. of Pb and Mo.  $(\text{AcO})_2\text{Pb}$  in  $\text{H}_2\text{O}$  gives with a little alizarin red (I) a violet soln., and with an excess a violet ppt. of the Pb salt of I; on addn. of  $(\text{NH}_4)_2\text{MoO}_4$  to the soln. the ppt. of  $\text{PbMoO}_4$  formed completely absorbs, at an adequate concn. of I, the Pb salt of I with violet color until the equiv. amt. of  $(\text{NH}_4)_2\text{MoO}_4$  is added. With the slightest excess of  $(\text{NH}_4)_2\text{MoO}_4$  the anion  $\text{MoO}_4^{2-}$  displaces the anion of I, and the color of the ppt. is sharply changed to rose with orange tint. Any further addn. of  $(\text{NH}_4)_2\text{MoO}_4$  does not affect the color of the ppt. In the expts. 0.5 cc. 1% I was added to the soln. of  $\text{Pb}(\text{NO}_3)_2$  or  $\text{Pb}(\text{OAc})_2$ , and the soln. was brought to a boil and titrated with a standard soln. of  $(\text{NH}_4)_2\text{MoO}_4$ . For the detn. of Mo, an excess of standard soln. of Pb salt was added to the soln. of a molybdate, the soln. was brought to a boil, 0.5 cc. 1% I was added, and the excess of Pb titrated back with a standard molybdate soln. The standard solns. of  $\text{Pb}(\text{NO}_3)_2$  and  $\text{Pb}(\text{OAc})_2$  were prep'd. by adding aq.  $\text{NH}_3$  to the ppt. of  $\text{Pb}(\text{OH})_2$  just forming and dissolving the ppt. with AcOH. The solns. used were 0.1235 N, 0.1133 N and 0.0151 N with concns. of AcOH of 0.1-0.4 N. The vol. of all titrated solns. was 40 cc. The titer of Pb solns. was detd. as  $\text{PbSO}_4$  and checked by electrolysis, and that of  $(\text{NH}_4)_2\text{MoO}_4$  by evapg. to dryness and gently igniting in a Pt dish. Good results were obtained at different concns. of the titrated solns., but with the concns. of a molybdate below 0.01 N the amt. of I should not exceed 0.5 cc. 1% soln. The method can be used for the detn. of oxides of Pb by dissolving them in  $\text{AcOH}$  or  $\text{HNO}_3$  and neutralizing with  $\text{NH}_3$ , in which case the concn. must be below 1 N for  $\text{NH}_3\text{OAc}$  or 2 N for  $\text{NH}_3\text{NO}_3$ . Titration of Pb and Mo by this method in the presence of the acetates and nitrates of heavy metals and other elements which form difficultly sol. salts with  $(\text{NH}_4)_2\text{MoO}_4$  is impossible.

CHAR BLANC

## ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION



Oxidation reduction indicators. I. New indicators for  
the titration of tin and antimony by bromates. II.  
G. Rakhmashin. *J. Applied Chem. U.S.S.R.*, 6,  
1179 (in German). 14,500 words. Source cited in  
U.S.A. HCL vol. 14, no. 1, with decomposition of  
benzoporphyrin. Basic indicator. Tin can be titrated by  
bromate in U.S.A. HCL soln. with benzoporphyrin  
bromate in U.S.A. HCL soln. with benzoporphyrin  
H. M. Proc. 66

Potentiometric determination of sulfates. I. I.  
Zhukov and Tz. G. Rakhimdzieva. *J. Gen. Chem.*  
(U.S.S.R.) 4, 802-8 (1934).—As the foundation for  
this detn. Kolthoff's method based on the use of Pb  
tri-ferrocyanide as an indicator electrode was used.  
Portions of  $\text{Na}_2\text{SO}_4$ , soln., were treated with alc., some  
 $\text{K}_3[\text{Fe}(\text{CN})_6]$ , and some  $\text{K}_4[\text{Fe}(\text{CN})_6]$ , and titrated with  
 $\text{Pb}(\text{NCH}_3)_4$ , soln. V. D. Karpenko

ARISTOV, Grigoriy Andrianovich; RAKHLIN, I., redaktor; SHIVELIEVA, A.,  
redaktor; IGHAT'YEVA, A., tekhnicheskiy redaktor

[Infinite universe] Vselennaya beskonechna. [Moskva] Mo-  
skovskii rabochii, 1955. 110 p. (MIRA 9:3)  
(Cosmogony)

PAREKHO, Pavel Petrovich; RAKHLIN, I.Ye., redaktor; YARMAKVA, Ye.a.,  
tekhnicheskiy redaktor

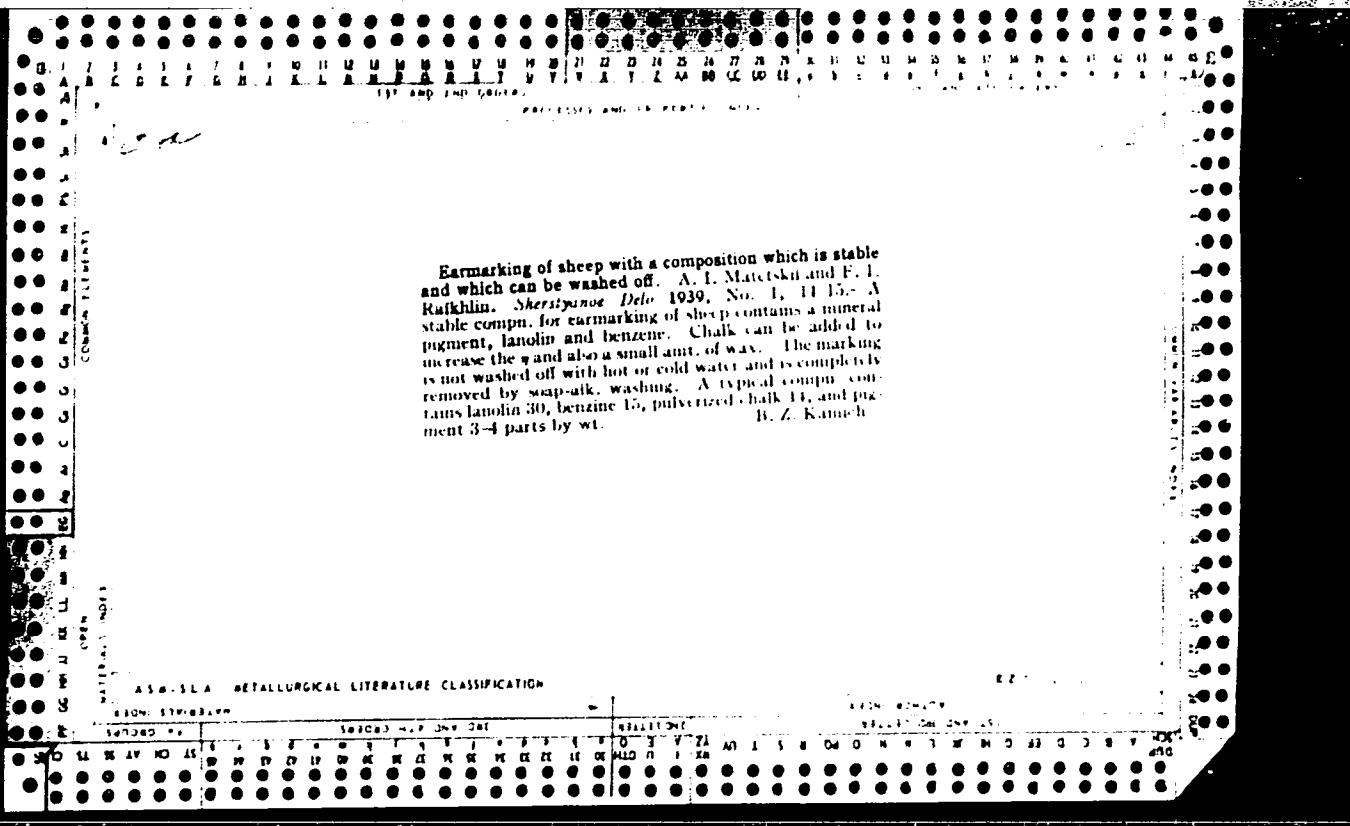
[In the world of stars] V mire zvezd. Moskva, Gos.izd-vo tekhniko-  
teoret. lit-ry, 1957. 95 p. (Populiarnye lektsii po astronomii,  
no.5) (MLRA 10:1C)  
(Astronomy)

*Ca*

Use of cationic soaps. A. I. Matetskii and F. I. Rakshin. *Tekstil. Prom.* 1941, No. 5, 35-6; *Chem. Zentr.* 1943, I, No. 4, 463.—Cationic soaps were prep'd. from cetyl and octodecyl ales. and mixts. of high-mol. ales. produced from cottonseed oil and seal oil. The following compds. were used as bases: pyridine, pyridine bases (b. p. 142-53°), trimethylamine, diethylamine. Expts. in the washing of wool and dyeing of cotton- and half-wool material showed cationic soaps are valuable products, and their use in the dyeing and finishing of fiber is desirable.  
Sonya G. Machelson

ASIN: A4 METALLURGICAL LITERATURE CLASSIFICATION

E2 1940-1949

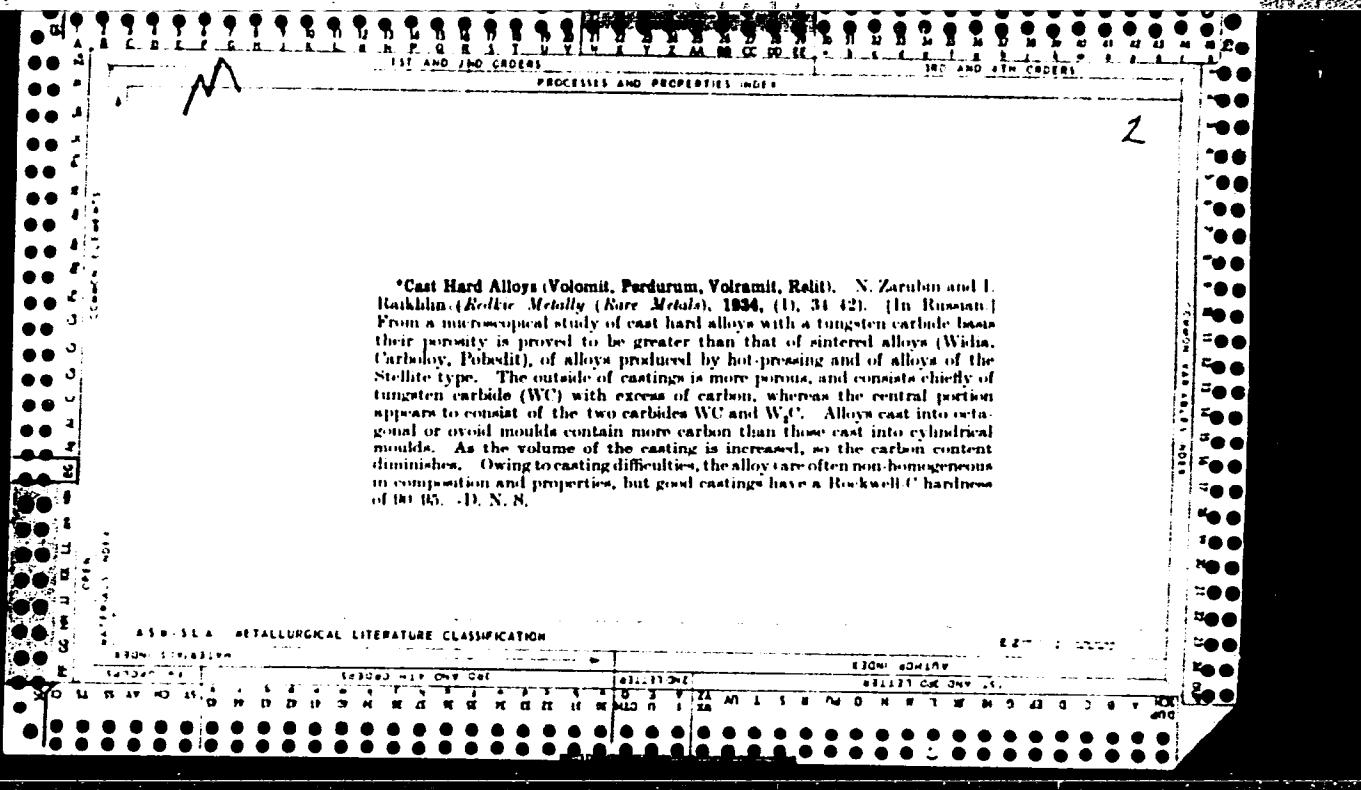


CA

22

**Stabilization of mineral-oil emulsions.** I. M. Rakhlin, *Tekstil. Prom.*, 7, No. 2, 34-5 (1947). In an effort to eliminate olein from wool-scouring emulsions, emulsions of solar oil and spindle oil, together with Kontakt (sulfonated naphthenic acid) and Nekal (butylnaphthalene-sulfonate) detergents were tested for stability. An emulsion of 20-25% spindle oil, 8% olein, 4% NH<sub>3</sub> and 68-63% H<sub>2</sub>O was stable after 240 hrs. Substitution of solar oil in this formula reduced this time to 24 hrs. An emulsion contg 3.5% Nekal gave 240-hr. stability with 20-30% spindle oil and 77-67% H<sub>2</sub>O. Substitution of solar oil decreased time of stability to 24 hrs. Increasing Nekal content to 7-10% reduced stability to 3 hrs. An emulsion consisting of 4% Kontakt, 20% spindle oil, 1% Na<sub>2</sub>CO<sub>3</sub>, and 75% H<sub>2</sub>O was recommended for use. M. S.

CONT'D



RAIKHLIN, N.T. (Moskva)

Compensatory adaptation modifications in the lungs in pneumoconioses.  
Arkh.pnt. 18 no.3:18-23 '56  
(MIRA 11:10)

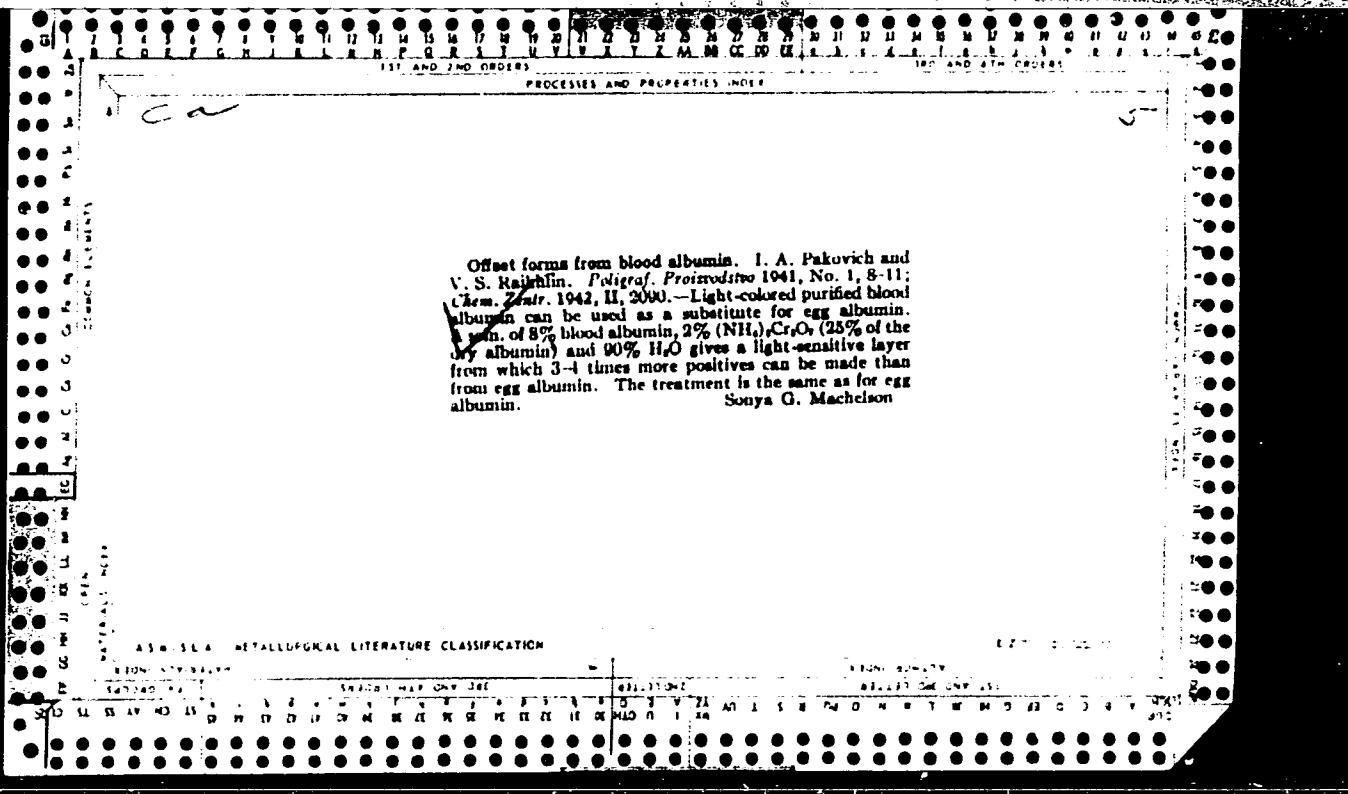
1. Iz kafedry patologicheskoy anatomii (zav. - chlen-korrespondent  
AMN SSSR prof. A.I. Strukov) I Moskovskogo ordena Lenina meditsinskogo  
instituta imeni I.M. Sechenova.  
(PNEUMOCONIOSES, physiol.  
compensatory adjustment modifications of lungs (Rus))

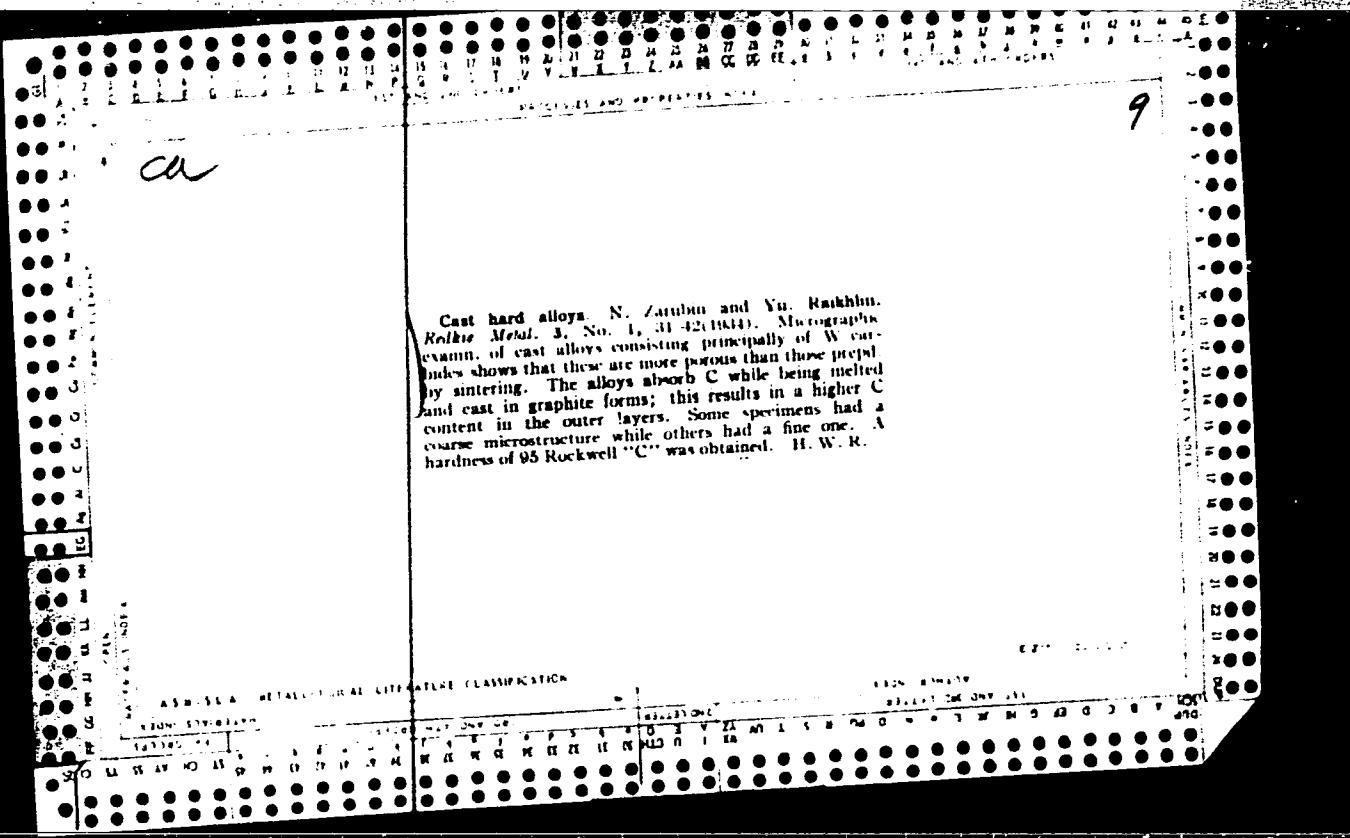
*CH*

7

Rapid method for determining tungsten. D. Raskin, J. A. Kostyuk. *J. Applied Chem. U.S.S.R.* 8, 283-7 (1958). The determination is carried out with phenacetate (ultrometrically) in the presence of "diamine azo-licht-scharlich 6 BS" indicator. Good results can be obtained with neutral or slightly acidic solutions. The excess of HNO<sub>3</sub> should not exceed 0.005 N and that of HCl 0.03 N toward the end of titration. Na and NH<sub>4</sub> nitrate are permissible in higher amounts. The determination cannot be made in the presence of NH<sub>4</sub>Cl or CH<sub>3</sub>COONH<sub>4</sub>. This method may also be used for deter. Pb (in W solns). Nine references.  
A. A. Raskin, 3

434 SLA METALLURGICAL LITERATURE CLASSIFICATION





## MEDICAL AND INDUSTRIAL EMPLOYMENT.

Dr. E. I., Ed., Professor

...olodye redkih metallov (Toxicology of Rare Metals) ...  
...nia 1953. 335 p. 1500 copies printed.

A. A. Khanidullin; Tech. Ed.: Yu. S. Bel'chikova.

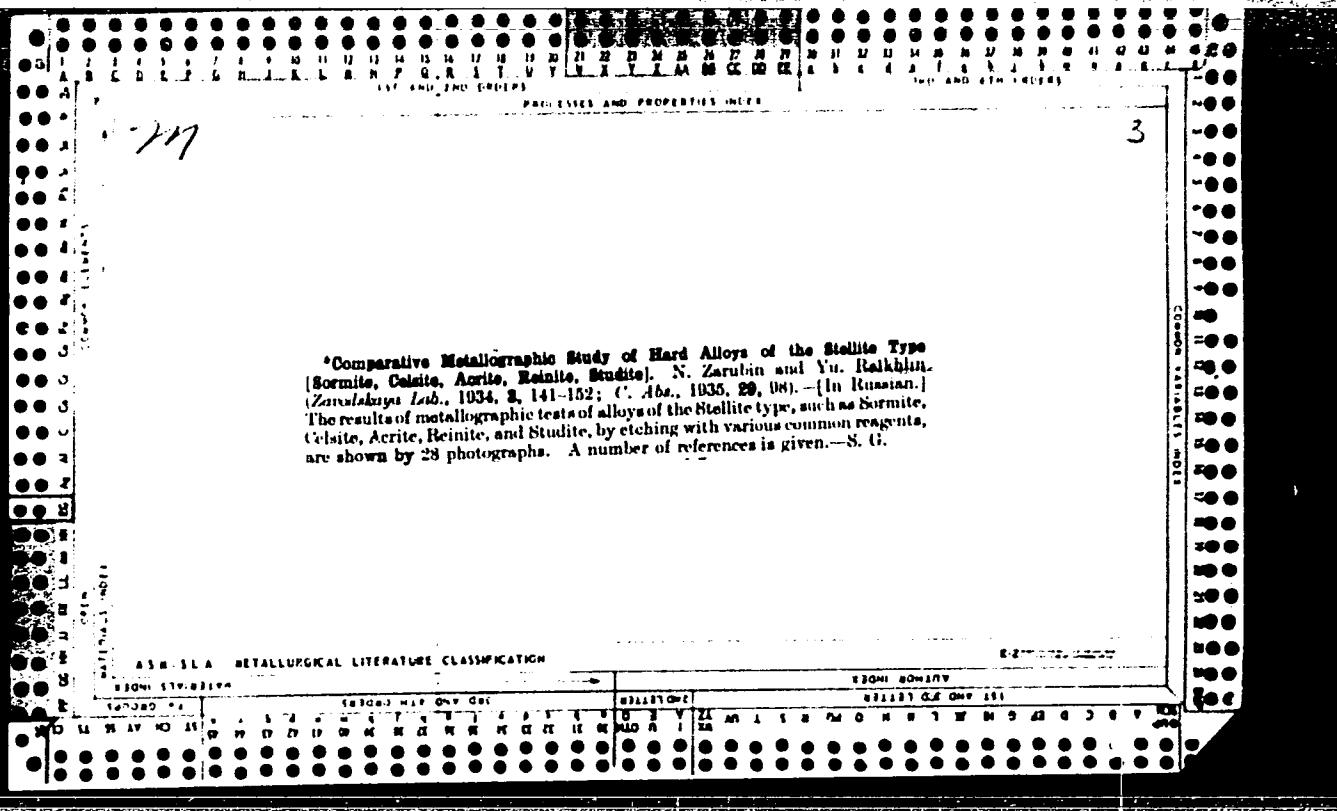
... a general information on the toxic effects of rare

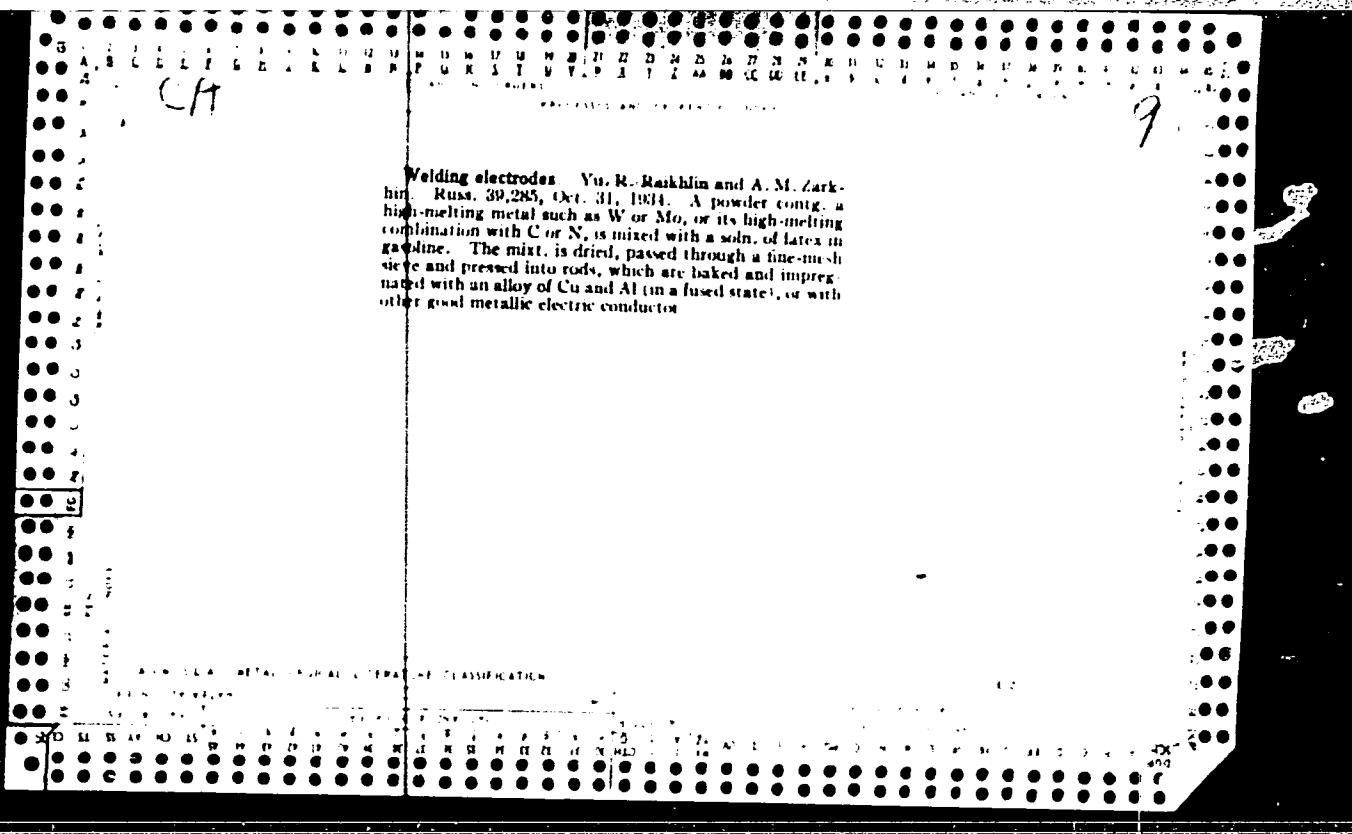
metals, their absorption and clinical application of rare  
metals in industry. The clinical picture and  
method of rare-metal poisoning is also given. There are 37  
appendices.

...E. Experimental Studies of the Effect on an Organism of  
Rare, Dispersed, and Other Metals Used in Industry ...  
Their Compounds.

10. Nickel. Yu. Yu. Mogilevskaya	151
11. Cobalt. Z. S. Kaplun (Second ed)	161
12. Uranium compounds. G. I. Rumyantsev	176
13. Zinc and Zinc oxide. I. Ya. Mogilevskaya	187
14. Rare earths. O. Ya. Mogilevskaya and N. I. Raikhlin	195

Comparative metallographic study of hard alloys of the Stellite type. N. Zarubin and Yu. Ralkhin. Zavodskaya Lab. 8, 141-63.—The results of metallographic tests of alloys of the Stellite type, such as Sormite, Celcite, Acrite, Reinitite, Percite, Studite, by etching with various common reagents are shown by 24 photographs. Over 24 references.





CA

9

**Welding electrodes.** Yu. R. Ranklin and A. M. Zarkhin. Russ. 38,715, Sept. 30, 1934. Electrodes are prep'd. from powdered hard metals, such as W, coated (electrolytically or by substitution) with a metal layer that is a good conductor, such as Cu ppt'd. from a soln. of its salts

ADDITIONAL DETAILS AND LITERATURE CLASSIFICATION

*4/16/01* *Class*

Method for making metalloceramic products. Yu. N.  
RALKIN AND A. M. ZARKHIN. Russ. 56,976. April 30,  
1910. 406. 17. Such products are made from powdered  
metals to which may be added diamond dust. The molded  
powder is saturated with molten copper under pressure.  
M. Ho

"APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R001344020008-9

RAIKHMAN, A.Z., inzh.

Controlling steam turbine disks. Energetik 5 no.9:17-19 S '57.  
(MIRA 10:10)  
(Steam turbines)

APPROVED FOR RELEASE: 03/20/2001

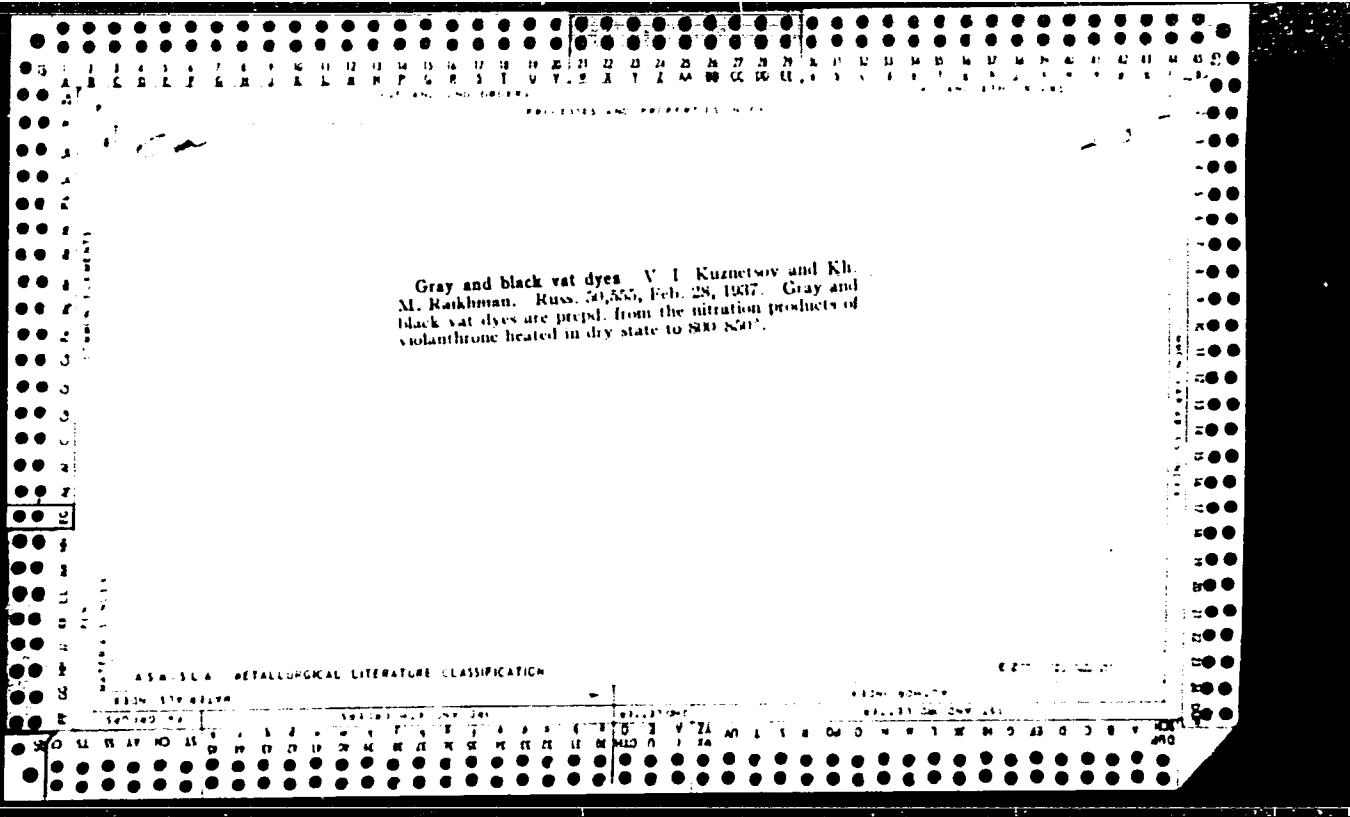
CIA-RDP86-00513R001344020008-9"

The reaction of arylamines with amino and hydroxy derivatives of quinolines in the presence of bisulfite. I. M. Kogan and Kh. M. Kalkman. *J. Applied Chem.* (U.S. S. R.) 12, 1308 (in French, 1309) (1959). - 8-( $\alpha'$ -Aminophenoxyamino)quinoline-5-sulfonic acid (I) (55% theory) was obtained by heating 8-aminoquinoline-5-sulfonic acid with  $p$ -C<sub>6</sub>H<sub>4</sub>(NH<sub>2</sub>)<sub>2</sub> in bisulfite soln. at 109-110° for 8 hrs. The same compd. was obtained by heating 8-hydroxyquinoline-5-sulfonic acid with  $p$ -C<sub>6</sub>H<sub>4</sub>(NH<sub>2</sub>)<sub>2</sub> in bisulfite soln. at the same temp. I can be diazotized and used in combination with the HO deriv. for the prepn. of dyes, but I itself cannot be coupled with diazonium compds. since it is oxidized by the diazonium compd. to quinone. The reaction of hydroxyquinoline with  $p$ -C<sub>6</sub>H<sub>4</sub>(NH<sub>2</sub>)<sub>2</sub> proceeded in two stages: (1) addn. of SO<sub>3</sub>Na to the C having the OH group and (2) reaction of the HO group with the NH<sub>2</sub> group of the diamine. The analogy between 6- and 8-hydroxyquinoline and  $\alpha$ - and  $\beta$ -naphthols was also investigated. Heating of 6-hydroxyquinoline with  $p$ -aminophenol in the presence of NaHSO<sub>3</sub> soln. yielded 76% of 6-( $p$ -hydroxyphenylamino)quinoline, m. 233.5-235°. Heating 8-hydroxyquinoline with the same phenol at the same temp. (115-20° for 27 hrs.), yielded 8-( $p$ -hydroxyphenylamino)quinoline, m. 151-152.5° (37%). The 2 compds. did not condense with diazo compds.

THE CONGRESS  
A. A. Podgorny

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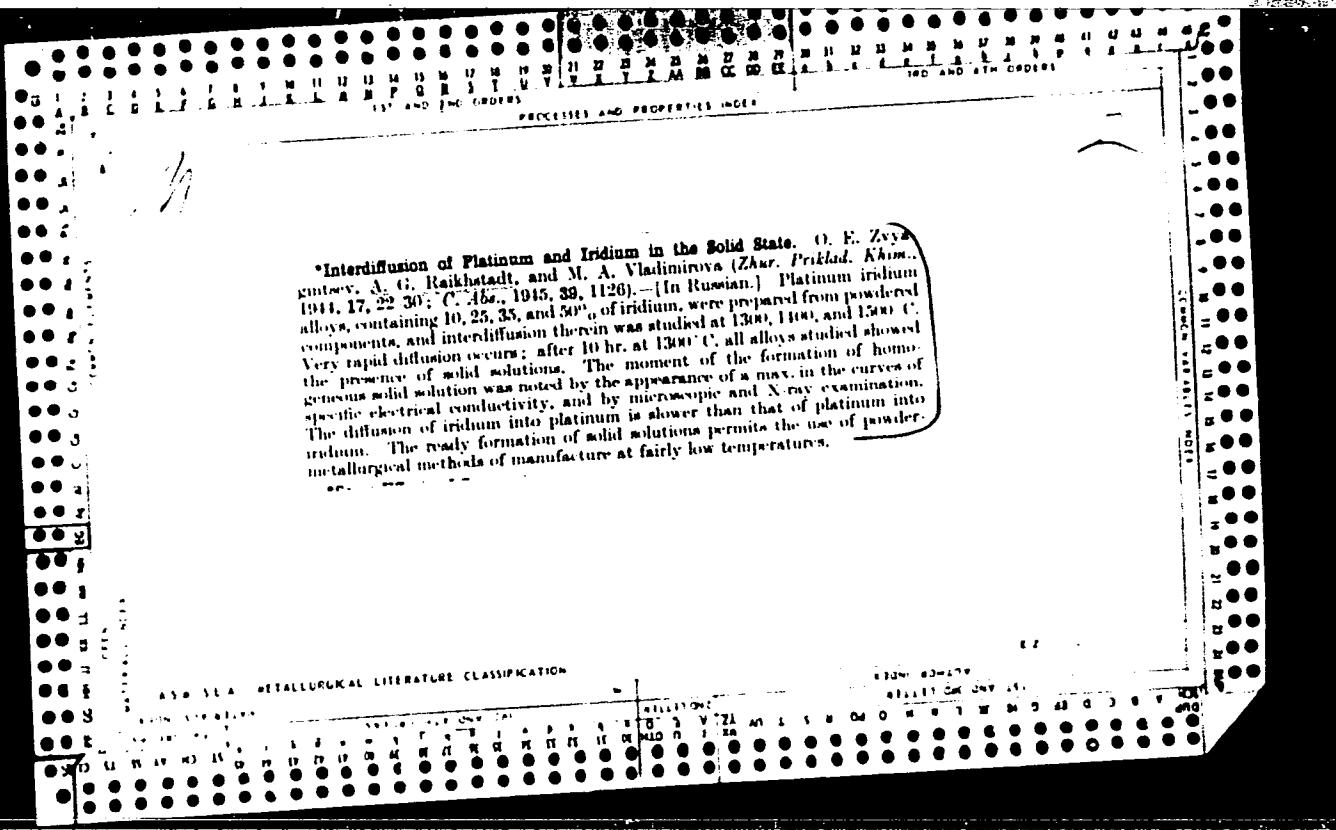
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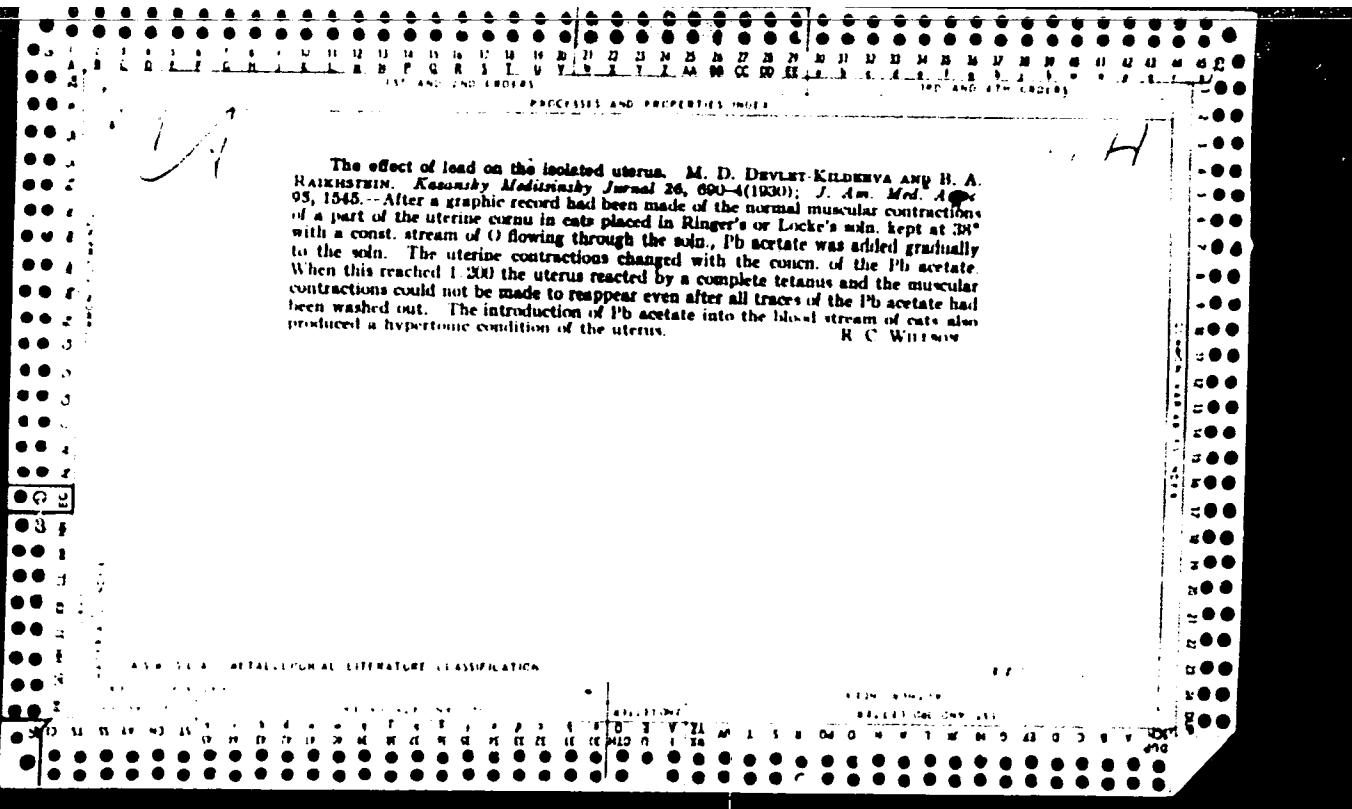
RAIKHRUD, A. Ya.

Quantitative regularities in the regulation of synthesis in the  
virus cell system. Vop. virus. 8 no.3:325-329 My-Je'63.

(MIRA 16:10)

1. Institut virusologii imeni D.I.Ivanovskogo AMN SSSR, Moskva.  
(VIRUS RESEARCH) CELL METABOLISM)





RECORDED, S. I.

Pa-2T49

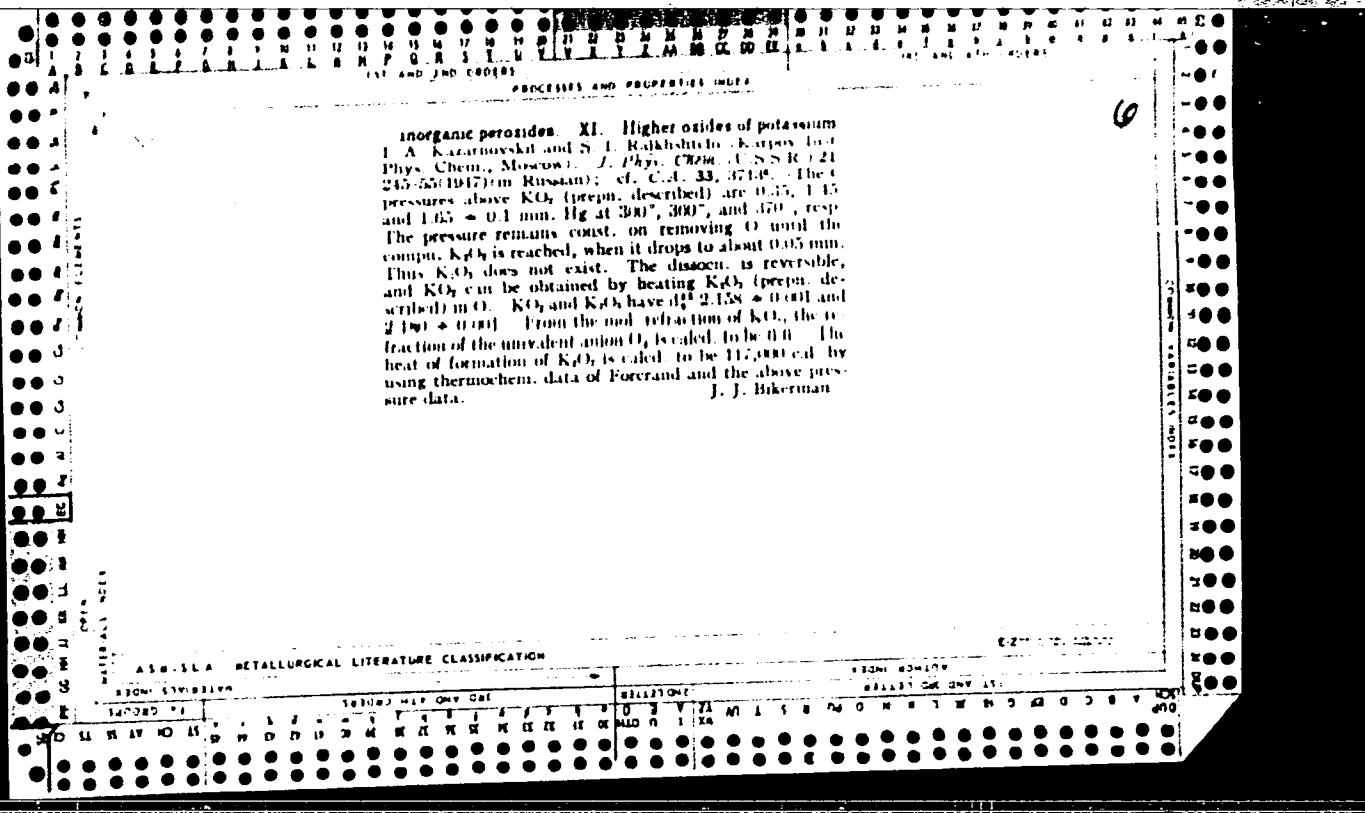
USSR/Physical Chemistry - Apparatus Mar 1947  
Elasticity Measurements

"An Apparatus for Determination of Dissociation Pressure," S I Raikhshtein, and I A Kazarnovskiy,  
4 pp

"Zhurn Fiz Khim" Vol XXI, No 3

Diagrams and operating data of subject equipment  
for elasticity measurements in dissociation of  
hard substances

2T49



Preliminary communications and discussion. Is there a trioxide of potassium? S. I. Barkhatkina and I. A. Kazarnovskii. *J. Phys. Chem.* U.S.S.R. 11, 133 (1938). The existence of  $K_3O_3$  was investigated by us using its dissociation point and that of its products of thermal disintegration up to  $K_2O_2$ . It was found that even the first measurements of the thermal disintegration kinetics at 300° and 0.1 mm. pressure pointed to the fact that  $K_3O_3$  is not an individual compd. but is a mixt. of  $K_2O_2$  with  $K_2O$ . The curve is smooth and no bend is found at the point corresponding to  $K_2O_2$ .

W. R. Bratt

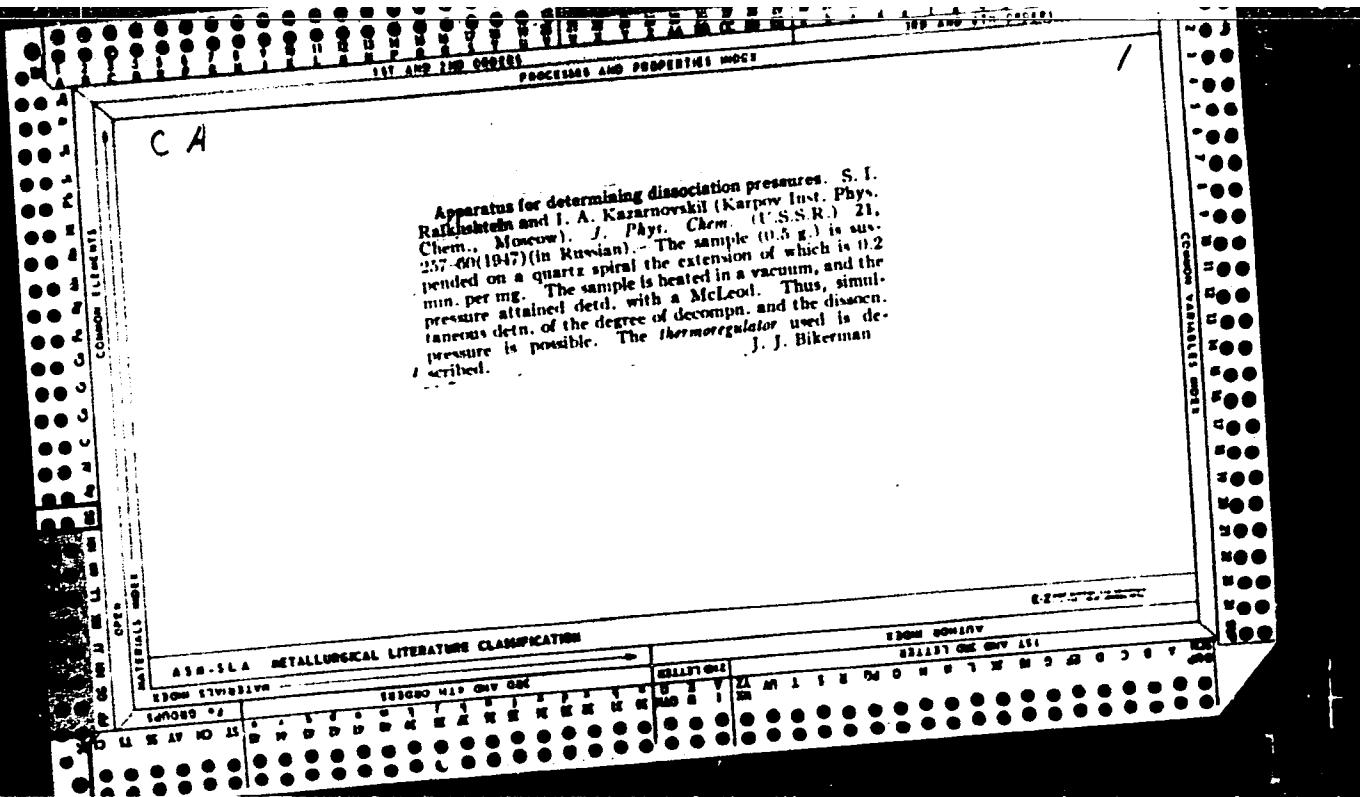
## AS-654 - METALLURICAL LITERATURE CLASSIFICATION

16

AB

Apparatus for the Determination of Dissociation Pressure. (In Russian.) S. I. Raikhshtein and I. A. Kazarnovskii. *Journal of Physical Chemistry (U.S.S.R.)*, v. 21, no. 3, 1947, p. 256-260.

The above apparatus for solid substances is described and diagrammed.



M. IKHTSABE, A.G.

Khimicheskie laboratorii po  
issledovaniyu uglei (Chemical laboratories for coal  
research). Moscow, Gletezhizdat, 1952. 158 p.

SO: Monthly List of Russian Accessions, Vol. 6, No. 1, April 1953

"APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R001344020008-9

MAKHOV, N. I.

MAKHOV, N. I. "The Janzeyursk Seismic Expedition." Vestnik Akad. Nauk S.S.R., Leninigrad,  
Extra Number, 1932, pp. 177-184.

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R001344020008-9"

RATIN, N. V.

Ratin, N. V. "On the Possibility of Observing Mohorovicic Phase During the Caucasian Earthquakes." Trudy Seismologicheskogo Instituta, No. 14, 1930, p. 1-10.

IVANOV, P., inzh.; KERVANBASHILY, St., inzh.; ARSOV, IA., inzh.; RAIKOV,  
K., inzh.

A new foundry binder based on bitumen. *Neftinostroene 13 no.4:*  
23-27 Ap '64.

RAIKOV, Raiko, inzh.

Distribution of power reserves in various power systems in  
Bulgaria. Elektroenergiia 14 no.1:3-5 Ja '63.

RAIKOV, F

"Let us mechanize the fight against the grape Paraconsporalec", p 131  
(ROZTRATYNA ZEME FILIP. Vol 6 #4. Apr. 1951, Bulgaria)

SO: Monthly List of RUSSIAN Accessions, /Library of Congress, August 1953, Uncl.

2.1 A 11

Кирилов (А.) & Райков (Б. В.). Действие на праховидните фунгициди върху  
капилемостта на зеленчуковите семена при максимално напрашване.  
[The effect of fungicidal dusts on the germination of vegetable seed when  
maximum dusting is employed.] -Reprinted from Семепроизводство [Seed  
Production], iv, 1-2, 6 pp., 1945.

In comparative tests conducted over a period of ten years in Bulgaria, the highest  
germination rates of vegetable seeds treated with fungicidal dusts were as follows:  
red cabbage with porzol [R.I.M., xvii, p. 460] 87.75 per cent., tillantin R. [ibid.,  
xxvi, p. 187] 85.75, control 79.5; white cabbage with copper carbonate [ibid.,  
xxvi, p. 324] 82.5, tillantin 78.75, control 60.5; dill [*Puucedanum graveolens*] with  
copper carbonate 76, ceresan 71.25, control 67.5; pepper with tillantin 98.25, con-  
trol 95.75; radish with copper carbonate 84.25, tillantin 82.25, control 44.25;  
lettuce with porzol 88, copper carbonate 81.75, control 81.5; eggplant with tillantin  
73.25, porzol 69.75, control 45. In one test granosan and ceresan completely con-  
trolled *Alternaria radicina* [ibid., xxv, p. 378] on heavily infested carrot seed.

RAIKOV, G., inzh.; ZLATEV, Zl.

Installation of cables at the mine hoisting machine with friction plates. Min delo 16 no.11:24-27 '61.

1. "Mashiproekt"(for Raikov) 2. Gl. mekhanik na Durzhavno minno predpriatie "Burgazski medni mini"(for Zlatev)

(Mining machinery) (Hoisting machinery)

RAIKOV, G.  
TECHNOLOGY

Novelties on the prespinning machines for worsted spinning. p. 19.

LEKA PROMISLJENOST. TEKSTIL. (Ministerstvo na lekata promishlenost) Sofiia.

Vol. ?, no. 6, 1958.

SO: Monthly List of East European Accessions (EEAI) LC

Vol. 8, No. 3  
Uncl. March 1959

RAIKOV, G.

Regulating the mechanisms of the flat carding machine.

P. 21, (Lika Promishlenost) Vol. 6, no. 2, 1957, Sofia, Bulgaria

SO: Monthly Index of East European Acessions (EEAI) Vol. 6, No. 11 November 1957

RADEV, R., inzh.; BALEV, V., inzh.; RAJKOV, Il., inzh.

Determining the depth of dressing of the Chukurovo coals.

Min delo 17 no.11:12-14 '62.

1. Bulgarska akademia na naukite.

RAIKOV, IVAN

Ot Buzludzha do vrukh Botev. Sofiia, Profizdat, 1957. 7sp. (From Buzludzha  
Mountain to Botev Peak. illus., fold map, bibl.)

SO: Monthly Index of East European Accessions (EEAI) Vol. 6, No.11 November 1957